

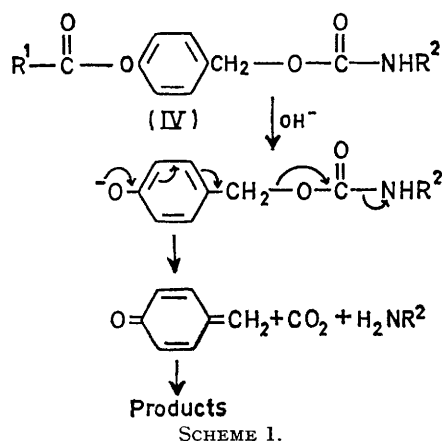
An Alkali-labile Substituted Benzyloxycarbonyl Amino-protecting Group

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Summary 4-Isopropylloxycarbonyloxybenzyloxycarbonyl, Z(4-PriOCO), a new amino-protecting group stable under the conditions which cause cleavage of the t-butoxycarbonyl and 2-(*p*-biphenyl)isopropylloxycarbonyl groups, can be removed in a slightly alkaline medium *via* a 1,6-elimination, by hydrogenolysis, or by HBr in acetic acid.

THE amino-protecting groups generally used in peptide synthesis are acid-labile. There has been a search for alkali-labile amino-protecting groups which can be removed selectively in the presence of the acid-labile groups. Carpino and Han¹ and Wunsch and Spangenberg² have used the 9-fluorenylmethoxycarbonyl and cyano-t-butoxycarbonyl groups which can be cleaved by base *via* a β -elimination reaction.



We set out to synthesise a substituted benzyloxycarbonyl group from which the amine could be regenerated by alkali via a 1,6-elimination mechanism involving a quinonemethide³ intermediate (Scheme 1). We prepared a series of crystalline glycine carbamates [(IVa; R¹ = Me) m.p. 91–92°, 50% yield; (IVb; R = EtO) m.p. 92–94°, 75%; (IVc; R = Pr¹O) m.p. 109–110°, 95%] as shown in Scheme 2.⁴ The chlorocarbonates (II) gave the stable crystalline *N*-methylimidazolium[†] derivatives (III)^{4†} which react rapidly with amino-acids in water.

We studied the stability of (IVa–c), and found that free glycine is quantitatively released under the conditions which cause cleavage of the benzyloxycarbonyl group (40% HBr in AcOH or H₂, Pd/C). The isopropylcarbonate (IVc) is the most stable in acid; it does not decompose in CF₃-CO₂H-CH₂Cl₂ which cleaves *t*-butoxycarbonyl-protected amines⁵ (see Table). This substituted benzyloxycarbonyl

TABLE

% Cleavage of Z(4-ROCO) amino-protecting group of glycine (reaction followed by quantitative determination of free glycine⁶ or by the change of absorbance at 290 nm)

Conditions (22°)	(IVa)	(IVb)	(IVc)
CF ₃ CO ₂ H-CH ₂ Cl ₂ ; 1–1, 1 h	10	2	<0.5
5% K ₂ CO ₃ , 1 h	100	80	35
0.1 N-NaOH, 10 min	100	100	100

† 4-Dimethylaminopyridine⁷ also gives stable derivatives but without improvement of the yield.

‡ (Ic) m.p. 61–62°, 80%; (IIc) ν_{CO} (liquid) 1760–1770 cm⁻¹, δ (CDCl₃) 5.29 (CH₂); (IIIc) m.p. 93°, ν_{CO} (Nujol) 1755 and 1785 cm⁻¹, δ (D₂O) 5.68 (CH₂), 87% yield from (Ic); (IIIa) m.p. 62°; (IIIb) m.p. 51°.

¹ L. A. Carpino and G. Y. Han, *J. Org. Chem.*, 1972, **37**, 3404.

² E. Wunsch and R. Spangenberg, *Chem. Ber.*, 1971, **104**, 2427.

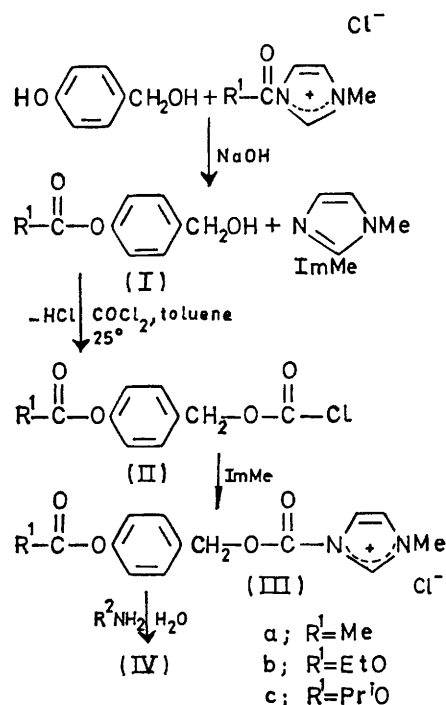
³ A. B. Turner, *Quart. Rev.*, 1964, **18**, 347.

⁴ E. Guibé-Jampel, G. Bram, and M. Vilkas, *Bull. Soc. chim. France*, 1973, 1021.

⁵ U. Ragnarsson, S. Karlsson, and G. Lindeberg, *Acta Chem. Scand.*, 1970, **24**, 2821.

⁶ J. Bartos, *Ann. pharm. franç.*, 1964, **22**, 383.

⁷ E. Guibé-Jampel and M. Wakselman, *Chem. Comm.*, 1971, 267.



group is easily cleaved in alkali. In 5% K₂CO₃ the reaction is slow but can be accelerated by the addition of 1 equiv. of hydrazine (98% cleavage after 2 h) and complete removal takes place rapidly in 0.1 N-NaOH.

By the same method, we have prepared other Z(4-Pr¹OCO)-amino-acids in good yields.

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