Biosynthesis of Flavonoid and Terphenyl Metabolites by the Fungus Aspergillus candidus

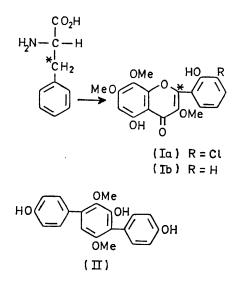
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Summary Chlorflavonin and two accompanying phenylalanine-derived metabolites, characterized from spectral evidence as dechlorochlorflavonin and 1,4-dimethoxy-2,4',4"-trihydroxy-p-terphenyl, are biosynthesized de novo by Aspergillus candidus.

FLAVONOIDS are generally accepted to be characteristic metabolites of higher plants,¹ and most accounts of their formation by micro-organisms have been either poorly documented² or subsequently found to be in error.³ A recent report on the isolation of the antibiotic chlorflavonin from *Aspergillus candidus* established its flavonoid structure (Ia), but did not distinguish between two possible origins: *de novo* synthesis or bioconversion of a flavonoid constituent in the corn steep nutrient. These alternatives have been tested with isotopically labelled substrates.

High (153%) specific incorporation of $[U^{-14}C]$ -D-glucose was observed. Exclusive labelling via methionine of the O-methyl groups is unlikely since $[Me^{-14}C]$ -L-methionine was a less efficient (32.7%) precursor than $[U^{-14}C]$ -Dglucose and both $[U^{-14}C]$ and $[3^{-14}C]$ -L-phenylalanine labelled the antibiotic (27.6 and 18.5% respectively). Degradation of chlorflavonin with alkali and decarboxylation of the recovered 3-chlorosalicylic acid⁴ established that label from $[3^{-14}C]$ -L-phenylalanine was located exclusively in C-3. We conclude that this unique chlorinated flavonoid is a true fungal metabolite.



We have isolated from cultures of A. candidus two additional metabolites biosynthetically labelled by $[U^{-14}C]$ -L-phenylalanine. One of these, $C_{18}H_{16}O_7$, m.p. 208° is probably dechlorochlorflavonin, (Ib) although an alternative structure with 5-hydroxy-6,7-dimethoxy substitution in ring A has not been excluded; λ_{max} (EtOH) 252 (inflex.), 265, 304 (inflex.), and 346 nm (log ϵ , 4.32, 4.41, 3.82 and 3.88);

 M^+ at m/e 344.0898, τ (60 MHz; CDCl_3; Me_4Si) -2.2 (1H, s, exchanges with D₂O, chelated OH), 1.97br (1H, s, exchanges with D₂O, unchelated OH), 2·15-3·1 (4H, complex m, ABCD system, ArH), 3.55 (1H, s, ArH), 6.04 (3H, s, OMe), and 6.13 (6H, s, OMe). Radioactivity from 14Clabelled (Ib) was incorporated irreversibly into (Ia).

The other metabolite, C20H18O5, m.p. 239° (decomp.), showed λ_{max} (EtOH) 225 (inflex.) and 275 nm (log ϵ , 4.06 and 4.06); M^+ at m/e 338.1168 with fragment ions at m/e245 $(M - C_6H_5O)^+$ and 154 $(M - C_6H_5O - C_6H_3O)^+$. The n.m.r. spectrum [100 MHz; (CD₃)₂CO; with Me₄Si] showed temperature-dependent hydroxy-signals at τ 1.63br (1H, s), 1.83br (1H, s), and 2.58 (1H, s); aromatic protons at τ 2.5 (2H, d, J 8.7 Hz) and 3.08 (2H, d, J 8.7 Hz) comprising one AA'BB' system, and at 2.76 (2H, d, J 8.8 Hz) and 3.17 (2H, d, J 8.8 Hz) comprising a second AA'BB' system. An aromatic proton at τ 3.53 (1H, s) was weakly coupled (J 0.5 Hz) to an aromatic methoxy-group at τ 6.32 (3H, s, 1 0.5 Hz) but was not coupled to a second aromatic methoxygroup at τ 6.63 (3H, s). The evidence is consistent with structure (II) for this metabolite. *p*-Terphenyl derivatives are rare in microfungi, the only other example being volucrisporin, a pigment of the Hyphomycete Volucrispora aurantiaca which is known to be derived biosynthetically from phenylalanine.5

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