Formation of the 7-Triphenylphosphonio-norbornen-7-ylium Dication: a Non-classical Dication

By PIETER SCHIPPER,* WIM A. M. CASTENMILLER, JAN W. DE HAAN, and HENK M. BUCK (Department of Organic Chemistry, Eindhoven University of Technology, The Netherlands)

the analogous reaction with the 7-methoxynorbornen- (see Scheme). 7-ylium cation produces the 7-methoxynorbornen-7-yldetected.

unsaturated norbornyl systems, we tried to prepare the stituted products with stronger nucleophiles, like PPh₃, and 7-triphenylphosphonium salts, which could serve as pre- the latter products apparently decompose under the recursors of these intermediates. A similar method was action conditions.² These nucleophiles underwent similar successful in the oxidation of diphenylmethyltriphenyl- reactions with the norbornen-7-ylium cation, but reaction phosphonium methylide when the related radical and of 7-anti-chloronorbornene with PPh₃ in formic acid, where

Summary While the norbornen-7-ylium cation dissolved dipositive ion were formed.¹ We discuss here the preparain liquid SO₂ does not undergo 7-substitution with PPh₃, tion of compounds (III) and (IV) and the dipositive ion (V)

Attempts to prepare the norbornadien-7-ylphosphonium triphenylphosphonium cation; the intermediate 7-tri- salt by treatment of norbornadien-7-yl chloride or the phenylphosphonio-norbornen-7-ylium dication could be norbornadien-7-ylium cation with PPh₃ in liquid SO₂ failed. Only weakly nucleophilic phosphines like $P(C_6F_5)_3$ appeared to form 7-substituted salts. Probably, charge delocalisation In order to investigate the 7-cations and 7-radicals of in the norbornadien-7-ylium gives rise to 2- or 3-subequilibrium conditions are prevailing, does yield the 7-substituted product.

Stabilisation of charge by double-bond participation in norbornen-7-ylium cations decreases with increasing electron donating ability of the 7-substituents.³ Reaction of PPh₃ with 7-chloro-7-methoxynorbornene gives rise only to substitution of PPh₃ at the 7-position.

Reaction of 7,7-dimethoxynorbornene with PCl_5 yields a mixture of the isomers (I) (76%) and (II) (24%), which reacts readily with PPh₃ in liquid SO₂ (-60°) to yield a mixture (28:72) of the isomeric cations (III) and (IV). At -60°, the ratio of the isomers is constant, but at *ca.* -14°, compound (IV) isomerises to (III), which is obtained in crystalline form at room temperature. Both isomers (III) and (IV) produce the dication (V) in liquid SO₂ with HSO₃F-SbF₅ at -60°. Quenching of this dication with methanol yields only (III). The structures of compounds (I)--(V) were confirmed by their n.m.r. (¹H and ¹³C) spectra.

In solution in liquid SO_2 , isomers (I) and (II) undergo an exchange reaction at higher temperatures. This interconversion and the formation of a mixture of products in the reaction with PPh₃ indicate absence of double-bond participation in the intermediate 7-methoxy-norbornen-7-ylium cation.

First-order kinetics were observed for the isomerisation of compound (IV) to (III) in liquid SO₂ at -14° ($k = 2.7 \times 10^{-4} \, \mathrm{s^{-1}}$). In liquid SO₂-HSO₃F, compound (III) is stable, whereas compound (IV) decomposes at a rate comparable to the rate of isomerisation in liquid SO₂. During the decomposition no 7-methoxy-carbonium ion was observed, although this cation was shown to be stable under these conditions by adding (I) and (II). Thus, isomerisation of (IV) to (III) proceeds *via* the dication (V), which is not stable in liquid SO₂-HSO₃F. In this dication, double-



² P. Schipper and H. M. Buck, *Phosphorus*, to be published.

⁸ P. G. Gassman and A. F. Fentiman, jun., J. Amer. Chem. Soc., 1969, 91, 1545; 1970, 92, 2549; H. G. Richey, jun., J. D. Nichols, P. G. Gassman, A. F. Fentiman, jun., S. Winstein, M. Brookhart, and R. K. Lustgarten, *ibid.*, p. 3783; R. K. Lustgarten, M. Brookhart, and S. Winstein, *Tetrahedron Letters*, 1971, 2, 141.



bond participation, which is increased by the PPh_{3}^{+} group explains the irreversible formation of compound (III) (vide supra).

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