

Contact Term Contribution to Lanthanide-induced ^{13}C Nuclear Magnetic Resonance Shifts in Pyridine and β -Picoline

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Summary There is a large contact term contribution to ^{13}C paramagnetic shifts induced by $\text{Eu}(\text{dpm})_3$, $\text{Pr}(\text{dpm})_3$, and $\text{Eu}(\text{fod})_3$ in pyridine and β -picoline; this contribution may not be due mainly to spin delocalization through π -bonds.

ALTHOUGH it is generally accepted that paramagnetic shifts induced by some lanthanide complexes are mainly due to pseudo-contact interaction,¹ it has been recently reported that abnormal shift values are observed for atoms (^{14}N), co-ordinated to the lanthanide ion² and for ^{13}C or ^1H nuclei close to the co-ordinating atom in some aliphatic compounds.³ Johnson *et al.* have reported that abnormal relative ^1H shift patterns are observed for substituted pyridine *N*-oxides and anilines with some lanthanide complexes, and that the shift patterns for these substrates with $\text{Eu}(\text{fod})_3$ are characteristic of a contact interaction attributed to π -spin density.⁴ For pyridine derivatives, the observed shifts induced by $\text{La}(\text{dpm})_3$ ($\text{La} = \text{Pr}, \text{Eu}, \text{and Yb}$) could be interpreted approximately in terms of a pseudo-contact interaction only by using a distance between nitrogen and lanthanide ion of 4 \AA .⁵ Huber *et al.* tentatively took account of the contact term contribution through π -bonding by using the Hückel spin distribution of quinoline anion radical, to obtain better predicted values,⁶ but the agreement is not good.

We report here observed ^{13}C paramagnetic shifts induced by $\text{Eu}(\text{dpm})_3$, $\text{Pr}(\text{dpm})_3$, and $\text{Eu}(\text{fod})_3$ for pyridine and β -picoline, and discuss the contact term contribution.

The observed ^1H and ^{13}C shifts for pyridine and β -picoline are shown in Table 1 with the related pseudo-contact shifts calculated from the geometric factor $(3\cos^2\theta - 1)/r^3$. Although the observed relative shifts for ^1H can be explained with the pseudo-contact term only by using a value of r_{N} (the nitrogen-lanthanide distance) of 3.00 \AA ,⁵ those for ^{13}C disagree completely with the values predicted on this basis. With $\text{Eu}(\text{dpm})_3$ and $\text{Eu}(\text{fod})_3$, the large upfield shift component is clearly included in the β - ^{13}C resonance. These results thus show the existence of a much larger contact term contribution for ^{13}C than for ^1H . The agreement between the observed relative shifts with those

calculated using only the pseudo-contact term for ^1H is presumably not co-incidental, and so the other term is not required. The most suitable values for K in the equation $S_{\text{pcs}} = (3\cos^2\theta - 1)/r^3 \times K$ for pyridine are 594, 1776, and 880 with $\text{Eu}(\text{dpm})_3$, $\text{Pr}(\text{dpm})_3$, and $\text{Eu}(\text{fod})_3$ respectively. The S_{pcs} values for ^{13}C calculated using $K = 594$ are shown in Table 2.

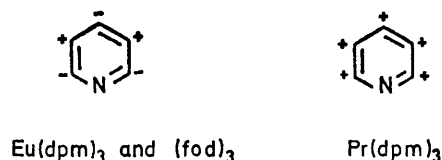


FIGURE. S_{CS} patterns for ^{13}C .

Observed S values for pyridine obtained with $\text{Eu}(\text{dpm})_3$ in CHCl_3 (or CDCl_3) are used to discuss the ^{13}C contact term in detail. The contribution to the lanthanide-induced shift other than that from the pseudo-contact term may be considered to be only that due to the contact term (S_{CS}) (see Table 2). Assuming that S_{CS} arises only through σ - π interaction parameters in the McConnell and the Karplus-Fraenkel equation⁸ were used.

$$S = S_{\text{pcs}} + S_{\text{CS}}$$

were estimated from McLachlan's spin distribution⁷ in the pyridine anion radical by using the approximation that the bonding orbital formed between an f -orbital on Eu containing an unpaired electron and a π^* -MO of pyridine may be regarded as a π^* -MO itself. The following values for σ - π interaction parameters in the McConnell and the Karplus-Fraenkel equation⁸ were used.

