## The Synthesis and Structure of 1,1,1-Trimethylhydrazinium 3-Methoxycarbonyl-pyrazole-5-carboxylate, a Typical Novel Intermediate for Acid-ester Synthesis

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Summary 1,1-Dimethylhydrazine was observed to monodemethylate a dimethyl ester at room temperature to give a 1,1,1-trimethylhydrazinium-ester (II), whose subsequent treatment with acid gave an acid-ester (III); a single crystal X-ray analysis of (II) was carried out.

Synthesis of compounds containing both an acid and an ester group (acid-esters) is tedious, particularly when the functional groups necessary for the formation of an anhydride, a common precursor for acid-esters, are not vicinal. We now report a new, simple acid-ester synthesis in which a dimethyl ester (I) is monodemethylated with excess 1,1-dimethylhydrazine, and the resulting 1,1,1-trimethylhydrazinium ester (II) converted to the corresponding acid-ester (III) (Scheme). The reaction was applicable to aliphatic, alicyclic, aromatic, and heterocyclic systems.

Half acid-esters of glutaric 1,2-cyclohexane, benzene, and 3,5-pyrazole were synthesized. The yield of the intermediate salts varies from 25—60% depending on the nature

‡ Details of the X-ray analysis will be published in Acta Cryst.

of the starting diester. The yield of acid-esters formed by adding acid to the salt was almost quantitative.

$$\begin{array}{c} \stackrel{H}{\stackrel{N}{\stackrel{}}} \stackrel{CO_2Me}{\stackrel{}} \stackrel{\bullet}{\stackrel{}} \stackrel{H_2NNMe_2}{\stackrel{room temp.}{\stackrel{}}} \stackrel{\bullet}{\stackrel{}} \stackrel{N}{\stackrel{}} \stackrel{OO_2H}{\stackrel{}} \stackrel{NNMe_3}{\stackrel{}} \stackrel{\stackrel{\bullet}{\stackrel{}}}{\stackrel{}} \stackrel{N}{\stackrel{}} \stackrel{NO_2H}{\stackrel{}} \\ \stackrel{CO_2Me}{\stackrel{}} \stackrel{(III)}{\stackrel{}} \stackrel{(IIII)}{\stackrel{}} \stackrel{(III)}{\stackrel{}} \stackrel{(III)}{\stackrel{}} \stackrel{(III)}{\stackrel{}} \stackrel{(III)}{\stackrel{}} \stackrel{(III)}{\stackrel{}} \stackrel{(III)}{\stackrel{}} \stackrel{(III)}{\stackrel{}} \stackrel{(III)}{\stackrel{}} \stackrel{(IIII)}{\stackrel{}} \stackrel{(III)}{\stackrel{}} \stackrel{(IIII)}{\stackrel{}} \stackrel{(III)}{\stackrel{}} \stackrel{(III)}{\stackrel{III}{\stackrel{}} \stackrel{(III)}{\stackrel{}} \stackrel{(III)}{\stackrel{}} \stackrel{(III)}{\stackrel{}} \stackrel{(III)}{\stackrel{III}{\stackrel{}} \stackrel{(III)}{\stackrel{}} \stackrel{(III)}{\stackrel{III}{\stackrel{}} \stackrel{(III)}{\stackrel{III}{\stackrel{}} \stackrel{(III)}{\stackrel{III}{\stackrel{}} \stackrel{(III)}{\stackrel{III}{\stackrel{}} \stackrel{(III)}{\stackrel{III}{\stackrel{}} \stackrel{(III)}{\stackrel{III}{\stackrel{}} \stackrel{(III)}{\stackrel{III}{\stackrel{}} \stackrel{(III)}{\stackrel{III}{\stackrel{III}{\stackrel{III}{\stackrel{}} \stackrel{(III)}{\stackrel{III}{\stackrel{\stackrel{III}{\stackrel{III}{\stackrel{III}{\stackrel{III}{\stackrel{III}{\stackrel{III}{\stackrel{III}{\stackrel{III}{\stackrel{III}{\stackrel{$$

A preliminary characterisation of (II) was carried out by elemental, i.r. and n.m.r. spectral analyses. [ $\delta$  (CF<sub>3</sub>CO<sub>2</sub>H) reference 2,2,3,3-tetradeuterio-3-(trimethylsilyl) propionate, 3.45 (NMe<sub>3</sub>), 4.05 (OMe), 7.6 (arom. H);  $\nu_{max}$  (KBr) 1730 (ester C=O), 1580 cm<sup>-1</sup> (carboxylate C=O)]. Unequivocal structural proof of (II) was given by X-ray crystallography.‡

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