

Silyl Esters of a Five-membered Cyclic Acyl Phosphate

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Summary The first example of a silyl ester of a five-membered cyclic acyl phosphate is described.

WE describe the first example of a silyl ester of a five-membered cyclic acyl phosphate (3), which was prepared from compound (1)¹ by the following procedure. A mix-

TABLE 1
¹H N.m.r. and i.r. spectral data for compounds (1)–(3)^a

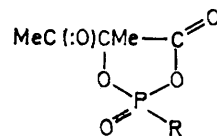
		$\tau(\text{Ac})$	$\tau(\text{MeC})$	Other ¹ H signals	$\nu_{\text{C-O}}/\text{cm}^{-1}$
(1)	<i>cis</i>	7.62	8.23	6.02 ^c	1748; 1838
	<i>trans</i>	7.66	8.13	6.02 ^c	—
(2)		7.60	8.33	5.43; 7.30 2.06; 4.0.86 ^d	—
(3)		7.60	8.16	9.60	1740; 1830

^a CDCl₃ solutions at 24°, Me₄Si internal standard, the integrated intensities are in the expected ratios. ^b Acetyl and phosphoryl oxygen *cis*. ^c *J* 12.0 Hz. ^d *J* 7 Hz.

ture of compound (1) (43.3 mmol) and γ -picoline (47.5 mmol) in CH₂Cl₂§ (35 ml) was stirred at 24° for 72 h. After evaporation (20°; 12 mmHg) and washing with benzene, the

kept at –25° until crystallization occurred. Filtration afforded pure (3) (70%).

The salt (2) and its esters, (1) and (3) are powerful



(1) R = OMe

(2) R = O[–]MeN⁺(Me)₂

(3) R = OSiMe₃

TABLE 2. ¹³C N.m.r. spectra of (1) and (3)

		$\delta(^{13}\text{C})/\text{p.p.m. vs. Me}_4\text{Si} = 0 (J_{\text{C-P}}/\text{Hz})$					
		Me-C-O	Me-C:O	C(:O)C(C:O)	CMe-C:O	Me-C:O	Other
(1)	<i>cis</i>	21.25 (3.0)	25.18 —	92.23 (6.6)	162.05 (13.2)	196.10 (3.0)	56.90 (8.8)
	<i>trans</i>	22.10 (4.4)	24.89 —	92.36 (6.6)	162.00 (13.2)	196.30 (4.2)	57.15 (6.6)
(3)		21.08 (3.7)	24.53 —	91.13 (5.9)	162.05 (13.2)	(196.80) (3.7)	0.45 —

^a Recorded on a Bruker HX-90-15'' n.m.r. spectrometer operating in the PFT-mode at 22.63 Hz; the shifts and coupling constants were taken from ¹H-broad-band-decoupled ¹³C spectra in CDCl₃ solution at 40° with CDCl₃ as internal lock; spectral window 4000 Hz. The information was gathered with a computer memory of 16 K points, yielding a precision in the reported shifts and constants of at least 0.02 p.p.m. and 0.5 Hz, respectively. The absolute sign of the coupling constants was not determined.

crude salt (2) was dissolved in the minimum amount of CH₂Cl₂, pentane (20 ml) was added, and the mixture was kept at –25° until crystals of pure (2) appeared, m.p. 71–72° (89%).§ Me₃SiCl (12.6 mmol) in CH₂Cl₂ (3 ml) was added to a solution of (2) in CH₂Cl₂ (15 ml) at 24°. After 24 h, evaporation (20°; 12 mmHg), treatment with benzene, filtration, and evaporation gave the crude trimethylsilyl derivative (3) which was covered with pentane (10 ml) and

phosphorylating agents, and hence very sensitive to moisture. The structure of compound (3) is based on the correspondence of its ¹H and ¹³C n.m.r. signals and its i.r. carbonyl stretching frequencies to those of compound (1), for which there is an independent X-ray structure determination.^{1c}

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§ All solvents were carefully dried, and all operations were carried out under strictly anhydrous conditions.

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