

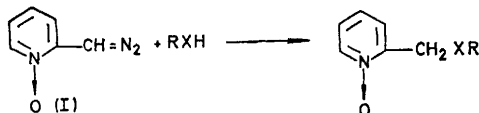
1-Oxidopyridin-2-yl-diazomethane: a Water-soluble Alkylating Agent for Nucleosides and Nucleotides

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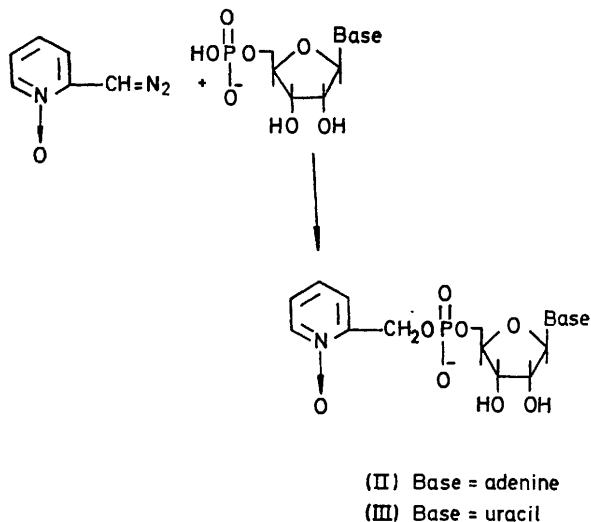
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Summary The water-soluble alkylating agent: 1-oxidopyridin-2-yl-diazomethane (I) introduces the 1-oxidopyridin-2-ylmethyl protecting group into acidic substances ($pK_a < 9.8$) including nucleotides; it can be removed by treatment with acetic anhydride followed by methanolic ammonia.

The development of procedures for the chemical synthesis of oligonucleotides depends to a significant extent on the design of a new protecting group with very specific properties.¹ Although diazomethane is useful for methylating reasonably acidic substances,² the methyl group is of no use as a protecting group, because of difficulties in its removal.³ We report here the synthesis of 1-oxidopyridin-2-yl-diazomethane (I) and its application in the protection of hydroxy-functions.†



2-Formylpyridine 1-oxide⁴ was converted into the corresponding *p*-tosylhydrazone,‡ m.p. 135–137° (50%), which was treated with NaOMe (1 equiv.) at 60°. Work up⁵ afforded compound (I) (30%, as CHCl₃ solution); λ_{max} (aq. MeOH) 557 nm; ν_{max} 2080 (N≡N⁺) and 1235 (N → O)



cm⁻¹. Compound (I) reacted rapidly with AcOH in CHCl₃ with evolution of nitrogen to afford 1-oxidopyridin-2-yl-methyl acetate, m.p. 67–68°, quantitatively. Results for other acidic substances are in Tables 1 and 2.

† The use of the 1-oxidopyridin-2-ylmethyl protecting group in polynucleotide synthesis has been discussed by Mizuno, *J. Org. Chem.*, 1972, 37, 39.

‡ Satisfactory elemental analyses were obtained for these compounds and those with m.p.s. listed herein.

In chloroform solution, (I) did not alkylate *m*-nitrophenol (pK_a 8.4), phenol (pK_a 10.0), and uridine (pK_a 9.8 and 12.34)⁶ In aqueous solution, prolonged treatment

TABLE 1

Reactions in CHCl₃ solution at 20° for 3 h^a

RXH	pK_a	M.p.	Yield
<i>p</i> -NO ₂ -C ₆ H ₄ -CO ₂ H ..	3.4	155–157°	Quant.
PhCO ₂ H	4.2	125–126°	Quant.
PhSH	6.5	98–100°	70%
<i>p</i> -NO ₂ -C ₆ H ₄ -OH ..	7.1	221–223°	71%

^a Satisfactory elemental analyses were obtained for all compounds.

