

## 1,11-Dipolar Cycloaddition

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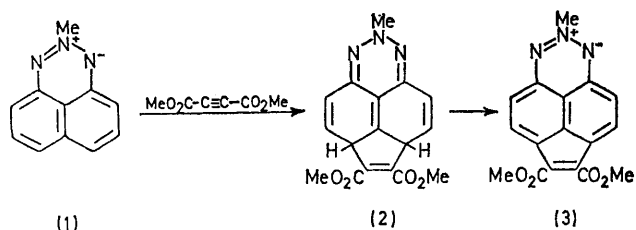
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**Summary** 2-Methylnaphtho[1,8-*de*]triazine undergoes 1,11-dipolar ( $12\pi + 2\pi$ ) cycloaddition to acetylenic esters to give 2-methylacenaphtho[5,6-*de*]triazines, after spontaneous dehydrogenation.

2-METHYLNAPHTHO[1,8-*de*]TRIAZINE (1)<sup>1</sup> is of interest since it incorporates the dipolar azimine function<sup>2</sup> for which the first 1,3-dipolar cycloaddition reactions have only recently been observed.<sup>3</sup>

Gradual addition of dimethyl acetylenedicarboxylate (2 equiv.) to the blue naphthotriazine (1) in refluxing *o*-dichlorobenzene over 3 h gave the red acenaphtho[5,6-*de*]triazine (3) (30–40%), m.p. 233°. The <sup>1</sup>H n.m.r. spectrum of (3) [ $\tau$ (CDCl<sub>3</sub>) 1.05 and 2.19 (4H, ABq, *J* 8.5 Hz), 5.08 (s, *N*-Me), and 5.88 (2 identical Me)] and its u.v. spectrum in ethanol [ $\lambda_{\max}$  247 ( $\epsilon$  18,500), 276 (13,200), 340sh (38,500), 352 (51,400), and 504 nm (6230)] are closely analogous to those reported for 2-methylacenaphtho[5,6-*de*]triazine.<sup>4</sup>

Formation of (3) is rationalised by a thermally allowed 1,11-dipolar cycloaddition to give (2), followed by dehydrogenation to give the stable  $14\pi$  aromatic system. In accord with this, the reaction was cleaner and the best yield was obtained in the presence of 3 equiv. of sulphur as dehydrogenating agent.



Analogous cycloadditions were observed with diethyl acetylenedicarboxylate and methyl propiolate, but initial indications are that reactions with olefinic dipolarophiles are more complex. 2-(2,4-Dinitrophenyl)naphtho[1,8-*de*]triazine reacts in a similar manner to the 2-methyl compound.

These reactions provide the first examples of  $12\pi + 2\pi$  cycloaddition and, together with the recently reported 1,5-dipolar cycloaddition reactions,<sup>5</sup> strikingly illustrate the potential extension of 1,3-dipolar cycloaddition to vinylous dipolar systems of more than  $4\pi$  electrons.

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<sup>3</sup> S. F. Gait, M. J. Rance, C. W. Rees, and R. C. Storr, *J.C.S. Chem. Comm.*, 1972, 688.

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