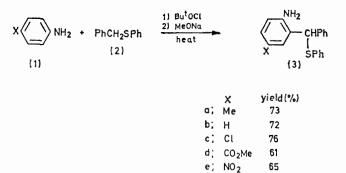
New Synthesis of Diphenylmethane Derivatives

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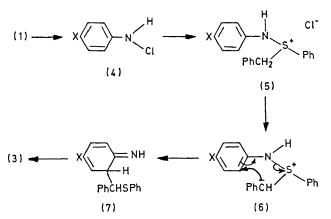
Summary A new, general synthesis of o-benzylanilines (2-aminodiphenylmethanes) from anilines has been developed, which involves the sequential reaction of an aniline with (i) t-butyl hypochlorite, (ii) phenyl benzyl sulphide, and (iii) sodium methoxide, followed by desulphurization with lithium aluminium hydride or Raney-nickel.

ALTHOUGH o-benzylanilines are useful synthetic intermediates (e.g., in the synthesis of fluorenes¹), the available methods² of preparation of o-benzylanilines have limitations. We have developed a new method for the o-alkylation of aromatic amines,³ and report here a modification of this



procedure which provides a general, high yield synthesis of o-benzylanilines.

In a general procedure, the aniline (1) (1 equiv.) and phenyl benzyl sulphide (2) (2 equiv.) were dissolved in a 3:1 mixture of MeCN-CH₂Cl₂ and the solution cooled to -40°. A solution of 1.3 equiv. of Bu^tOCl in CH₂Cl₂ at -78° was added dropwise and the reaction mixture was stirred for ca. 4 h at -40° and then allowed to warm to -20° over 3 h. A methanolic solution of sodium methoxide was then added and the reaction mixture was heated. Work-up and purification gave (3).



Mechanistically, the reaction involves initial chlorination of (1) to give (4) followed by reaction of (4) with (2) to give the azasulphonium salt (5). Treatment with base yields the ylide (6) which on heating undergoes a Sommelet-Hauser type rearrangement⁴ to give the dienone imine (7). Hydrogen shift and accompanying re-aromatization gave (3).

Desulphurization of (3) was readily accomplished via reduction with lithium aluminium hydride. When (3a), (3b), or (3c) was stirred with an equimolar amount of lithium aluminium hydride at 25° for 12 h, we obtained (8a) (89%), (8b) (95%), and (8c) (83%), respectively. Since lithium aluminium hydride desulphurization of (3d)

[†] All new compounds had satisfactory elemental analyses and spectra.

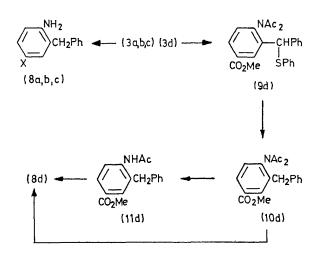
¹ W. D. Zahler and R. Huisgen, Chem. Ber., 1963, 96, 765. ² H. Nozaki, R. Noyori, and K. Sisido, Tetrahedron, 1964, 20, 1125; F. G. Mann and F. H. C. Stewart, J. Chem. Soc., 1954, 4127; H. J. Scheifele, jun., and D. F. DeTar, Org. Chem., 1952, 32, 8; H. Hellmann and W. Unseld, Annalen, 1960, 631, 82; O. Fischer and

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³ P. G. Gassman and G. D. Gruetzmacher, J. Amer. Chem. Soc., 1973, 95, 588.

 ⁴ M. Sommelet, Compt. rend., 1937, 205, 56; G. C. Jones and C. R. Hauser, J. Org. Chem., 1962, 27, 3572; G. C. Jones, W. O. Beard, and C. R. Hauser, *ibid.*, 1963, 28, 199; M. G. Burdon and J. G. Moffatt, J. Amer. Chem. Soc., 1966, 88, 5855; *ibid.*, 1967, 89, 4725; J. Doucet and A. Robert, Compt. rend., 1971, 272 A, 1562; P. Claus, Monatsh, 1971, 102, 913; P. Claus, N. Vavra, and P. Schilling, ibid., 1072.

⁵ J. J. Brown and R. K. Brown, Canad. J. Chem., 1955, 33, 1819.

could not be accomplished without reduction of the methoxycarbonyl group, an alternative desulphurization process was developed. Direct Raney-nickel reduction of (3d) was



complicated by extensive reduction of the activated aromatic ring. In order to avoid this difficulty (3d) was converted into (9d) according to the method of Brown.⁵ Reduction of (9d) with W-2 Raney-nickel in ethanolic tetrahydrofuran gave (10d) (93%). Treatment of (10d) with BF₃-MeOH in methanol removed the acetyl protecting groups to give (8d) in 79% overall yield from (3d). The selective removal of one acetyl group could be accomplished by treating (10d) with sodium methoxide in methanol under reflux for 12 h, which gave 91% of (11d).

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