

## Molecular Structure of *cis*-Dichloro[(*R*)- $\alpha$ -methylbenzylamine][(*S*)-1,2,2-trimethylpropyl (*R*)-vinyl ether]platinum(II)

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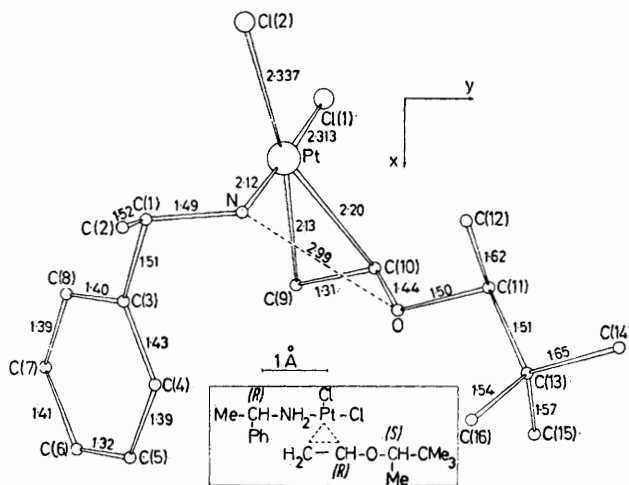
**Summary** The X-ray structural analysis of the title compound showed that the absolute configuration of the chiral centre formed by complexation of the vinyl group to Pt<sup>II</sup> is *R* and gave indications on the conformation of the amine and olefin ligands.

The crystals form plates normal to [100];  $a = 25.56 \pm 0.03$ ,  $b = 11.37 \pm 0.01$ ,  $c = 6.84 \pm 0.01$  Å;  $V = 1988$  Å<sup>3</sup>;  $D_m = 1.65$ ,  $D_c = 1.67$  g cm<sup>-3</sup>,  $Z = 4$ ;  $F(000) = 968$ .

ALTHOUGH transition metal-olefin complexes have been extensively investigated, complexes of olefinic compounds containing electron releasing substituents have only recently received attention.<sup>1</sup>

As part of our investigation<sup>2</sup> into the stereospecificity of complexation of chiral vinyl derivatives to Pt<sup>II</sup>, we report here the X-ray structure of *cis*-dichloro[(*R*)- $\alpha$ -methylbenzylamine][(*S*)-1,2,2-trimethylpropyl (*R*)-vinyl ether]platinum(II) (I). Complex (I) was prepared by displacement of ethylene with (*S*)-1,2,2-trimethylpropyl vinyl ether† from the corresponding ethylene-Pt<sup>II</sup> complex.<sup>3</sup> Crystallization from a mixture of acetone, diethyl ether, and pentane (1:1:2) gave a diastereoisomerically homogeneous sample of (I) (m.p. 134–135°) as supported by an n.m.r. spectrum that showed a single resonance for the *t*-butyl group of the complexed vinyl ether.

† The preparation and the optical purity determination will be reported elsewhere.



Space group:  $P2_12_12_1$  by systematic absences. Cu- $K_\alpha$  radiation (1.5418 Å),  $\mu = 158.7 \text{ cm}^{-1}$ . The intensities of 1764 symmetrically independent reflections were measured by means of a Nonius microdensitometer on integrated equi-inclination multiple-film Weissenberg photographs taken about  $c$  axis; the intensities were corrected for Lorentz, polarization, and absorption factors.

The structure was solved by means of the heavy-atom method; the full-matrix least-squares refinement of atomic co-ordinates, platinum and chlorine anisotropic temperature factors, and C, N, and O isotropic temperature factors converged to a final  $R$  value of 0.071. Further refinement is in progress. Two enantiomorphous solutions of the structure were possible, but knowledge of the absolute configuration of the amine and the vinyl ether used ( $R$  and  $S$  respectively) allows us to assign<sup>4</sup> an absolute configuration  $R$  to the chiral centre of the co-ordinated vinyl group. The shape of the molecule and bond lengths for the examined  $SSR$  diastereoisomer are shown in the Figure (e.s.d.'s are 0.008 for Pt-Cl, 0.03 for Pt-N and Pt-C, 0.04 for N-C and O-C, and 0.05 Å for C-C bond lengths). The complex shows almost undistorted square-planar arrangement of the

ligands around the Pt atom with the two chlorines in a *cis* position. The best plane through Pt, Cl(1), Cl(2), and N does not bisect the co-ordinated double bond, C(9) being 0.32 Å below while C(10) is 0.99 Å above it. Moreover, the double bond is slightly displaced from the perpendicular ( $5.4^\circ$ ) to the plane, C(9) being closer to Pt. Unlike similar olefin complexes,<sup>5</sup> the plane Pt, C(9), C(10) is perpendicular to the co-ordination plane within the error limits.

The dihedral angles C(9), C(10), O, C(11) and C(10), O, C(11), C(13) are  $0^\circ$  and  $42^\circ$  respectively. The Pt, N, C(1) plane deviates from perpendicularity to the Pt, Cl(1), Cl(2) plane (dihedral angle  $77^\circ$ ). Furthermore while C(2) is in a "quasi anti" position with respect to Pt, the benzene ring is bent toward Pt [C(3)  $\cdots$  Pt distance 3.51 Å] and lies approximately in a plane containing the C(1)-H bond. Finally, the short distance of 2.99 Å between the N and O atoms strongly suggests the presence of an intramolecular hydrogen bond which could play an important role in determining the conformation of co-ordinated vinyl ether.

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