$Q_{\text{CH}}^{\text{H}} = -23.0 \text{ G}$ in the equation $a^{\text{H}} = Q_{\text{CH}}^{\text{H}} \rho_{\text{C}}^{\text{H}}$, $Q_{\text{CN}}^{\text{C}} = 4.0 \text{ G}$ and $Q_{\text{NC}}^{\text{C}} = -20.5 \text{ G}$;⁹ Q^{C} values for C and H, and C and C are given by

$$a^{\text{C}} = (S^{\text{C}} + \sum_i Q_{\text{CX}(i)}^{\text{C}}) \rho^{\pi} + \sum_i Q_{\text{X}(i)\text{C}}^{\text{C}} \rho_i^{\pi}$$

TABLE 1. Observed paramagnetic shifts, S^a

			Pyridine			
			α	β	γ	
Eu(dpm) ₃	¹³ C	-67.5	+ 3.5	-23.0
			¹ H	-12.16	- 4.46	- 3.86
Pr(dpm) ₃	¹³ C	+93.5	+25.8	+25.8
			¹ H	+35.3	+12.7	+10.6
Eu(fod) ₃	¹³ C	-43.5	+14.5	-11.0
			¹ H	-18.0	- 6.25	- 6.10
Calc. ^b	¹³ C	1.00	0.42	0.34
			¹ H	1.00	0.36	0.30

			β -Picoline				
			α	β	γ	β -Me	
Eu(dpm) ₃	¹³ C	{ -78.3	- 4.7	-26.6	- 8.9
				{ -75.3	- 3.8		
			¹ H	-17.5	- 6.1	- 5.8	- 4.0
Pr(dpm) ₃	¹³ C	{ +86.5	+22.6	+25.9	+ 9.7
				{ +79.5	+21.8		
			¹ H	+38.6	+14.0	+11.2	+ 8.6
Calc. ^b	¹³ C	1.00	0.42	0.34	0.21
			¹ H	1.00	0.36	0.30	0.22

^a S is the lanthanide-induced shift in p.p.m. obtained by linear extrapolation to a complex-substrate ratio of 1.0. Negative signs designate shifts to lower field. ¹³C spectra were recorded at 22.63 MHz with a Hitachi R-22 spectrometer with a R-228 proton wide band decoupler and a A-1600A signal averaging analyser, and ¹H spectra with a Hitachi R-20A spectrometer at 60 MHz, both spectrometers operating in the frequency-swept mode. Solutions were ca. 2.0M for ¹³C and ca. 0.5M for ¹H. The amount of lanthanide complex was continuously varied up to a complex-substrate molar ratio of 0.2. CHCl₃ was the solvent for ¹³C and CDCl₃ for ¹H.

^b Geometric factors relative to the α -position.

TABLE 2. Contact term contribution in the Eu(dpm)₃-pyridine system

	S_{obs}^a	$(3\cos^2\theta - 1)/r^3$	S_{pcs}^b	S_{cs}^c	S_{π}	a^c or a^H (in G)
α -H	-12.16	0.02045	-12.15	- 0.01	+ 3.00B	- 3.00
β -H	- 4.46	0.00742	- 4.41	- 0.05	- 0.32B	+ 0.32
γ -H	- 3.86	0.00615	- 3.65	- 0.21	+ 9.66B	- 9.66
α -C	-67.5	0.03032	-18.01	-49.49	+14.56B	- 3.64
β -C	+ 3.5	0.01274	- 7.57	+11.07	+32.64B	- 8.16
γ -C	-23.0	0.01036	- 6.15	-16.85	-61.60B	+15.40

^a Observed shifts. ^b Pseudo-contact term using $K = 594$ and $r_N = 3.00$ Å. ^c $S_{obs} - S_{pcs}$.

From the contact shift equation and the relation $\gamma_H/\gamma_C = ca. 4$, the π -contact term is $-a^H \cdot B$ for ¹H and $-4a^C \cdot B$ for ¹³C, where B is a constant. Comparison of S_{cs} with the S_{π} values for ¹H in Table 2 shows that the limiting B value is ca. 0.1 at most, and, therefore, S_{π} for ¹³C is $-0.4 a^C$. The π -contact contribution for ¹³C would thus be very small, even if it were present at all, and may be due mainly to another spin transfer mechanism which affects ¹³C much more than ¹H. The shift patterns due to this effect with Eu(dpm)₃ and Eu(fod)₃ are downfield for α - and γ -¹³C and up field for β -¹³C (Figure). The sign alternation is opposite

to that observed for the Ni(acac)₂-pyridine system where the σ -contact term has already been shown to be the main contributor to the paramagnetic shift.¹⁰ S_{cs} values for ¹³C in the Pr(dpm)₃-pyridine system are positive for all positions

Although the spin-transfer mechanism is still under investigation, we assume that a positive spin induced directly at α -¹³C by Eu contributes to the ¹³C paramagnetic shifts induced by Eu(dpm)₃ and Eu(fod)₃.

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