TABLE 2

Alkylation in aqueous solution (pH 4.5) at 20° for 2 h

Compound	pK_a	Product	Yield
Adenosine 5'-phosphate ..	ca. 1	(II) ^a	84%
Uridine 5'-phosphate	ca. 1	(III) ^b	89%

^a Purified by DEAE-cellulose column chromatography. Enzymatic hydrolysis of the purified product (IV) with venom phosphodiesterase afforded adenosine 5'-phosphate and 1-oxidopyridin-2-ylmethanol, ratio 1:1. ^b Purified by DEAE-cellulose column chromatography. The structure was established by comparison (u.v. and *R_f*-values on paper electrophoresis) with an authentic sample prepared by a general method (including deacetylation) from 2',3'-di-*O*-acetyluridine 5'-phosphate and 1-oxidopyridin-2-ylmethanol with mesitylsulphonyl chloride as a condensing agent.

(20 h; room temperature) of uridine 5'-phosphate with excess of (I) afforded the protected derivative (IV) (85%) whose enzymatic hydrolysis with venom phosphodiesterase, followed by alkaline phosphatase treatment, afforded the protected uridine (V). These results coupled with those in

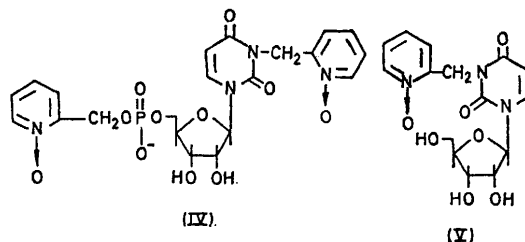
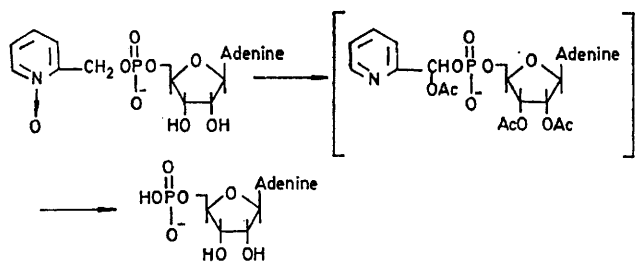


Table 1 indicate that in chloroform solution (I) could alkylate acidic substances with pK_a values less than ca 7.5.



In aqueous solution, however, the critical pK_a value increases to *ca.* 10.

Deblocking of (II) could be achieved by treatment with Ac_2O at 60° for 35 h (or 20° for 4 days), followed by methanolic ammonia (saturated ammonia, room temp., overnight). Recovery of adenosine 5-phosphate was 84%.

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¹ C. B. Reese 'Colloques Internationaux du C.N.R.S.,' Paris, 1970, No. 182, p. 319.

² L. E. Fieser and M. Fieser, 'Reagents for Organic Syntheses,' Wiley, New York, 1968, p. 191; A. Schonberg, 'Preparative Organic Photochemistry,' Springer-Verlag, New York, 1968, p. 275; R. Gompper, *Adv. Heterocyclic Chem.*, 1963, 2, 245.

³ J. F. W. McOmie, *Adv. Org. Chem.*, 1963, 3, 191; D. Lednicer, *ibid.*, 1972, 8, 179.

⁴ D. Jerchel and Heidler, *Annalen*, 1958, 613, 153; W. Mathes and W. Sauermilch, *ibid.*, 1958, 618, 152.

⁵ B. Eistert, W. Kurze, and G. W. Muller, *Annalen*, 1970, 732, 1; M. Regitz, *Chem. Ber.*, 1966, 99, 2918.

⁶ R. M. Izatt, J. H. Rytting, L. D. Hansen, and J. J. Christensen, *J. Amer. Chem. Soc.*, 1966, 88, 2641. The absence of alkylation of uridine in chloroform may have been due to its insolubility. Alkylation in aqueous solution is now being undertaken.