¹³C Nuclear Magnetic Resonance of Organophosphorus Compounds: Dihedral Dependence of the ${}^{2}J_{PCC}$ Spin Coupling in Four-membered Cyclic Phosphines

By George A. Gray*†

(Oregon Graduate Center for Study and Research, Beaverton, Oregon 97005)

and SHELDON E. CREMER

(Department of Chemistry, Marquette University, Milwaukee, Wisconsin 53233)

Summary The α -methyl ${}^{2}J_{POC}$ in 2,2,3,4,4-pentamethylphosphetans is stereospecific with respect to the exocyclic phosphorus substituent; the coupling is large (27-37 Hz) when trans and small (0-5 Hz) when cis to the exocyclic phosphorus substituent.

ALBRAND et al.¹ have previously showed that the two-bond ³¹P-¹H coupling in phosphines exhibits a dihedral dependence. This coupling-constant stereospecificity has permitted easy determination of stereochemistry and allowed analysis of reaction mechanisms, particularly in phosphe-



tans. We have noted a similar stereospecificity in the methyl ${}^{2}J_{PCC}$ in (Ia—c) and (IIa—c).

The data are given in the Table and were obtained from high-resolution 25.1 MHz ¹³C natural abundance spectra using an HA-100 with noise-decoupling and time-averaging. Full details will be provided in a subsequent publication. The α -methyls were assigned on the basis of intensity and upfield shift of axial with respect to equatorial methyls in cyclic systems.²

There is a possibility that a sign reversal may occur as a function of the dihedral angle, as shown by Albrand et al.¹ However, the couplings are different by an order of magnitude in size so that a knowledge of the coupling sign is not critical for obtaining stereochemical orientation to the

TABLE. ²J_{PCC} in 2,2,3,4,4-pentamethylphosphetans $(\pm 0.1 \text{ Hz})C$

Methyl	Isomer (I)		Isomer (II)	
Orientation	Axial	Equatorial	Axial	Equatorial
a: X = Ph	31.8	2.5	4.9	27.8
b: $X = Me$	30.5	2 ·1	4.3	26.9
$: X = Cl \dots$	37.1	0.0	2.5	33.5

exocyclic phosphorus substituent. This can be extended to singly substituted α -carbon atoms. E.g., in 2,2,3,3,-4-pentamethyl-1-phenylphosphetan the 4-methyl has a ${}^{2}J_{PCC}$ of 19.4 Hz, suggesting a trans-relationship to the phenyl substituent. X-Ray studies³ of its stereochemically related⁴ oxide bear out this orientation; ¹H n.m.r. studies are also in agreement with this stereo-chemical assignment.⁴

Stereospecificities in coupling data such as this, not only have great theoretical value in exploring the nature of bonding, but also in filling the very practical need for reliable techniques in structure assignment. The stereospecificity noted for ${}^{2}J_{PCC}$ will prove especially useful in cases where the ${}^{2}J_{PCH}$ is difficult to obtain from complicated ¹H spectra.

One of us (S.E.C.) acknowledges partial support from the National Cancer Institute, the Marquette University Committee on Research, and the Alfred P. Sloan Foundation for a Research Fellowship (1971-1973).

(Received, 10th December 1971; Com. 2105.)

† Present Address: Eastern Applications Laboratory, Varian Associates, 25 Highway 22, Springfield, New Jersey 07081.

¹ J. P. Albrand, D. Gagnaire, and J. B. Robert, Chem. Comm., 1968, 1469; J. P. Albrand, D. Gagnaire, J. Martin, and J. B. Robert, Bull. Soc. chim. France, 1969, 40.

² D. K. Dalling and D. M. Grant, J. Amer. Chem. Soc., 1967, 89, 6612; G. A. Gray and S. E. Cremer, submitted for publication.

^a A. Fitzgerald and C. N. Caughlan, personal communication.
^a S. E. Cremer, B. C. Trivedi, and F. L. Weitl, J. Org. Chem., 1971, 36, 3226; see ref. 28 therein.