## 2-Methylenebicyclo [3,2,1]octa-3,6-diene, a Bishomofulvene-type Structure

## By Mitsuru Sakai $\dagger$

(Hanagatami Co., Fukui, 915-02 Japan)

Summary From a study of its n.m.r. and u.v. spectra and its dipole moment, it has been shown that the resonance contribution of a bishomofulvene-type structure is important in 2 -methylenebicyclo[3,2,1]octa-3,6-diene (2).

We report the syntheses and properties of 2-methylenebicyclo $[3,2,1]$ octa- 3,6 -diene (2) and 2 -methylenebicyclo-
[3,2,1]oct-3-ene (4). The olefin (2) was obtained in good yield either from the reaction of $(\mathbf{1})^{1}$ with triphenylphosphine methylide in THF or from the addition of methylmagnesium iodide to (1) followed by dehydration of the resulting methylcarbinol. The structure of (2) followed from its ${ }^{1} \mathrm{H}$ n.m.r. $\ddagger[\tau 3.60-4.20(4 \mathrm{H}), 4.95(1 \mathrm{H}, \mathrm{m}), 5.19(1 \mathrm{H}, \mathrm{m})$, $6.71(1 \mathrm{H}, \mathrm{m}), 7.18(1 \mathrm{H}, \mathrm{m})$, and $7.85(2 \mathrm{H}, \mathrm{m})]$ and mass

[^0]$\ddagger$ Detailed analysis of ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ n.m.r. spectra will be reported in a forthcoming full paper.
spectra. The latter spectrum confirmed the molecular weight (118) and the base peak was attributed to $(M-1)^{+}$,

(1)

(2)

(4)

(5)

(5a)
corresponding to $\mathrm{C}_{9} \mathrm{H}_{9}{ }^{+}$. Compound (4) was similarly prepared from (3) ${ }^{2} \tau 4 \cdot 20(2 \mathrm{H}, \mathrm{m}), 5 \cdot 20(2 \mathrm{H}, \mathrm{m})$, and $7 \cdot 5-$ $8.5(8 \mathrm{H}, \mathrm{m})]$.
The most striking physical properties of (2) are its ${ }^{13} \mathrm{C}$ n.m.r. and u.v. spectra and its dipole moment. ${ }^{3}$

The ${ }^{13} \mathrm{C}$ n.m.r. chemical shifts of $\mathrm{C}-2$ and $\mathrm{C}-9$ in (2) are $\delta$ $\left(\mathrm{Me}_{4} \mathrm{Si}\right) 144 \cdot 5$ and $110 \cdot 3$, respectively. Under identical conditions, the ${ }^{13} \mathrm{C}$ n.m.r. chemical shifts of $\mathrm{C}-2$ and $\mathrm{C}-9$ in (4) are $\delta\left(\mathrm{Me}_{4} \mathrm{Si}\right) 153.5$ and 102.5 , respectively, i.e., it could be taken that the C-2 in (2) is more electronegative than that in (4) and the C-9 in (2) is more positive than that in (4). ${ }^{4}$ This suggests that resonance contribution of (2a) is important in (2).
Measurement of the u.v. spectrum of (2) showed a broad absorption having fine structure at $243 \mathrm{~nm}(\epsilon 12,000)$ (calculated ${ }^{5} \lambda_{\text {max }} 232 \mathrm{~nm}$ ). The olefin (4) has a spectrum with $\lambda_{\max }$ at 230 nm , and $\epsilon_{\max } 5000$. This bathochromic effect of 13 nm in (2) can be accounted for in a similar manner $[c f .(5) \longleftrightarrow(5 a)]$.
The dipole moment was found to be 0.59 D , which suggests a high degree of charge separation in (2).§
These data for (2) are in accord with a delocalized bishomofulvene-type resonance structure (2a). ${ }^{6}$

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(6)
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§ The dipole moment was determined by measuring the dielectric constants and densities of solutions of varying molarity at $20^{\circ}$ in benzene and extrapolating.
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[^0]:    $\dagger$ Present address: Department of Chemistry, University of Alberta, Edmonton, Alberta, Canada.

