

## Solvolysis of 10,10-Dibromotricyclo[4,3,1,0<sup>1,6</sup>]decane. Evidence for an Intermediate Bicyclo[4,3,1]dec-1(10)-ene Derivative

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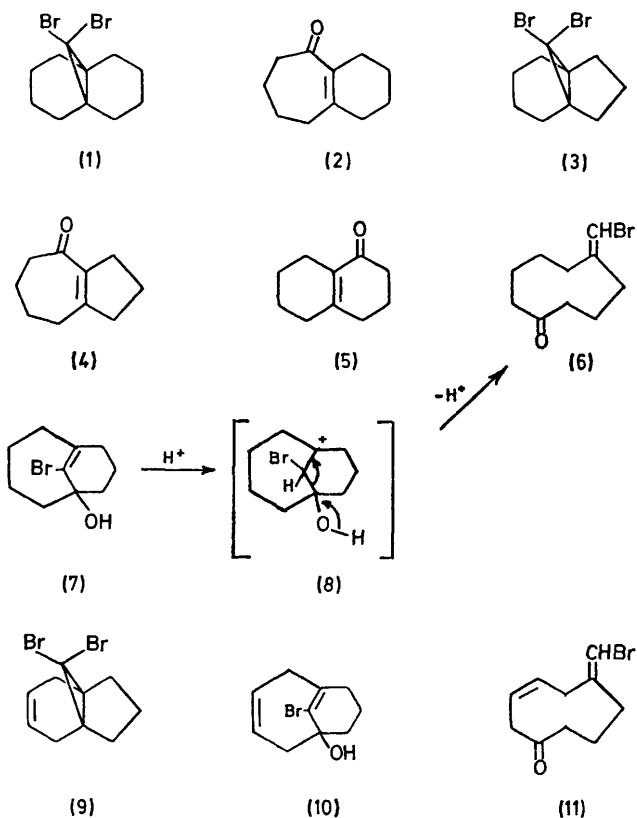
**Summary** Both (3) and (9) readily undergo Ag<sup>+</sup>-promoted solvolysis in aqueous acetone solution to give monocyclic ketones [(6) and (11), respectively] as the major products; it seems likely that the latter compounds result from the fragmentation of highly-strained bicyclic intermediates [(7) and (10), respectively].

Ag<sup>+</sup>-promoted ring-expansion of (3) by a mechanism involving ionization of the bromine atom *endo* to the five-

We recently found<sup>1</sup> that (1) rapidly undergoes silver perchlorate promoted hydrolysis in aqueous acetone (5:95, v/v) solution at 20° to give (2) in good yield. This product (2), which has also been obtained<sup>2</sup> by treating (1) with silver nitrate in methanol solution, would appear to be formed by a mechanism involving a 1,2-alkyl shift. In order to ascertain whether a skeletal rearrangement occurs generally when the more usual ring-expansion reaction is unfavourable,<sup>1</sup> we have examined the solvolysis of 10,10-dibromotricyclo[4,3,1,0<sup>1,6</sup>]decane (3).

When the latter compound (3), which was prepared in high yield by the catalytic hydrogenation of (9),<sup>3</sup> was subjected to Ag<sup>+</sup>-promoted solvolysis under the conditions described above for the hydrolysis of (1), it was completely converted within 10 min into a mixture of 5-bromomethyl-encyclononanone (6) (50%), bicyclo[5,3,0]dec-1(7)-en-2-one (4) (15%) and an unidentified product (<5%). The formation of (4), which was identified by comparison with authentic material,<sup>4</sup> is analogous to the formation of (2) from (1).† The major product was identified as (6) on the basis of its n.m.r. [ $\tau$  (CDCl<sub>3</sub>) 3.98 (1H, s), 7.5–8.3 (14H)], i.r. [ $\nu_{\max}$  (film) 1702s, 1618w cm<sup>-1</sup>], and high resolution mass [Found:  $M^+$  at  $m/e$  230.0308. Calc. for C<sub>10</sub>H<sub>15</sub><sup>79</sup>BrO: 230.0307] spectra, and the elemental analysis of its 2,4,6-tri-isopropylbenzenesulphonylhydrazone,<sup>5</sup> m.p. 127–128° (decomp.).

It seems likely that (6) results from the fragmentation of a highly-strained intermediate, 10-bromobicyclo[4,3,1]dec-1(10)-en-6-ol (7), possibly *via* the carbonium ion (8). The intermediate (7) is the expected product of the normal



membered ring and concerted disrotatory opening of the resulting cyclopropyl cation in the allowed direction.<sup>6</sup> The instability of (7) can be explained in terms of its being a

† It is noteworthy that the isomeric ketone (5), which could have been formed from (3) by a mechanism involving the migration of the five-membered ring, was not detected in the products.

bridged *trans*-cycloheptene derivative rather than by invoking Bredt's rule.<sup>7</sup> It is especially noteworthy that the solvolysis of (1) does not lead to a similar fragmentation process. Attempts to trap (7) as a bicyclic adduct have so far been unsuccessful.

The Ag<sup>+</sup>-promoted hydrolysis of (9) was also found to proceed rapidly at 20° under the above conditions and 6-bromomethylenecyclonon-3-en-1-one (11) (50%) was obtained as the major product. It seems likely that this reaction also proceeds *via* an intermediate bicyclo[4,3,1]-

dec-1(10)-ene derivative (10). It would appear to follow from the sharpness of the bromomethylene proton resonances in their n.m.r. spectra that (6) and (11) are both pure geometrical isomers and thus that the fragmentation reaction is stereospecific.

One of us (M.R.D.S.) thanks the S.R.C. for the award of a Research Studentship.

(Received, 30th August 1972; Com. 1513.)

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