Cycloaddition of Cycloheptatrienethione with Dienophiles

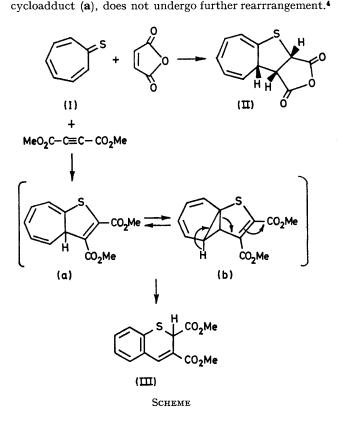
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Summary The cycloaddition of cycloheptatrienethione (I) with maleic anhydride gives a rare $[\pi 8_s + \pi 2_s]$ type of adduct (II), while with dimethyl acetylenedicarboxylate it gives an unusual rearrangement product (III).

ALTHOUGH cycloheptatrienethione (I), the S-analogue of tropone, has been synthesized,¹ its thermal instability has so far precluded a detailed chemical study. We report here on its cycloaddition with dienophiles. Tropones have been reported to react normally with maleic anhydride and dimethyl acetylenedicarboxylate giving 1,4-cycloadducts.² We found that (I) reacted smoothly and rapidly, with immediate disappearance of the colour, with some dienophiles, undergoing a relatively rare $[\pi 8_s + \pi 2_s]$ type of cycloaddition.3

Reaction of (I) with maleic anhydride in benzene solution at 10° for 30 min gave a 1,8-cycloadduct (II), m.p. 98-99°, in quantitative yield; τ (CDCl₃, Me₄Si): 3·29-3·65 (3H, complex m, 6-, 7-, 8-H), 3.80 (1H, m, 5-H), 4.60 (1H, dd, J 9.0 and 5.5 Hz, 4-H), 5.00 (1H, d, J 9.0 Hz, 2-H), 5.72 (1H, dd, J 10.2 and 9.0 Hz, 3-H), and 6.93 (1H, ddt, J 10.2, 5.5, and 1.2 Hz, 3a-H).[‡] Conversely, reaction of (I) with dimethyl acetylenedicarboxylate under similar conditions yielded the thiachromene (III), pale yellow prisms, m.p. 91—92° (ca. 65%); τ (CDCl₃, Me₄Si): 2·21 (1H, s, 4-H), 2.58-2.97 (4H, complex m, Ar-H), and 5.29 (1H, s, 2-H). Assignments and coupling constant values were confirmed by spin decoupling experiments. We suggest that (III) may be formed via the sequence shown in the Scheme. The cycloadduct (a) was not detected spectroscopically even at low temperature in CCl_4 solution. We feel that the [8+2] type of cycloadduct (a) should give rise to a norcaradiene intermediate (b) which could then smoothly rearrange owing to the electron-withdrawing group at C-2. It should be noted that a 1,8-cycloadduct of heptafulvene



with dimethyl acetylenedicarboxylate, unlike the 1,8-

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[†] Satisfactory elemental analyses were obtained for all new compounds.

§ Detailed analyses of the decoupling will be reported elsewhere.

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² T. Nozoe, T. Mukai, and J. Minegishi, Proc. Japan. Acad., 1952, 28, 287; T. Nozoe, T. Mukai, T. Nagase, and Y. Toyooka, Bull, Chem. Soc. Japan, 1960, 33, 1247; S. Ito, H. Takeshita, Y. Shoji, Y. Toyooka, and T. Nozoe, Tetrahedron Letters, 1968, 3215; T. Uehara, and Y. Kitahara, Chem. and Ind., 1971, 27, 354; S. Ito, A. Mori, Y. Shoji, H. Takeshita, Tetrahedron Letters, 1972, 2685.
³ R. B. Woodward, and R. Hoffmann, 'The Conservation of Orbital Symmetry,' Verlag Chemie-Academic Press, New York, 1970, 2020.

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