A Convenient Synthesis of Alkylated 4-Hydroxy-2-pyrones

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Summary Various alkylated 4-hydroxy-2-pyrones were synthesized from the $\beta\delta$ -diketo-esters (11), (13), and (14), which were easily prepared from the $\gamma\delta$ -acetylenic β keto-esters (3), (4), and (7) via the corresponding enamino- β -keto-esters (8)-(10).

The present study was undertaken to establish a convenient method for the preparation of 3,6- and 5,6-dialkyl- and 3,5,6-trialkyl-4-hydroxy-2-pyrones corresponding to the pyrone units in luteoreticulin,¹ citreoviridin,² and aureothin.³ Syntheses of 3,6-4 and 5,6-dialkyl-4-hydroxy-2-pyrones⁵ have been carried out extensively in recent years, but no 3,5,6-trialkyl derivatives have been synthesized.

The Grignard reagents (1) and (2) reacted with a solution of ethyl methylmalonyl chloride in tetrahydrofuran at -20° to give the $\gamma\delta$ -acetylenic β -keto-esters (3) (65%) and (4) (85%), respectively, as viscous oils. The imidazolide of tetrolic acid reacted with the malonate magnesium enolates (5) and (6)⁶ in tetrahydrofuran to give the esters (3) (50%)and (7) (50%). Addition of diethylamine to (3), (4), and (7) in ether afforded the corresponding enamino- β -ketoesters, (8) (74%), (9) (48%), and (10) (41%). Hydrolysis of the enamine (9) with N-hydrochloric acid at 50° gave ethyl 2-methyl-3,5-dioxoheptanoate (11) (81%). Ring closure with polyphosphoric acid at 100° for 2 h then furnished 6-ethyl-4-hydroxy-3-methyl-2-pyrone (12) (51%), m.p. 182-183°. When the enamines (8) and (10) were heated with 3 equiv. of methyl iodide at 100° for 12 h, predominant C-alkylation was observed on the carbon atom of the enamine function. Without purification, the alkylated enamines were hydrolysed with boiling water for 1 h to give ethyl 2,4-dimethyl-3,5-dioxohexanoate (13) (54%) and ethyl 4-methyl-3,5-dioxohexanoate (14) (74%). Treatment of (13) with polyphosphoric acid at 100° for 1 h afforded 3,5,6-trimethyl-4-hydroxy-2-pyrone (15) (54%), m.p. 153-

 $R^{1}C \equiv C \cdot C \cdot O \cdot C + C O_{2}Et$ RC≡CMgBr (3) $R^1 = R^2 = Me$ (1) R = Me(2) R = Et (4) $R^1 = Et$, $R^2 = Me$ R^3 R^2 (7) $R^1 = Me$, $R^2 = H$ R¹CO·ĊH·CO·ĊH·CO,Et $(11) R^{1} = Et, R^{2} = Me, R^{3} = H$ $(13) R^1 = R^2 = R^3 = Me$ (14) $R^1 = R^3 = Me_1 R^2 = H$ OEt (5)R = Me(6) R = H $R^{1}C = CH \cdot CO \cdot CH \cdot CO_{2}Et$ NEt, (8) $R^1 = R^2 = Me$ (9) $R^{1} = Et, R^{2} = Me$ $(12) R^{1} = Et_{R}^{2} = Me_{R}^{3} = H$ (10) $R^1 = Me_1 R^2 = H$ $(15) R^{1} = R^{2} = R^{3} = Me$ (16) $R^1 = R^3 = Me_1 R^2 = H$

154°. 5,6-Dimethyl-4-hydroxy-2-pyrone (16) was prepared (50%) from (14) in the same way. The physical properties (m.p., i.r., and n.m.r.) of (16) were in good agreement with those cited in the literature.^{5a}

All new compounds gave satisfactory spectral and analytical data.

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