Structures of the Mating-type-specific Prohormones of Mucorales

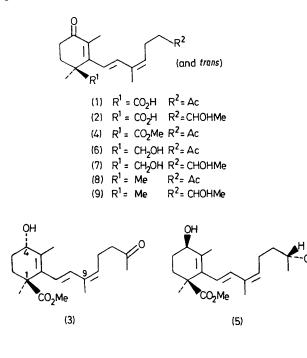
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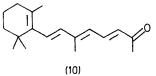
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Summary Specific prohormones from separate mating types of Blakeslea trispora are shown to have structures (3) (from plus) and (6)—(9) (from minus); each is specifically produced by one mating-type and only converted into the active hormones (1) and (2) by the other.

characterized as (3): C₁₉H₂₈O₄ by high-resolution m.s.; perturbed triene u.v. absorption $[\lambda_m(Et_2O) 276(sh)283 (log$ ϵ 4.18), 296 (sh) nm, unchanged by NaBH₄]; i.r. v_{max} (CHCl₃) 1725 (s, br:ester + saturated ketone >C = O), 3460 cm^{-1} (br: HO); $M^+m/e = 320$, changed to 322 by Na-BH4, to 362 by Ac2O-pyridine, and to 334 by transesterification with NaOEt. The n.m.r. spectrum of P+ was fully interpretable in terms of the predominantly 9-cis-structure (3) by reference to measurements for the known methyl 9-cis- and 9-trans-trisporate B (4) and for the $NaBH_4$ reduction product, mainly (5) (see below) from methyl trisporate C. The stereochemistry of (3) at C-1 follows from that of $(2)^7$ and at C-4 is tentatively assigned from m.s. data: given the preferred conformation of methyl trisporate C,⁷ the preferred product of its NaBH₄ reduction is predictably⁸ (5) in which the possibility of H-bonding between the cis-1,4 ester and OH groups brings the latter into an axial arrangement favouring dehydration; in the mass spectrum of (5) (and of related compounds), loss of the 4-OH as water is a very favoured process, whereas in the m.s. of natural

IN heterothallic mucoraceous fungi,^{1,2} isolated mycelia of *plus* or *minus* mating-type produce small amounts of specific 'prohormones' conveniently termed P⁺ and P⁻ respectively,^{3,4} such that only mycelium of the opposite mating-type will convert them into trisporic acids (1,2) which are the true hormones eliciting sexual differentiation in both mating-types.^{1,5} We report chemical characterizations of the prohormones. Culture media⁶ from *plus* Blakeslea trispora afforded 2—4 mg/1 of a complex mixture of neutral Et₂O soluble metabolites. Direct assays against minus Mucor mucedo⁵ of silica gel scraped from t.l.c. of the mixture localized active prohormone in one zone, with R_f and u.v. absorption agreeing with data ^{3,4} for P⁺. From 151 of culture medium, 1.6 mg of P⁺ was purified and





P+ this process is relatively much less prominent. The epimeric configuration shown in (3) is therefore suggested.

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Similar work on the complex of neutral metabolites from minus B. trispora confirmed that materials with P- prohormone activity occur in two well-separated chromatographic regions,³ both heterogenous. From the more polar zone we obtained the trisporols (6, 7) and from the less polar zone the corresponding gem-dimethyl compounds (8, 9). These are all known compounds⁵ and were characterized by u.v., m.s., and chromatographic data. In bioassays (with plus M. mucedo) the relative activities of (6), (7), and (8 and 9) are approximately as 50:10:1.

Labelling experiments confirm that the prohormones, like (1) and (2) in mixed plus-minus cultures,⁹ are formed via the C-18 ketone (10). In the single strains, the overall synthesis of (1, 2) is repressed, but some steps still occur though at a minimal (un-induced, or 'gratuitous') level. We have shown² that distinctive features in the resulting complex mixtures of minor metabolites arise because the pattern of gratuitous reactions is different in the two mating-types. The structures now assigned to the active components of these mixtures, (3) in plus and (6)—(9) in minus, accord with that analysis since both their specific production from (10) by one mating-type and their specific transformation into (1) and (2) by the other are consistent with what is known ^{2,5} about the patterns of the matingtype-linked gratuitous reactions.

Recent observations¹⁰ on some Mucor species in which the copulatory outgrowths are aerial suggested that the volatile 'zygotropic hormones' of these species are in effect the P+ and P- prohormones. This too accords with the structures now assigned, since all are much less polar than the hormones proper, (1) and (2), and their volatility at room temperature, though slight, is experimentally demonstrable.

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