Intramolecular C-Alkylation in $\beta\gamma$ -Unsaturated Diazo Ketones : A New Synthetic Route to Angularly fused Cyclobutanones and γ -Lactones

By Usha R. Ghatak^{*} and Baijayanti Sanyal (née Maitra)

(Department of Organic Chemistry, Indian Association for the Cultivation of Science, Jadavpur, Calcutta-700032, India)

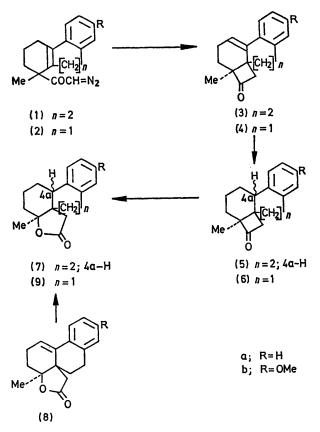
Summary A new synthesis of angularly fused cyclobutanones by the acid-catalysed intramolecular C-alkylation of $\beta\gamma$ -unsaturated α -diazomethyl ketones is described along with their transformation into γ -lactones.

RECENTLY a new annelation reaction of synthetic potentiality has been developed^{1,2} for the introduction of a bicyclo-[3,2,1]octanone group into a number of monocyclic and polycyclic systems by intramolecular acid-catalysed cyclisation of the corresponding $\gamma\delta$ -unsaturated α -diazomethyl ketones. We now present some novel examples in which an angularly fused four-membered ring ketone is formed in high yields by π -bond participation in the displacement of nitrogen from the protonated diazomethyl carbonyl function in a few $\beta\gamma$ -unsaturated α -diazomethyl ketones. To our knowledge, these represent the first reported intramolecular electrophilic addition reaction of diazo-ketones leading to four-membered ring ketones.^{3,4}

The diazo ketone (1a), † m.p. 104-106°, prepared quantitatively from the corresponding acid,⁵ in CHCl₃ (0·1-0.05 M) rapidly loses nitrogen when treated with 2 to 4 equivalents of 70% aqueous $HClO_4$ or 48% aqueous HBF_4 or conc. H_2SO_4 at 5-10° to give the unsaturated cyclobutanone (3a), m.p. 91–92° [ν_{max} (CHCl₃) 1765 cm⁻¹; λ_{max} (EtOH) 255 nm (log ϵ 4.07); δ (CDCl₃) 1.20 (3H, s) and 6.37 (1H, br t, J 4Hz)] (80-99%). Reaction of the diazoketone (1a) in CHCl₃ with 57% HI, under Wolfrom reduction conditions⁶ again gave the cyclobutanone (3a) (60%), as the only isolable product after chromatography. The methoxy-analogue (1b), ikewise with HClO₄ or HI, produced the cyclobutanone (3b), m.p. 102-103° (50-70%) The hydrofluorene derivative (2a),8 m.p. 102-103° with $HClO_4$ or HI afforded the strained cyclobutanone (4a), m.p. 77—78° [ν_{max} (CHCl₃) 1768 cm⁻¹; λ_{max} (EtOH) 255 (log ϵ 4·12) and 290 nm (log ϵ 3·68); δ (CDCl₃) 1·30 (3H, s), 6.26 (1H, br t, J 5 Hz)] (50-55%). Compared to (3) this cyclobutanone is relatively unstable at room temperature.

Catalytic hydrogenation of the styrenoid bond in the ketones (3a) and (3b) [Pd-C (10%) in EtOH] gave the corresponding saturated ketones (5a), m.p. 102° and (5b),

m.p. 100—101°, (90%). N.m.r. spectral and t.l.c. properties of these ketones indicated their stereohomogeneities. The stereochemistry at the C-4a asymmetric centre shown in (5) is based upon the least-hindered approach of the catalysed surface from the opposite phase of the rigidly bound cyclobutanone ring in the olefins (3), as observed with other rigid systems.² Catalytic hydrogenation of the relatively planar unsaturated ketone (4a), however, pro-



† Compounds described here are all racemates; satisfactory analytical and spectral data were obtained for new compounds.

duced a mixture of the diastereoisomers (6a) in ratio of ca. 80:20 (n.m.r.). The stereochemistry of the main stereoisomer of (6a), m.p. 79-80°, separated from the mixture (ca. 60%) could not be assigned at this stage.⁹

