Rearrangements of the Gibbane Skeleton during Dehydrogenations with 2,3-Dichloro-5,6-dicyanobenzoquinone

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Summary Gibbanes in which ring A is aromatic react with 2,3-dichloro-5,6-dicyanobenzoquinone giving allylic carbonium ions (via Δ^{4b} -enes) which then undergo Wagner-Meerwein rearrangement; in 7-hydroxy-compounds, the 7,8-bond migrates to C-6 to give a 7-ketone, whereas in 7-deoxy-gibbanes, the 9,9a-bond migrates to position 4b to give a 9a,10-ene.

TREATMENT of the methyl ester (1) of dihydroallogibberic $acid^1$ with an excess of 2,3-dichloro-5,6-dicyanobenzoquinone (ddq) in boiling dioxan gave a 95% yield of a



mixture of 8-epimeric keto-esters from which one pure epimer was isolated by chromatography. The latter, for which structure (5) or (6) was deduced from its u.v., i.r., and n.m.r. spectra, was shown to be the α -methyl epimer (5) because reduction with sodium borohydride, which took place from the less hindered α -face, gave the alcohol (7) in whose n.m.r. spectrum the methyl doublet had shifted downfield from $\tau 8.93$ to 8.63. Structure (5) for the ketoester was confirmed by (a) treatment with base in deuterium



oxide which gave the trideuterio-derivatives (8) and (9) $[m/e \ 299 \ (M^+)$ and 255 $(M-\text{CD}_2=\text{C}=\text{O})$; no peak at $\tau 5.95, \therefore 10\text{-D}$] and (b) oxidation with selenium dioxide to the α -diketone (10) followed by heating with Pd-C gave methyl 1,7-dimethylfluorene-9-carboxylate.² Similarly, reaction of methyl allogibberate³ (2) with ddq gave the rearrangement product (11) in 97% yield.

As expected, treatment of methyl gibberate¹ (12) with ddq did not give the above type of rearrangement. The reaction yielded methyl dehydrogibberate² (14), and a high yield of products resulting from substitution at C-6 of the latter, *e.g.* the ethyl ethers (15) and (16) derived from traces of ethanol present in the dioxan. However a rearrangement

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product, which was identified as the keto-ester (17) by u.v., i.r., and n.m.r. spectroscopy was isolated in low yield. Structure (17) was confirmed by hydrogenation to the tetrahydro-derivative (19) and by oxidation with selenium dioxide in ethanol which gave methyl 1,7-dimethylfluorene-9-carboxylate. Analogous rearrangements took place when methyl deoxogibberate (cf. ref. 2) (13) and the ester $(3)^{\dagger}$ were treated with ddq, the products (ca. 25%) being the esters (18) and (21) respectively. Hydrogenation of the former gave the tetrahydro-compound (20).

Treatment of the 8-hydroxy-ester⁴ (4) with ddq afforded the keto-ester (22).

These rearrangement products are believed to arise by initial formation of the Δ^{4b} -ester followed by removal of a

hydride ion by ddq from C-6 to give an allylic carbonium ion,⁵ e.g. (23), which, unlike the 2-methylene-3-norbornyl cation,⁶ undergoes Wagner-Meerwein rearrangement in either of two ways. If a 7-hydroxy-group is present the lone pair on the oxygen aids [see (23)] migration of the 7,8-bond to C-6 to give products typified by structure (6) in high yield; similarly an 8-hydroxy-group assists rupture of the 7,8-bond. However, in the absence of either a 7- or an 8-hydroxy-group nucleophilic attack at C-6 predominates and rearrangement follows a more unusual course; the 9,9a-bond migrates to C-4b and the product is a 9a,10ene. All new compounds gave satisfactory spectra and analytical data.

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[†] Derived from gibberellin A₇ by acid catalysed aromatisation of ring A⁴ followed by hydrogenation of the ring D double bond.

- ¹ B. E. Cross, J. Chem. Soc., 1954, 4670.
- ² B. E. Cross, J. F. Grove, J. MacMillan, and T. P. C. Mulholland, J. Chem. Soc., 1958, 2520. ³ T. P. C. Mulholland, J. Chem. Soc., 1958, 2693.
- ⁴ B. E. Cross, R. H. B. Galt, and J. R. Hanson, Tetrahedron, 1962, 18, 451.
- ⁵ I. H. Sadler and J. A. G. Stewart, Chem. Comm., 1969, 773; F. E. Lutz and E. F. Kiefer, Tetrahedron Letters, 1970, 4851. ⁶ C. W. Jefford and W. Wojnarowski, Chem. Comm., 1968, 129.