## **Regioselective Intramolecular Photocyclisation in the Prostaglandin Series**

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Summary Irradiation of  $PGA_2$  methyl ester (1) affords the cage compounds (2)—(4).

Here we report an example of intramolecular photochemical cycloaddition occurring with the  $PGA_2$  molecule, which provides optically active tricyclic cage compounds. Irradiation of  $PGA_2$  methyl ester (1) at  $-70^\circ$  for 90 min, in tetrahydrofuran solution, affords a mixture of three isomeric cycloadducts.

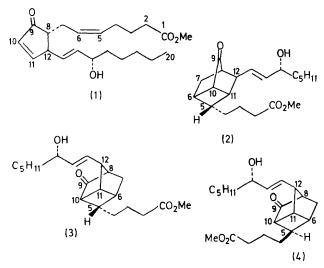
Preparative t.1.c. allows separation of the saturated ketone (2)<sup>+</sup> as the major component (18%). The structure of compound (2) is supported by the presence of an i.r. band at 1755 cm<sup>-1</sup>, typical of similar cyclopentanones,<sup>5</sup> the absence of intense u.v. absorption above 220 nm, and the n.m.r. spectrum which shows only two olefinic protons at  $ca. \delta 5.48$ , compared to six in (1). Moreover, the c.d. curve exhibits a very weak negative Cotton effect ([ $\theta$ ]<sub>306</sub> -460) because the molecule is almost completely symmetrical but for the four carbon chain at C-5 which seems to be located, at least partly, in a front octant.

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THE broad outlines of the reaction pathway of the addition of photoexcited cyclopentenones to alkenes have been elucidated,<sup>1,2</sup> and photochemical cycloadditions have been reported in numerous classes of natural products, including the prostaglandin family.<sup>3</sup> In addition, it has been shown that intramolecular enone-ene photocyclisations can lead to substituted cyclobutane derivatives.<sup>1,4</sup>

<sup>&</sup>lt;sup>‡</sup>Satisfactory elemental analysis and/or mass spectra were obtained for all new compounds. I.r., u.v., and n.m.r. spectra were in agreement with the proposed structures.

The second component (15%) is the keto-derivative (3). In this tricyclic compound the saturated ketone is located



in a six-membered ring, so that the i.r. carbonyl band and the methyl ester absorption appear in the same region. The ketone (3) presents only two olefinic protons and exhibits a positive c.d. curve ([ $\theta$ ]<sub>295</sub> + 5640), since the C-12 carbon is twisted in a positive octant.

The third component (8%) of the reaction is a stereoisomer of (3). Structure (4) is assigned to this substance. Its physical properties are reminiscent of these of its configurational isomer (3). Furthermore, as expected, compound (4) also presents a positive Cotton effect  $([\theta]_{295} +$ 5150).

The formation of the major compound (2) can be rationalised as resulting from an intramolecular photochemical addition process, due to the stereoproximity of the cis-5,6double bond<sup>6</sup> which adds to positions C-11 and C-10, respectively, from the  $\alpha$ -side of the PGA<sub>2</sub> methyl ester (1), through a stable biradical.<sup>2,7</sup>

Isomers (3) and (4) result from internal photoinduced  $\alpha$ -addition of the 5,6-cis-double bond to C-10 and C-11, respectively. The initial bond is formed between C-6 and C-11, and the diradical intermediate is rotationally equilibrated before ring closure.7§

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§ During the previous photochemical additions no internal cycloaddition was detected.<sup>3</sup> This may be due to the nature of the solvent,<sup>1</sup> the absence of external substrate with which to react, and/or to the fact that when the previously mentioned cycloadditions<sup>8</sup> were very fast processes, the intramolecular photochemical additions reported here are slower reactions, thus allowing the formation of the requisite biradicals.

<sup>1</sup> See: P. G. Sammes, Quart. Rev., 1970, 24, 37; P. de Mayo, Accounts Chem. Res., 1971, 4, 41; P. G. Bauslaugh, Synthesis, 1971, 2,

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<sup>4</sup> G. Ciamician and P. Silber, *Chem. Ber.*, 1908, **41**, 1928; G. Büchi and I. M. Goldman, *J. Amer. Chem. Soc.*, 1957, **79**, 4741. <sup>5</sup> See: J. L. Charltow, P. de Mayo, and L. Skattebøl, *Tetrahedron Letters*, 1965, 4697.

<sup>6</sup> A n.m.r. study of prostaglandin complex with tris(dipivalomethanato)europium(III) has shown that no isomerisation of the double bonds occurs during these photochemical reactions: G. A. Garcia, E. Diaz, and P. Crabbé, Chem. and Ind., 1973, in the press.

<sup>7</sup> Cf. W. L. Dilling, T. E. Tabor, F. P. Boer, and P. P. North, J. Amer. Chem. Soc., 1970, 92, 1399, and references cited.