Baeyer-Villiger oxidation of the saturated ketone (5a) with m-chloroperbenzoic acid in CH2Cl2 produced the γ -lactone (7a), m.p. 173—174° [ν_{max} (CHCl₃) 1750 cm⁻¹; δ (CDCl₃) 1.42 (3H, s)] (68-72%). The same lactone was also obtained as the sole product through hydrogenation [Pd-C (10%) in EtOH] of the corresponding unsaturated lactone (8a), m.p. 149° [ν_{max} (CHCl₃ 1755 cm⁻¹; λ_{max} (EtOH) 250 nm (log ϵ 4·18); δ (CDCl₃) 1·42 (3H, s) and 6.20 (1H, br t, J 5 Hz)], which was prepared (90%) by alkaline hydrogen peroxide oxidation of the unsaturated ketone (3a). Similarly the methoxy analogues (5b) and (3b) were converted into the corresponding γ -lactones (7b),

m.p. 177-178° and (8b) m.p. 166-167°. The stereochemical results in the hydrogenation of rigid unsaturated lactones (8a) and (8b) to the corresponding saturated lactones support the stereostructures assigned to (5a) and (5b). The saturated ketone (6a), m.p. 79-80° also gave the corresponding γ -lactone (9a) m.p. 122° [ν_{max} (CHCl₃) 1755 cm⁻¹, δ (CDCl₃) 1.42 (3H, s)] (80%) through oxidation with *m*-chloroperbenzoic acid in CH_2Cl_2 .

The present investigations thus provide a simple synthetic route to angularly fused polycyclic cyclobutanones¹⁰ which have potentialities, particularly for stereospecific introduction of angular substituents¹¹ in polycyclic systems.

We thank Professor A. K. Bose and Dr. Goutam Gupta for some of the n.m.r. spectra.

(Received, 17th July 1974, Com. 884.)

¹ R. Malharbe, Helv. Chim. Acta, 1973, 56, 2845.

² P. N. Chakrabortty, R. Dasgupta, S. K. Dasgupta, S. R. Ghosh, and U. R. Ghatak, Tetrahedron, 1972, 28, 4653.

* The cyclobutanone formation by intramolecular C-H insertion in the Wolff rearrangement and copper catalysed decomposition of ⁶ The cyclobutatione formation by intramolecular C-H insertion in the wolff rearrangement and copper catalysed decomposition of diazoketone have been recorded see, e.g., J. P. Tresca, J. L. Fourrey, J. Polonsky, and E. Wenkert, *Tetrahedron Letters*, 1973, 895.
 ⁶ An example of the formation of a four membered oxa-ketone has been reported in a base-catalysed conversion of 1-ocetoxy-1-diazo-acetylcyclohexane into 1-oxaspiro[3,5]nonan-3-one: J. R. Marshall and J. Walker, *J. Chem. Soc.*, 1952, 467.
 ⁶ U. R. Ghatak, N. R. Chatterjee, A. K. Banerjee, J. Chakravarty, and R. E. Moore, *J. Org. Chem.*, 1969, 34, 3739.
 ⁶ M. L. Wolfrom and R. L. Brown, *J. Amer. Chem. Soc.*, 1943, 65, 1516.
 ⁷ Diazo-ketone was prepared from the corresponding acid; U. R. Ghatak and N. R. Chatterjee, *Indian J. Chem.*, 1971, 9, 804.
 ⁸ Diazo-ketone was prepared from the corresponding acid; U. R. Ghatak R. Dasgunda and J. Chakravarty. *Tetrahedron*, 1074, 20

⁸ Diazo-ketone was prepared from the corresponding acid; U. R. Ghatak, R. Dasgupta, and J. Chakravarty, Tetrahedron, 1974, 30, 187

⁹ See for examples: H. O. House, R. G. Carlson, H. Müller, A. W. Noltes, and C. D. Slater, J. Amer. Chem. Soc., 1962, 84, 2614;
¹⁰ B. M. Trost and M. J. Bogdanowicz, J. Amer. Chem. Soc., 1973, 95, 5321, and refs. therein.
¹¹ U. R. Ghatak and S. Chakrabarty, J. Amer. Chem. Soc., 1972, 94, 4756.