## Stereospecific Reduction and Methylation of Two 1-Iodo-trans-cycloalkenes

By Mark S. Baird

(Department of Organic Chemistry, The University, Newcastle upon Tyne NE1 7RU)

Summary Two 1-iodo-trans-cycloalkenes, obtained by ring expansion of suitable gem-di-iodocyclopropanes with silver perchlorate in methanol, can be converted into the trans-cyclo-alkene, 1-deuterio-trans-cyclo-alkene, and the 1-methyl-trans-cycloalkene.

There have been a number of recent reports concerning the ring expansion of exo-8-bromobicyclo[5,1,0]octanes to

3-methoxy-trans-cyclo-octenes and of 8,8-dibromobicyclo-[5,1,0] octanes to 2-bromo-3-methoxy-trans-cyclo-octenes using AgClO<sub>4</sub> in MeOH; thus the parent systems (I; X = H, Y = Br) and (I; X = Y = Br) give (IIa) and (IIb) respectively. These reactions may also be applied to the preparation of trans-cyclononenes; in this way (III; X = Y = Br) gives (IVb).

Reaction of cycloheptene with KOBut and iodoform3

J.C.S. CHEM. COMM., 1974

OMe
$$\alpha_{i} X = H \quad d_{i} X = D$$

$$b_{i} X = Br \quad e_{i} X = Me$$

$$C_{i} X = I$$
(II)
$$(IV)$$

gave 8,8-di-iodobicyclo [5,1,0] octane  $(I; X = Y = I) \dagger (58\%)$ as a brown oil, stable for several weeks at 25°. Treatment of (I; X = Y = I) with AgClO<sub>4</sub> in MeOH at 25° gave the trans-cyclo-octene derivative (IIc) (90%); the n.m.r. spectrum of (IIc) was similar to that of (IIb), and contained a double doublet at  $\tau$  3.95 (I = 10 and 5.5 Hz). Treat-

ment of (IIc) with MeLit in ether for 15 s at 25-35° followed by quenching with H<sub>2</sub>O gave (IIa) (60%), identical by n.m.r. and i.r. spectroscopy with an authentic sample.§ If the reaction was repeated, but quenched with D<sub>2</sub>O, the product was (IId), but if the products were stirred with an excess of MeI before adding water, (IIe) was the major product.

Reaction of cis, cis-cyclo-octa-1,5-diene with iodoform and KOBu<sup>t3</sup> gave (III; X = Y = I) (63%); ¶ treatment of (III; X = Y = I) with  $AgClO_4$  in MeOH gave (IVc) (89%). Reaction of (IVc) with MeLi in ether at 25° followed by H<sub>2</sub>O quenching gave (V) as the major product, together with (IVa) and (IVe). However, reaction of (IVc) with LiCuMe<sub>2</sub> (from Cu<sub>2</sub>I<sub>2</sub> and MeLi<sup>6</sup>) for 35 min followed by quenching with H2O gave (IVa) (60%) relatively cleanly; the geometry of the 2,3-bond was confirmed as trans on examination of the n.m.r. spectrum, which showed a 16 Hz coupling between the two protons.\*\* By working up the above reaction with D<sub>2</sub>O instead of H<sub>2</sub>O the deuteriated olefin (IVd) could be readily obtained (66%). It was also possible to obtain the methylated compound (IVe) (80%) as the major product by reaction of (IVc) with MeLi at 0° followed by treatment with excess of MeI.

The reactions described above provide a useful extension to the known syntheses of trans-cyclo-olefins, particularly when the exo-bromobicyclo [n,1,0] alkane necessary for stereospecific ring opening to the trans-cyclo-olefin is not readily available, as is the case for (III; X = H, Y = Br).

(Received, 19th September 1973; Com. 1310.)

- † Satisfactory microanalyses or high resolution molecular weights have been obtained for all new compounds.
- ‡ For lithiation of other 1-halogenocycloalkenes see ref. 4. MeLi obtained from Alfa Inorganics.
- § We thank Professor C. B. Reese for providing the spectra of authentic (IIb).
- ¶ This was stable for long periods at  $0^\circ$ , but decomposed when kept at  $25^\circ$  in daylight for several weeks; a crystalline di-iodide, m.p.  $58-59^\circ$ , was isolated from the products, and characterised as (VI; X = I) by comparison of its spectra with those of (VI;  $X = Br),^{2,5}$
- \*\* The four olefinic protons appeared as a complex signal in the n.m.r. spectrum of (IVa), but were differentially shifted in the presence of [Pr(fod)<sub>2</sub>].
- 1 (a) C. B. Reese and A. Shaw, J. Amer. Chem. Soc., 1970, 92, 2566; (b) M. S. Baird and C. B. Reese, Chem. Comm., 1970, 1644; (c) Tetrahedron Letters, 1971, 4637.

  <sup>2</sup>C. B. Reese and A. Shaw, Chem. Comm., 1970, 1365, 1367.

  - <sup>8</sup> J. P. Oliver and U. V. Rao, J. Org. Chem., 1966, 31, 2696.
  - <sup>4</sup> J. C. Giacomoni, A. Cambon, and E. Rouvier, Bull. Soc. chim. France, 1970, 3097.

  - M. S. Baird and C. B. Reese, J. Chem. Soc. (C), 1969, 1808.
    E. J. Corey and G. H. Posner, J. Amer. Chem. Soc., 1967, 89, 3911.
  - <sup>7</sup>C. J. Osborn, T. C. Shields, B. A. Shoulders, C. G. Gardenas, and P. D. Gardner, Chem. and Ind., 1965, 766.