## cis-Influence in Platinum(II) Complexes Demonstrated by <sup>129</sup>I Mössbauer Spectroscopy

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Summary <sup>129</sup>I Mössbauer spectra of  $trans-L_2PtI_2$  show that the neutral ligands exert a range of *cis*-influences in opposition to their *trans*-influence.

trans-INFLUENCE effects are well documented and many techniques have been used to study this phenomenon.<sup>1</sup> Much less is known about the *cis*-influence, but it is thought to be relatively small and is reported to vary in the same or the opposite direction to the *trans*-influence.<sup>1</sup> We now report <sup>129</sup>I-Mössbauer data for the complexes *trans*-L<sub>2</sub>PtI<sub>2</sub> which demonstrate a significant *cis*-influence in opposition to the *trans*-influence.

Spectra were obtained with source (<sup>66</sup>Zn<sup>129</sup>Te,  $t_{\pm} = 70$  min) and absorbers (*ca.* 90% enriched in <sup>129</sup>I) at 4.2 K. The eight-line spectra were computer fitted using the isomer shift ( $\delta$ ), the quadrupole coupling constant ( $e^2qQ$ ), the square of the asymmetry parameter ( $\eta^2$ ), the overall intensity (using the ratios expected for a random powder), and the half-width (equal for all lines) as variables. The results are shown in the Table together with derived values for  $h_p$  and  $h_{sr}^{2,3}$ 

[e.g. the difference in  $h_p$  between  $(py)_2MX_2$  and  $(R_3P)_2MX_2$ is 0.02 for Pd–Cl and Pt–Cl but 0.07 for Pt–I]. This difference is presumably related to the more covalent character of the metal–iodine bond. The n.q.r. results were originally interpreted in terms of long-range electrostatic effects.<sup>4</sup> However, in view of the inverse cubic dependence of eq on distance, the major part of the electric field gradient must arise in the valence-shell of the halogen atom. Thus, the  $e^2qQ$ -values and  $h_p$  reflect the charge donated from iodide to platinum and the strength of the Pt–I bond. The order of bond strength is therefore L = Ndonor < S-donor < P-donor, *i.e.* the *cis*-influence operates in the opposite direction to the *trans*-influence.

Most theoretical treatments of the *trans*-influence suggest that it must operate through metal p-orbitals,<sup>1</sup> which cannot account for a *cis*-influence. M.O. calculations by Zumdahl and Drago<sup>5</sup> suggest that the *cis*- and *trans*-influences should run parallel, which is contrary to our observation. A satisfactory explanation may be found in Syrkin's theory of the *trans*-effect,<sup>6</sup> in which *d*-s hybridisation is used to show that a ligand which forms a strong bond

TABLE. <sup>129</sup>I Mössbauer data for trans-L<sub>2</sub>PtI<sub>2</sub> at 4.2 K<sup>a</sup>

L		$\delta_{znTe}$ /(mm s <sup>-1</sup> )	e <sup>2</sup> qQ <sup>129</sup> /(mm s <sup>-1</sup> )	$e^2 q Q^{127}/h^{ m b}$ /MHz	η	Γ/ (mm s <sup>-1</sup> )	$\chi^2$ /degrees of freedom	$h_p$	$h_s$
Et,P	••	-0.35(2)	$-36 \cdot 3(2)$	-1183	0.21(2)	0.89(5)	191/233	0.52	0.06
Me <sub>s</sub> S	••	-0.24(1)	-34.7(1)	-1131	0.13(1)	1.13(1)	625/233°	0.49	0.05
Et,S	••	-0.25(1)	$-32 \cdot 3(1)$	-1052	0.12(1)	1.00(2)	283/242	0.46	0.04
β-Picoline		-0.30(1)	$-32 \cdot 3(1)$	-1052	0.07(4)	$1 \cdot 24(3)$	224/242	0.46	0.05
Pyridine		-0.23(1)	-31.5(1)	-1026	d`́	$1 \cdot 11(2)$	888/243°	0.45	0.04
ΝΉ,		-0.32(1)	-30.5(1)	-994	d	1.67(3)	$754/226^{\circ}$	0.43	0.05

<sup>a</sup> Numbers in parentheses are uncertainties in the last figure. <sup>b</sup> Calculated from<sup>2</sup>  $32.58 e^2 q Q^{129}/(\text{mm s}^{-1}) = (e^2 q Q^{127}/h)/\text{MHz}$ . <sup>c</sup> Large  $\chi^2$  due to intensity-constraint. <sup>d</sup> Computer fitting gave  $\eta^2 < 0$ . Data shown are for  $\eta$  constrained to zero.

The isomer shifts and  $h_s$ -values show no systematic trends and are probably equal within the experimental error. The quadrupole coupling constant values show well defined, systematic variations, clearly indicating the operation of a *cis*-influence. Similar trends have been seen in n.q.r. data for related chloro-complexes,<sup>4</sup> but these show much less sensitivity to change in the neutral ligand to platinum would strengthen the bonds in the *cis*-positions. This is clearly compatible with our data.

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