Synthesis of 2β-Hydroxy-3,17-dioxoandrost-4-en-19-al and its Facile Aromatisation into Estrone

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Summary 2β -Hydroxy-3,17-dioxoandrost-4-en-19-al was synthesised and was shown to aromatise rapidly into estrone in the presence of water at neutral pH.

THE stoicheiometry of the biosynthesis of estrogens requires 3 moles each of oxygen and NADPH per mole of estrogen formed.¹ This is consistent with the participation of three enzymatic hydroxylations in the aromatisation process. Two of these have been identified to take place on the C-19 methyl group as the initial steps in the biotransformation of androgens into estrogens leading successively to the 19hydroxy- and 19-aldehyde-intermediates.² The nature and site of the third hydroxylation is presently unknown and the possibility that the C-2 position might be involved suggested the synthesis of the isomeric 2-hydroxy derivatives of the 19-hydroxy- and 19-oxo-androst-4-ene-3,17-diones as potential intermediates in the biosynthesis of estrone.

Reaction of 19-acetoxyandrost-4-ene-3,17-dione (1b) with N-bromosuccinimide provided 19-acetoxy- 6β -bromoandrost-4-ene-3,17-dione (2). When (2) was refluxed in glacial acetic acid for 12 min in the presence of potassium acetate,³ 2β ,19-dihydroxyandrost-4-ene-3,17-dione diacetate (3b) and 2α ,19-dihydroxyandrost-4-ene-3,17-dione diacetate (4b) were obtained. The $2\alpha\mbox{-isomer}$ (4b) was also obtained in superior yield by another independent route. Sulphuric

R'O R^2 n Ŕ4 (5) **a**; $R^1 = H$, $R^2 = OH$ (1) a; $R^1 = R^2 = R^3 = R^4 = H$ **b**; $R^1 = Ac$, $R^2 = R^3 = R^4 = H$ $R^1 = Ac$, $R^2 = R^3 = H$, $R^4 = Br$ **b**; $R^1 = OH$, $R^2 = H$ \mathbf{c} ; $\mathbf{R}^1 = \mathrm{OSiMe}_2 \mathrm{Bu}^t$, **a**; $R^1 = R^3 = R^4 = H, R^2 = OH$ $\tilde{R}^2 = \tilde{H}$ (3) **b**; $R^1 = Ac$, $R^2 = OAc$, $R^3 = R^4 = H$ c; $R^1 = R^3 = R^4 = H$, $R^2 = OSiMe_2Bu^t$ (4) **a**; $R^1 = R^2 = R^4 = H$, $R^3 = OH$ **b**; $R^1 = Ac$, $R^2 = R^4 = H$, $R^3 = OAc$

acid rearrangement⁴ of 4β , 5β -epoxy-19-hydroxyandrost-3,17-dione which in turn was obtained from the alkaline hydrogen peroxide oxidation of 19-hydroxyandrost-4-ene-

3,17-dione (1a) gave 2\alpha,19-dihydroxyandrost-4-ene-3,17dione (4a). Acetylation of (4a) gave the diacetate (4b) identical in all respects with that obtained from the 6bromo-compound (2). Brief alkaline hydrolysis of the other diacetate (3b) yielded 2β , 19-dihydroxyandrost-4-ene-3,17-dione (3a). The hydrolysis proceeded without rearrangement since reacetylation regenerated the diacetate (3b).

Oxidation of the 2α , 19-dihydroxy compound (4a) with the chromic oxide-pyridine complex gave 2a-hydroxy-3,17-dioxoandrost-4-en-19-al (5a). Similar oxidation of the epimeric (3a) failed to stop at the aldehyde stage and yielded a 2,19-lactone product. Since anchimeric assistance of the 2β -hydroxy group was clearly involved in the facile oxidation of the 19-hydroxy, access to the desired 19-aldehyde required masking of the 2β -hydroxy function in (3a). Reaction of (3a) with dimethyl-t-butylsilyl chloride in the presence of imidazole⁵ proceeded selectively give 2β , 19-dihydroxyandrost-4-ene-3, 17-dione 2-dito methyl-t-butylsilyl ether (3c). Oxidation of (3c) with the chromic oxide-pyridine complex provided 2β -hydroxy-3,17dioxoandrost-4-en-19-al dimethyl-t-butylsilyl ether (5c). Removal of the masking silvl ether was achieved with acetic acid in tetrahydrofuran⁵ to give the desired 2β -hydroxy-3,17-dioxoandrost-4-en-19-al (5b).

The 2β -hydroxy-19-aldehyde derivative (5b) in phosphate buffer pH 7 at room temperature was rapidly and quantitatively converted into estrone. In contrast the other 2-hydroxy derivatives (3a), (4a), and (5a) remained unchanged under these conditions. The known facts of estrogen biosynthesis including the β stereochemistry of C-2 hydrogen loss⁶ are accommodated by the 2β -hydroxy- Δ^{4} -3 keto-19-aldehyde structure (5b). The uniquely facile aromatisation of (5b) under physiological conditions suggests that it may be the enzymatic end product in estrogen biosynthesis with its nonenzymatic collapse to estrone as the final step in the pathway.

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