

Stereochemistry of the Biological Divinyloxiran–Dihydro-oxepin Rearrangement: X-Ray Studies of the Sesquiterpenoids Miscandenin and Dihydromikanolide

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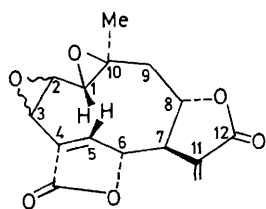
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Summary The stereochemistries of the sesquiterpenoids dihydromikanolide and miscandenin have been determined by X-ray crystal-structure analysis: the results define the stereochemistry of the biological divinyloxiran–dihydro-oxepin rearrangement.

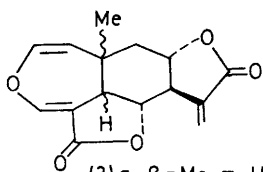
SEVERAL closely related sesquiterpenoid dilactones have been isolated from *Mikania* species.¹ The most prominent, mikanolide, was formulated as (1a) on the basis of chemical behaviour and extensive n.m.r. studies. A minor constituent, miscandenin, was assigned stereochemistry (3a) because of its probable derivation from the hypothetical common precursor (2) by way of a [3,3]sigmatropic (Cope) rearrangement. Previously studied, albeit simpler, examples of this rearrangement had produced one of the two

possible *trans*-fused elemadienes *via* a chair-like transition state depending upon the conformational restraints implicit in the initial germacradiene.²

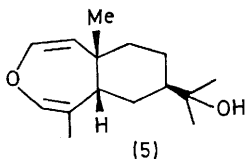
However, reconsideration of the Cope rearrangement which, as was pointed out earlier,¹ can also be looked upon as the homoelectrocyclic closure of a divinyloxiran to a dihydro-oxepin led us to the realization that the *cis*-fused divinyloxiran system and the configuration at C-6 impose limiting geometrical constraints on (2). These factors imply that miscandenin should be represented not by (3a) but either as (3b) [derived from (2) by way of rotation of the C-1,C-10 double bond through the ten-membered ring and closure *via* a boat-like transition state] or as (3c) [derived from (2) *via* a half-chair-like transition state]. The n.m.r. data of miscandenin¹ can, in fact, be accom-



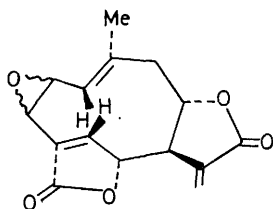
(1) a; 2,3-epoxide, *cis* and α
b; 2,3-epoxide, *cis* and β



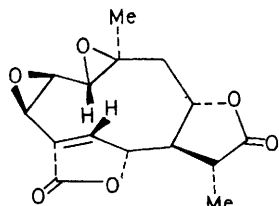
(3) a; β -Me, α -H
b; β -Me, β -H
c; α -Me, β -H



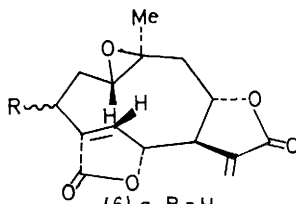
(5)



(2)



(4)



(6) a, R = H
b, R = β -OAc
c, R = α -OAc

modated more satisfactorily in terms of (3b) rather than (3c).

To settle the question, crystal structures of miscandenin and dihydromikanolide¹ have been determined in Glasgow. We find that these compounds have the stereochemistry depicted in (3b) and (4), respectively; the stereochemistry of mikanolide is therefore (1b) which differs from the original suggestion in the orientation of the C-2,3-epoxide. The stereochemistry of miscandenin is in accordance with the suggestion that (1b) and (3b) are formed from a common precursor (2) (C-2,3-epoxide, *cis* and β). The same ring junction stereochemistry has been attributed to occidenol (5).³ The results require no revision in the stereochemistry of deoxymikanolide (6a) but intimate that scandenolide¹ could be (6b) rather than (6c).

The X-ray intensity data for miscandenin and dihydromikanolide were measured on a four-circle diffractometer with Mo- K_{α} radiation and the crystal structures elucidated by direct methods. The atomic parameters were adjusted by least-squares procedures and the analyses are at $R = 0.070$ over 1222 reflections for (3b) and $R = 0.079$ over 1486 reflections for (4). The molecular conformation of dihydromikanolide is very similar to the germacranolide conformation described earlier for shiromodiol.⁴ The *trans* double bond in the ten-membered ring of (4) has a torsion angle of -163° and is thus appreciably distorted from ideal geometry.

W. H. acknowledges support by a U.S. Public Health Service Grant from the National Cancer Institute. P.J.C. and G.A.S. thank the S.R.C. for a grant to purchase the diffractometer.

(Received, 7th May 1973; Com. 648.)

¹ W. Herz, P. S. Subramaniam, P. S. Santhanam, K. Aota, and A. L. Hall, *J. Org. Chem.*, 1970, **35**, 1453.

² K. Takeda, *Pure Appl. Chem.*, 1970, **21**, 181; K. Takeda, I. Horibe, and H. Minato, *J. Chem. Soc. (C)*, 1970, 1142; K. Takeda, K. Tori, I. Horibe, M. Ohtsuru, and H. Minato, *ibid.*, p. 2697.

³ B. Tomita and Y. Hirose, *Tetrahedron Letters*, 1970, 235.

⁴ R. J. McClure, G. A. Sim, P. Coggon, and A. T. McPhail, *Chem. Comm.*, 1970, 128.