

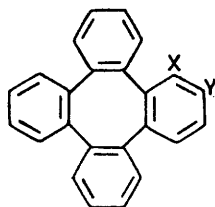
Resolution and Optical Stability of Tetrabenzocyclo-octatetraene Derivatives

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Summary The tetrabenzocyclo-octatetraene derivative (2) has been partially resolved, and a lower limit for the barrier to racemization of 45 kcal mol⁻¹ has been determined.

WE have previously shown¹ that the tetrabenzocyclo-octatetraene derivative (1) exists as a mixture of diastereomers which do not interconvert rapidly on the n.m.r. time scale at 98° ($\Delta G_{98}^{\ddagger} > 21$ kcal mol⁻¹).† In order to obtain a more meaningful estimate of the barrier to ring inversion in such compounds, we have prepared optically active (2), and studied its optical stability at elevated temperatures.



- (1) X=H, Y=CH(OH)Me
- (2) X=H, Y=CO₂H
- (3) X=H, Y=CHO
- (4) X=CHO, Y=H

Treatment of tetraphenylene with titanium tetrachloride and dichloromethyl methyl ether in methylene chloride gave a mixture of (3) and (4), which were separated by fractional crystallization from methylene chloride-hexane.¹ Oxidation of (3) with silver oxide in ethanol, followed by acidification, yielded racemic (2), m.p. 281–284°.‡ Partial resolution of (2) by fractional crystallization of the brucine salt from ethanol gave (+)-(2), m.p. 285–287°, [α]₃₂₀ +23.3° (c 0.45 in EtOH), and (-)-(2), m.p. 287–288°, [α]₃₂₀ -8.4° (c 0.57 in EtOH). A sample of (+)-(2) showed no change in rotation within experimental error (ca. ±10%) after heating at 251° for 36.0 h (0.038 molal solution in biphenyl). This result yields a lower limit to racemization, and hence a lower limit to ring inversion, of $\Delta G_{251}^{\ddagger} > 45$ kcal mol⁻¹.

A comparison of the barrier for ring inversion in (2) with the barriers reported for less sterically hindered cyclo-octatetraene derivatives¹ is a striking confirmation of our thesis that non-bonded interactions constitute the major source of structural rigidity exhibited by such systems.¹

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† Added in proof. Independently, A. Rosdahl and J. Sandström (*Tetrahedron Letters*, 1972, 4187) have reported a lower limit to ring inversion of 26 kcal mol⁻¹ for another derivative of tetrabenzocyclo-octatetraene.

‡ Satisfactory elemental and mass spectral analyses were obtained for (2). N.m.r. spectral data were in agreement with the structure assigned.

* G. H. Senkler, jun., D. Gust, P. X. Riccobono, and K. Mislow, *J. Amer. Chem. Soc.*, 1972, **94**, 8626.