Synthesis of Pseudotropine

By Joseph J. Tufariello* and Eugene J. Trybulski

(Department of Chemistry, State University of New York at Buffalo, Buffalo, New York 14214)

involves a nitrone induced oxidative cyclization.

WE have recently reported1 the synthesis of supinidine using nitrones as key intermediates and would now like to report a novel entry into the tropane class of alkaloids2 based on a nitrone induced intramolecular cycloaddition. Thus, 4-nitrobut-1-ene,3 upon reaction with acrolein in

methanol containing sodium methoxide, followed by acidification of the medium with dry hydrogen chloride, affords nitro-acetal (I) [b.p. 64-68°/0.05 mm; i.r. (film) 6.1, 6.45, and 7.2 μ]. The nitro-acetal (I) was converted into the nitrone olefin (II) by treatment with zinc in aqueous ammonium chloride solution followed by acidification with hydrochloric acid. The nitrone olefin (II) [δ (CDCl₃, Me₄Si) $5.1 \text{ (m, 2)}, 5.6 \text{ (m, 1)}, 6.8 \text{ (m, 1)}; i.r. \text{ (film) } 6.1, 6.3 \mu$] was heated in toluene to afford cycloadduct (III) [m.p. 46-48°; δ (CDCl₃, Me₄Si) 1·2 (m, 2), 1·7—2·4 (m, 6), 3·5 (m, 2), 4·7 (t, 1, J 5 Hz)]. This reaction not only forms a carbon-

Summary A synthesis of pseudotropine is described which carbon bond essential to the completion of the synthesis but results in the concomitant and stereospecific oxidation of a centre destined to become the carbinol carbon of the natural product.

The isoxazolidine (III) was methylated with methyl iodide to give the quaternary salt (IV), which gives pseudotropine (V) upon reduction with lithium aluminium hydride. The pseudotropine so prepared is identical with an authentic sample from tropinone.4

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- J. J. Tufariello and J. P. Tette, Chem. Comm., 1971, 469.
 G. Fodor, in 'Chemistry of the Alkaloids,' ed. S. W. Pelletier, Van Nostrand Reinhold, New York, 1970, p. 431; G. Fodor in 'The Alkaloids,' vol. IX, ed. R. H. F. Manski, Academic Press, New York, 1967,p. 246.
 - M. H. Benn and M. G. Ettinger, Chem. Comm., 1965, 445.

⁴ R. Willstatter, Chem. Ber., 1896, 29, 936.