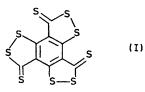
X-Ray Crystal Structure of Benzo[1,2-c; 3,4-c'; 5,6-c"]tris[1,2]dithiole-1,4,7-trithione: a Sulphur Analogue of Coronene

By LARS K. HANSEN and ABSJØRN HORDVIK*

(Chemistry Section, University of Tromsø, P.O. Box 790, 9001 Tromsø, Norway)

Summary The sulphocarbon benzo[1,2-c; 3,4-c'; 5,6-c''] tris[1,2]dithiole-1,4,7-trithione has a coronene-like structure in the crystalline state with S-S bond lengths in the range 2.420 (8)-2.433 (8) Å.



THE sulphocarbon (I) has been synthesized by two methods.^{1,2} The structural formula of (I) suggests that there may be delocalized σ -bonding in the linear three-sulphur sequences analogous to that in the three-sulphur sequence

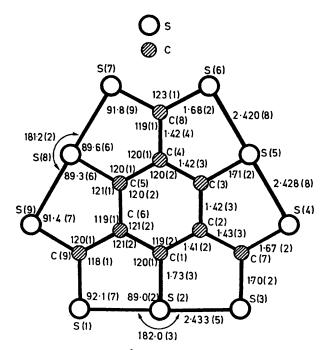


FIGURE. Bond lengths (Å) and bond angles (°) in $S_{\mu}C_{\mu}$.

of 6a-thiathiophthen.³ However, attempts to find experimental evidence for this assumption have failed.² The molecular structure of (I) has now been studied by

The six fused disulphide rings of (I) are equal within experimental error, cf. Figure, and the symmetry of the molecule is approximately D_{3h} . Furthermore, compound (I) has a 24π -electron system and may therefore be described as a sulphur analogue of coronene.^{4,5} The C-C bonds in the central ring of (I) as well as those in the central ring of coronene^{4,5} are found to be somewhat longer than the aromatic C-C bond, 1.397 Å, in benzene.

The S-S distances in (I) indicate that there is delocalized bonding in the linear three-sulphur sequences of the molecule. However, the S-S distances in (I) of 2.43 Å are longer than those of 2.36 Å in 6a-thiathiophthen.³

In crystals of (I) there are a series of intermolecular sulphur-sulphur contacts in the range 3.47-3.70 Å. The distance between the planes of adjacent molecules is 3·47 Å.

The crystals of C₉S₉ are brown-red, monoclinic, space group I2/c with unit cell dimensions a = 7.485(4), b =10·096(6), c = 15.792(5) Å, $\beta = 90.87(4)^{\circ} Z = 4$, $D_c = 2.21$ g cm⁻³, $D_m = 2.2$ g cm⁻³. Using Mo- K_{α} radiation, 437 out of 900 independent

reflections in the range $0 < \theta < 23^\circ$, for which $I \ge 2\sigma(I)$ were accepted as observed. The structure was solved by the heavy atom (S) method and refined by full-matrix leastsquares. With anisotropic temperature coefficients for sulphur and isotropic for carbon the final R is 0.086.

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