Journal of

The Chemical Society,

Chemical Communications

NUMBER 5/1973

7 MARCH

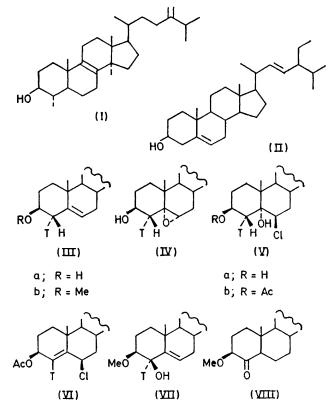
Inversion of the 4β-Hydrogen during the Conversion of the Sterol Obtusifoliol into Poriferasterol by *Ochromonas malhamensis*

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Summary The conversion of $[2,2,4-^{3}H_{3}]$ obtusifoliol into poriferasterol by the alga Ochromonas malhamensis has been demonstrated and proceeds with retention of at least 30% of the axial 4β -hydrogen which is inverted into the equatorial 4α -position of poriferasterol.

DEMETHYLATION of 4,4-dimethyl sterols in animals¹ and plants² proceeds by loss of the 4α -methyl group with the 4β -methyl group epimerising into the 4α -position of the product 4-monomethyl sterol. In the subsequent demethylation of the 4α -methyl sterol the fate of the 4β hydrogen is unknown but the preparation of tritiumlabelled obtusifoliol (I), a suggested phytosterol precursor,³ and its ready conversion into poriferasterol (II) by Ochromonas malhamensis has now permitted a study of this aspect of the C-4 demethylation.

 $[2,2,4-^{3}H_{3}]$ Obtusifoliol (29.8 mCi/mmol) was prepared from unlabelled obtusifoliol by the methods described previously.⁴ The $[2,2,4-^{3}H_{3}]$ obtusifoliol (30 μ Ci) was incubated with *O. malhamensis* for six days and the sterol (2% incorporation) isolated and analysed by g.l.c. with sample trapping which revealed that 95% of the radioactivity was present in the major component⁵ poriferasterol (II). To facilitate the determination of the fate of the C-4 tritium the tritiated poriferasterol was mixed with $[1,7,15,22,25-^{14}C_{5}]$ poriferasterol[‡] and the mixture crystallised to constant specific radioactivity and ³H: ¹⁴C ratio of 9·18:1. A sample of the poriferasterol (partial structure IIIa) was epoxidised to give the derivatives§ (IV), ³H: ¹⁴C ratio 9·03:1, which was converted into the chlorohydrin^{6,7} (Va), ³H: ¹⁴C ratio



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 \ddagger [1,7,15,22,26-14C₅]Poriferasterol was kindly donated by Dr. G. H. Beastall and was obtained by the culture of O. malhamensis in the presence of [2-14C] mevalonic acid.

\$ All compounds were fully characterised by m.p., i.r., u.v., n.m.r., and mass spectrometry. Purity was established by t.l.c. and where possible also by g.l.c.

9.02:1. Formation of the monoacetate (Vb) followed by dehydration with SOCl₂ in pyridine⁷ gave the acetate (VI), ³H: ¹⁴C ratio 9.00: 1, which was then reduced with LiAlH₄^{8,9} to produce the dienol (IIIa, poriferasterol), ³H:¹⁴C ratio 9.11:1. It is established that the dehydration of alcohols of type (Vb) occurs trans-diaxially¹⁰ and this was verified in the present work by taking (24S)-24-ethyl-4 β -[²H₁]cholesta-5,22-dien-3 β -ol through the above sequence of reactions with a resulting loss of deuterium.

Since no changes in the ³H:¹⁴C ratios were observed the above results established that either tritium was in the 4α -position in the poriferasterol or alternatively no tritium was located at C-4. To check the latter possibility a portion of the dual labelled poriferasterol was converted into the tosylate and then heated under reflux in anhydrous methanol¹⁰ to give the methyl ether (IIIb), ${}^{3}H: {}^{14}C$ ratio $8\cdot82:1$. The methyl ether (IIIb) was then refluxed in acetic acidbenzene with SeO_{2}^{11} to produce (VII), ${}^{3}H$: ${}^{14}C$ ratio 8.95: 1. The retention of all the tritium in this allylic oxidation would be predicted if the tritium was in the equatorial 4α -position and the reaction occurred with retention of configuration. The configuration of the 4β -hydroxy-group was established by the n.m.r. spectrum [τ 5.76 (d, J 4 Hz, eq-4-H) and 4.33 (deshielded m, olefinic 6-H)]. Further confirmation of this point was obtained by preparation of samples of (24S)-24-ethylcholesta-5,22-dien-3 β ,4 α -diol by SeO₂ oxidation of the corresponding 3β -monol, and by basic hydrolysis of 4β -acetoxy-(24S)-24-ethylcholesta-5,22dien-3 β -ol obtained by Br₂-silver acetate treatment¹² of (24S)-24-ethylcholesta-5,22-dien-3 β -ol. The two compounds were identical and formed the same acetonide, which could only occur with a trans-diequatorial or a cis-diol. Since the n.m.r. spectrum revealed an equatorial C-4 proton

the C-4 hydroxy-group must have been β -orientated and this established the stereochemistry of the SeO₂ oxidation.

