Synthesis of a Dialkyl Phosporoisocyanidate

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Summary The first compound with the >P(O)NC structure and its rearrangement to a dialkyl phosphorocyanidate, as shown by i.r. and ³¹P n.m.r. spectroscopy, are described.

The synthesis of diaryl phosphoroisoselenocyanatidates has been described only recently.¹ We find that the reaction

of equimolar amounts of the dioxaphosphorinan² (I), δ (³¹P) + 1.8 p.p.m. (MeCN, H_8PO_4 ext. standard; +ve shift to higher field) with KSeCN in dry MeCN solution gives the isoselenocyanatodioxaphosphorinan (II), m.p. 64—67° (from benzene-n-heptane, 1:1) δ (³¹P + 29.0 p.p.m. (C₆H₆), v_{NCSe} (KBr) 1295br,s and v_{PO} 1305br,s cm⁻¹.† Treatment of (II) with triethyl phosphite in benzene

 \dagger All operations with (II), (IV), and (V) have to be carried out under dry nitrogen under scrupulously dry conditions. Compounds (II), (IV), and (V) gave mass spectra fully consistent with their postulated structures.

caused cleavage of the C=Se bond³ with formation of the phosphoroselenoate⁴ (III) and the isocyanodioxaphosphorinan (IV), the first compound with the >P(O)NCstructure, m.p. $70.5-72^{\circ}$ (from ether-n-heptane, 1:1),



 $\delta~(^{31}{\rm P})~+~34{\cdot}5$ p.p.m. $({\rm C_6H_6})~\left|{}^1J~(^{31}{\rm P}{-}^{14}{\rm N})\right|~~13{\cdot}9$ Hz, $\nu_{\rm NC}$ (C_6H_6) 2080sh,m and v_{PO} 1340br,s cm⁻¹. Compound (IV) is very reactive and unstable and readily isomerizes to the cyano-compound (V), m.p. 79–81°, δ (³¹P) + 28.5 p.p.m. (C_6H_6) , v_{CN} (C_6H_6) 2210sh,m and v_{PO} 1330br,s cm⁻¹. Compound (V) was obtained independently by reaction of the dioxaphosphorinan (VI)⁵ with BrCN.[‡] The rearrangement (IV) \rightarrow (V) was followed by i.r. and ³¹P n.m.r. spectroscopy.

The proton-noise decoupled ³¹P n.m.r. spectrum of (IV) shows a 1:1:1 triplet, resulting from the coupling of ³¹P with ¹⁴N (I = 1). Such splitting, to our knowledge, has not been observed in the spectra of other >P(O)-N< compounds and strongly supports the presence of the NCgroup.6

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‡ Yamada et al. (S. Yamada, Y. Kasai, and V. Shioiri, Tetrahedron Letters, 1973, 1595) have claimed on the basis of i.r. data that in the reaction of triethyl phosphite with BrCN both phosphoro-cyanide and -isocyanide are formed and after several weeks at room temperature the isocyanide had decomposed. Our ³¹P n.m.r. results showed that the reaction of 2-methoxy-5,5-dimethyl-1,3,2-dioxaphosphorinan (R. S. Edmundson, Tetrahedron, 1964, 20, 2781) with an equimolar amount of BrCN in benzene solution gave a mixture of the 2-cyano-compound (V) and the corresponding 2-bromo-compound $\delta^{(31}P) + 14\cdot7$ p.p.m. (C₆H₆), (W. J. Stee and A. Zwierzak, *Canad. J. Chem.*, 1967, 45, 2513), in a 57:43 ratio. Traces of (IV) (< 5%) were evident from the ³¹P n.m.r. spectrum of the reaction mixture recorded immediately after the exothermi creaction of (VI) with BrCN.

- ¹ T. Gabrio and G. Barnikow, Z. Chem., 1969, 9 (5), 183.
- ² R. S. Edmundson, Tetrahedron, 1968, 21, 2379.
 ³ L. I. Stangeland, T. Austad, and J. Songstad, Acta Chem. Scand., 1973, 27, 3919.
- ⁴ W. J. Stec, A. Okruszek, B. Uznanski, and J. Michalski, Phosphorus, 1972, 2, 95.
- ⁵ A. Zwierzak, Canad. J. Chem., 1967, 45, 2502.
 ⁶ 'Isonitrile Chemistry,' ed. I. Ugi, Academic Press, New York, London, 1971, p. 5.