

Polar Substituents in Pericyclic Reactions: Photochemistry of 2-Cyanobenzobarrelene

By CHRISTOPHER O. BENDER* and SYDNEY S. SHUGARMAN

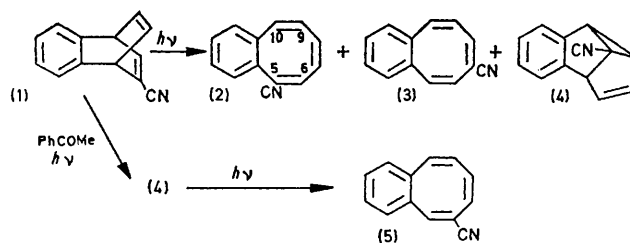
(Department of Chemistry, University of Lethbridge, Lethbridge, Alberta, Canada)

Summary Deuterium labelling studies show that 7-cyanobenzocyclo-octene, the major product from the direct irradiation of 2-cyanobenzobarrelene, derives from initial $2\pi + 2\pi$ bonding between the vinyl and vinyl-cyano units.

INTEREST in the effects of polar substituents on pericyclic processes has been heightened recently by the configuration-interaction calculations of Epiotis¹ which indicate that for some polar $4n$ systems reaction in *anti*-Woodward-Hoffmann² mode may occur. We report here the powerful influence of the cyano-group on the mechanistic course of the photochemical transformation of benzobarrelene into benzocyclo-octene.

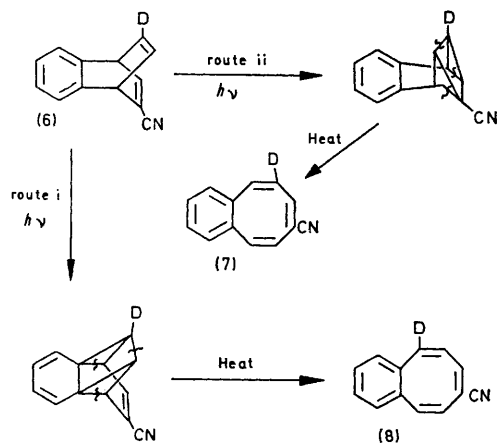
2-Cyanobenzobarrelene (1)[†] was prepared by the treatment with CuCN of 2-bromobenzobarrelene, isolated from the addition of benzyne to bromobenzene. Direct irradiation (450-W medium-pressure Hanovia lamp; Corex filter;

cyclohexane) of (1) led cleanly to three isomeric primary photoproducts: (2) (17%), (3) (54%), and (4) (29%). Under similar conditions, direct irradiation of (4) gave (5). Upon triplet sensitization (Pyrex filter, acetophenone), (1) yielded only (4). Thus (2) and (3) seem to derive from the singlet excited state of (1).



[†] All new compounds were characterized by i.r., u.v., and n.m.r. spectra, and gave combustion analyses in agreement with the proposed structures.

The photochemical formation of cyclo-octenes from barrelenes is generally considered to proceed *via* an initial concerted $2_{\pi} + 2_{\pi}$ cycloaddition (*e.g.*, Scheme).³



SCHEME

Whereas there is only one possible $2_{\pi} + 2_{\pi}$ cycloaddition route to (2), that involving initial bridging between the benzo- and vinylcyano-units, (3) could be formed *via* two concerted pathways: initial benzo-vinyl bridging (route i, Scheme) or initial vinyl-vinylcyano-bridging (route ii).

¹ N. D. Epiotis, *J. Amer. Chem. Soc.*, 1972, **94**, 1941.

² R. B. Woodward and R. Hoffmann, 'The Conservation of Orbital Symmetry,' Verlag Chemie-Academic Press, 1970.

³ H. E. Zimmerman, R. S. Givens, and R. M. Pagni, *J. Amer. Chem. Soc.*, 1969, **90**, 6096.

In order to distinguish between these two possibilities the deuterium labelled material [(6) 93% $^2\text{H}_1$] was synthesised from *para*-deuteriobromobenzene and irradiated directly. The n.m.r. spectrum of the deuteriated 7-cyano-product revealed that it was exclusively the β -labelled compound (7). Thus only the vinyl-vinylcyano-route is used in the formation of (3) from (1).

The presence of the cyano-group in the vinyl bridge of benzobarrelene dramatically changes the initial bonding preference. For benzobarrelene itself Zimmerman and his co-workers³ have shown that the benzo-vinyl bonding pathway predominates to the extent of $94 \pm 2\%$. In the present case benzo-vinyl bridging is now the minor pathway, and that exclusively to the substituted vinyl unit. The $2_{\pi} + 2_{\pi}$ photocycloadditions of (1) are thus not controlled by the polar *AX* mechanism.¹ Epiotis has reasoned that for *AX* $2_{\pi} + 2_{\pi}$ processes, reaction from the first singlet excited state should show *s + a* stereoselectivity. Consequently, for the rigid barrelene system, where geometry demands $2_{\pi} + 2_{\pi}$ cyclizations to occur only in the *s + s* mode, bonding to the polar vinylcyano-bridge should be the least favoured.

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