To determine the presence of tritium at C-4 the radioactive 4β -alcohol (VII) was reduced¹¹ to 3β -methoxy-(24S)-24-ethylcholestan-4 β -ol, ³H: ¹⁴C ratio 8.86: 1. Oxidtion with Jones' reagent then gave the ketone (VIII), 3H:14C ratio 7.87:1, with a loss of 10.5% of tritium. In a duplicate degradation, production of (VIII) resulted in a loss of 12.3%of tritium whilst labelled poriferasterol obtained by incubation of 24-ethylidene [2,2,4-3H3] lophenol with O. malhamensis4 lost 16% of tritium upon conversion into (VIII).

It can therefore be concluded that tritium was located at C-4 of the poriferasterol (II) and that it was exclusively in the 4α -position. Thus demethylation of the 4α -monomethyl precursor (I) must have proceeded by a mechanism in which at least 30% of the original 4β -hydrogen was retained and inverted to the 4α -position. The fact that only about 10-12% of the tritium content of the poriferasterol was in the 4α -position compared with a possible maximum of 33% may indicate that some exchange of the C-4 proton occurs during 4α -demethylation, for example by enolisation of the 3-oxo-compound which is implicated¹³ as an intermediate. Alternatively it may be a consequence of the basic enolisation process employed⁴ to prepare the [2,2,4-3H₃]obtusifoliol which may favour enolisation towards the less substituted C-2 with the result that much less tritium is incorporated into the 4β - than the 2α - and 2β positions. A final decision between these two explanations awaits further investigation.

We thank the S.R.C. for financial support, the N.I.H. for a Postdoctoral Fellowship to F.F.K., and Drs. W. Amarego and I. F. Cook for helpful discussions.

(Received, 28th November 1972; Com. 1988.)

¹ K. B. Sharples, T. E. Snyder, T. A. Spencer, K. K. Maheshwari, G. Gahn, and R. B. Clayton, *J. Amer. Chem. Soc.*, 1968, 90, 6874; K. B. Sharples, T. E. Snyder, T. A. Spencer, K. K. Maheshwari, J. A. Nelson, and R. B. Clayton, *ibid.*, 1969, 91, 3394. ² E. L. Ghisalberti, N. J. de Souza, H. H. Rees, L. J. Goad, and T. W. Goodwin, *Chem. Comm.*, 1969, 1403; F. F. Knapp and H. J.

- Nicholas, *ibid.*, 1970, 399. ^{*} L. J. Goad and T. W. Goodwin, Progr. Phytochem., 1972, 3, 113.
- ⁴ J. R. Lenton, J. Hall, A. R. H. Smith, E. L. Ghisalberti, H. H. Rees, L. J. Goad, and T. W. Goodwin, Arch. Biochem. Biophys., 1971, 143, 664.
- ⁵ M. C. Gershengorn, A. R. H. Smith, G. Goulston, L. J. Goad, T. W. Goodwin, and T. H. Haines, Biochemistry, 1968, 7, 1698.
- ⁶ R. A. Baxter and F. S. Spring, J. Chem. Soc., 1943, 613.
 ⁷ F. S. Spring and G. Swain, J. Chem. Soc., 1939, 1356.
- ⁸ R. T. Ireland, T. I. Wrigley, and W. G. Young, J. Amer. Chem. Soc., 1959, 81, 2818.
 ⁹ D. J. Collins and J. J. Hobbs, Austral. J. Chem., 1964, 17, 677.
- ¹⁰ D. N. Kirk and M. P. Hartshorn, 'Steroid Reaction Mechanisms,' Monograph 7, Reaction Mechanisms in Organic Chemistry, Elsevier, 1968.
- ¹¹ B. R. Brown and D. M. L. Sandbach, *J. Chem. Soc.*, **1963**, **5313**.
 ¹² V. A. Petrow, O. Rosenheim, and W. W. Starling, *J. Chem. Soc.*, **1943**, **135**.
 ¹³ A. D. Rahimtula and J. L. Gaylor, *J. Biol. Chem.*, **1972**, **247**, 9.