

Synthesis of a Cyclopropa[*b*]naphthalene

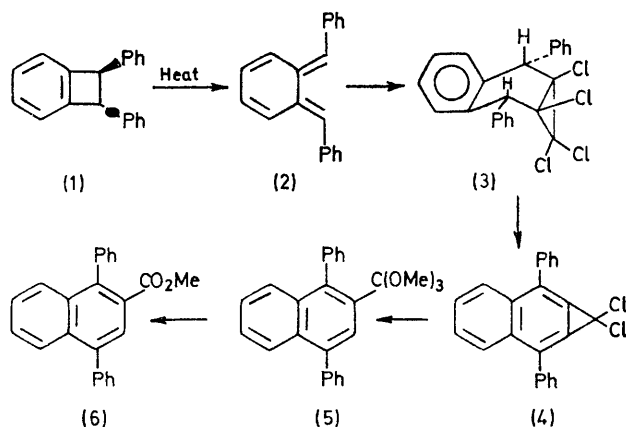
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Summary Dehydrochlorination of the benzobicycloheptene (3) gives 1,1-dichloro-2,7-diphenylcyclopropa[*b*]naphthalene (4) in low yield; the three-membered ring is readily cleaved by protic solvent.

WHILE the chemistry of cyclopropabenzene is well established,¹ no other aromatic system incorporating a linearly fused cyclopropene ring has been reported. Although cyclopropa[*b*]naphthalene is expected to have a high degree of π -bond localisation,^{2,3} prior attempts to obtain derivatives of this ring system have failed.³ We now report the synthesis of a cyclopropa[*b*]naphthalene derivative.

Of the synthetic methods available for the synthesis of cyclopropabenzene,¹ the dehydrohalogenation of substituted bicyclo[4,1,0]hept-3-enes^{4,5} appeared to be the most amenable to modification for the synthesis of cyclopropa[*b*]naphthalenes. The electrocyclic ring-cleavage product (2) of the cyclobutabenzene (1) is known⁶ to undergo [π 4 + π 2]cycloaddition and this consequently offers a



simple synthesis of the desired benzobicycloheptene (3). Indeed, thermolysis of (1)⁷ in the presence of tetrachlorocyclopropene⁸ gives (3)† (80%), m.p. 171—172°. De-

† Satisfactory microanalytical and spectral data were obtained for all new compounds reported.

hydrochlorination of (3) with a slight excess of KOBu^{\dagger} in tetrahydrofuran, in a manner strictly analogous to that used to obtain the corresponding *gem*-dichlorocyclopropabenzene,⁵ resulted in discolouration and formation of the cyclopropanaphthalene (4)† [25%; m.p. 177—179°; δ (CDCl_3) 7.51—7.53 p.p.m. (complex m), ν_{max} (KBr) 1675, 1358, 1000, 770, 755, 715, and 695 cm^{-1}] which was isolated by fractional crystallisation.

The chemistry of (4) parallels that of the *gem*-dichlorocyclopropabenzene analogue. For example, treatment of

(4) with methanol results in cleavage of the three-membered ring and formation of the naphthoate (6)⁹ *via* the orthoester (5). An *X*-ray structure determination of (4) is being undertaken.

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† A mass value accurate to 1.6 p.p.m. was obtained for $\text{C}_{23}\text{H}_{14}^{35}\text{Cl}^{37}\text{Cl}$.

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