Hyperacidic Media.† A Novel Route to 13α-Steroids

By Jean-Claude Jacquesy,* Rose Jacquesy, Serge Moreau, and Jean-François Patoiseau (Laboratoire de Chimie XII, Faculté des Sciences, 86022 Poitiers, France)

Summary Treatment of pregnane-3,20-diones (1) by HF–SbF $_5$ or HSO $_3$ F–SbF $_5$ leads to a mixture of compounds (1)—(4); the postulated intermediate is an opened ion due to a " β cleavage" of the C(13)–C(17) bond.

Because of their pharmacological interest¹ we have been looking for a new, less tedious synthesis of the 13α -pregnanes.

We now report a high-yield, one-step synthesis of these compounds. Treatment of compounds (1a) or (1b) with hyperacidic media (HF-SbF₅ or HSO₃F-SbF₅) gives an equilibrated mixture of compounds (1)—(4) (Table).

The isomers are separated by column chromatography on silica gel; isomers (1a) and (2a) are best separated as their 3-monoacetates.

Treatment of a mixture of the 17α and β isomers of (5) gave a mixture which had lost no deuterium.

Thus the reaction does not appear to go through the intermediacy of a classical enol (cleavage of a C-H bond), but rather through an opened ion formed by a C-C bond cleavage² as shown in the Scheme. This opened ion may adopt several conformations leading, after reclosure, to the observed isomers. The proposed mechanism is consistent with the following observations:

- (i) The kinetic study shows that each of the isomers formed is a primary product of the reaction.
- (ii) The triketone (6) is recovered under the usual reaction conditions in agreement with the fact that a positive charge in the 13 position is destabilized by the proximity effect of the protonated 11 keto-function.

This reaction is limited to the acyclic ketone of the pregnane dione system: because of the free rotation around the C(17)—C(20) bond, some appropriate conformations exist in which the π -orbital of the protonated carbonyl is properly eclipsed with the bond to be broken.

This unprecedented reaction may be generalized to other flexible systems and permits, in a single step, isomerisations

† For previous parts in this series see J. C. Jacquesy and J. P. Gesson, Tetrahedron, in the press.

TABLE

Starting material	Reaction conditions	Yield (%)			
(1a), $(2a)$, or $(3a)$	HF-SbF ₅ ; a 0 °C; 7 h	(1a) 23·5	(2a) 67·5	(3a) 7·5	(4a) 1
"(1b)" or (3b)	HSO_3F-SbF_5 ; b 25 °C; 24 h $HF-SbF_5$; a 0 °C; 7 h	(1b) 22	(2b) 69	(3b) 8	(4b) 1

^a Molar ratio SbF₅: substrate 14:1; molar ratio SbF₅: HF 0·1:1. b 15% non-polar products are formed in this case.

SCHEME

in the α and/or β position of a carbonyl group without any exchange with the medium. New compounds described in this paper gave satisfactory elemental analyses.

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² D. M. Brouwer and H. Hogeveen, Rec. Trav. Chim. Pays-Bas, 1970, 89, 211; D. M. Brouwer and J. A. Van Doorn, ibid., 1971,

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