

The Ring Contraction Step in Gibberellin Biosynthesis

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Summary The conversion of *ent*-7 α -hydroxykaurenoic acid (**1**) into gibberellin A₁₂-aldehyde (**2**) and *ent*-6 α ,7 α -dihydroxykaurenoic acid (**3**) involves the loss of one hydrogen atom at C-6.

WE have previously demonstrated the conversion of *ent*-7 α -hydroxykaurenoic acid (**1**) by ring contraction into gibberellin A₁₂(GA₁₂)-aldehyde (**2**) and by hydroxylation to *ent*-6 α ,7 α -dihydroxykaurenoic acid (**3**) in a cell-free system from the endosperm of *Cucurbita maxima* seed.^{1,2} The formation of GA₁₂-aldehyde (**2**) is part of the pathway by

which the cell-free system converts mevalonate into several gibberellins including gibberellin A₄.^{2,3} In the presence of Mn²⁺, however, GA₁₂-aldehyde (**2**) is converted only into GA₁₂ (**4**), further steps being inhibited. The *ent*-6 α -hydroxylation is an alternative reaction which takes place simultaneously with the ring contraction both in the presence and absence of Mn²⁺. The incubation of *ent*-7 α -hydroxykaurenoic acid (**1**) in the presence of Mn²⁺ leads to the formation of only four major products: GA₁₂-aldehyde (**2**), GA₁₂ (**4**), *ent*-6 α ,7 α -dihydroxykaurenoic acid (**3**), and an oxidation product of the latter, an unidentified *ent*-kaurenoic acid derivative. We now present results on the

mechanism of the ring contraction and *ent*-6 α -hydroxylation using this system and ^3H : ^{14}C double labelled substrate.

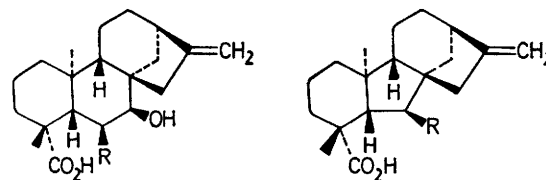
TABLE. Radio-activity (d.p.m.) of *ent*-7 α -hydroxykaurenoic acid and products

| Compound | ^3H | ^{14}C | ^3H : ^{14}C |
|---|--------------|-----------------|--------------------------------|
| <i>ent</i> -7 α -hydroxykaurenoic acid (1) | | | |
| Initial (1/10th aliquot) | 6792 | 3335 | 2.04 |
| Recovered | 28555 | 12654 | 2.26 |
| GA ₁₂ -aldehyde (2) | 6993 | 7126 | 0.98 |
| GA ₁₂ (4) | 4279 | 4748 | 0.90 |
| <i>ent</i> -6 α ,7 α -Dihydroxykaurenoic acid (3) | 6496 | 5841 | 1.11 |
| " " " metabolite " .. | 4506 | 3502 | 1.29 |

ent-6,6- $^3\text{H}_2$ -7 α -Hydroxykaurenoic acid was prepared from *ent*-7-oxokaurenoic acid by exchange with MeONa–MeO ^3H followed by reduction with AlPr 1_3 –Pr ^1OH and was mixed with ^{14}C -labelled *ent*-7 α -hydroxykaurenoic acid prepared from 2- ^{14}C mevalonate in the cell-free system. The specific activity of the ^3H -label in the mixture was 1.9 $\mu\text{Ci}/\mu\text{mol}$; the amount of ^{14}C was varied to produce ^3H : ^{14}C ratios from 1.39 to 2.04. The doubly labelled substrate was incubated with endosperm preparation in the presence of Mn $^{2+}$ and the usual co-factors.¹ The products were isolated by t.l.c. and their radioactivity measured by liquid scintillation counting using internal standardisation. In several instances the identity of the products was confirmed by g.l.c. Their identification by combined g.l.c.–m.s. has been described previously,^{1,2}

The results presented in the Table show that the ^3H : ^{14}C ratios in the products are only half of the ratio in the substrate. Thus the formation of GA₁₂-aldehyde (2) and *ent*-6 α ,7 α -dihydroxykaurenoic acid (3) involves the loss of one hydrogen atom from the 6-position in *ent*-7 α -hydroxykaurenoic acid (1). As would be expected from this mechanism an isotope effect was observed. Since the ^3H and ^{14}C labels are present in different molecules in the

mixed substrate, the isotope effect leads to a slower conversion of ^3H -molecules than ^{14}C -molecules and the ^3H : ^{14}C ratio increases in unchanged substrate as the reaction proceeds. Up to a 7-fold increase in this ratio was observed in experiments with the reaction nearly complete with respect to the ^{14}C -substrate.



(1) R = H
(3) R = OH

(2) R = CHO
(4) R = CO₂H

Based upon a ^3H : ^{14}C ratio of 2:1 in the substrate, the average ^3H : ^{14}C ratios for GA₁₂-aldehyde (2) and *ent*-6 α ,7 α -dihydroxykaurenoic acid (3) in seven experiments were 0.98 and 1.12 respectively. A lower ratio 0.82 was obtained for GA₁₂ (4) and is probably due to partial exchange during the oxidation step. Our results conflict with those of Hanson *et al.*,⁴ who observed no loss of ^3H in GA₁₂-aldehyde (2) isolated from the culture of the fungus *Gibberella fujikuroi* after incubation with [1,1- $^3\text{H}_2$,1- ^{14}C]geranyl pyrophosphate. They suggested on the basis of a low (0.005%) incorporation that the *ent*-6 α -hydrogen atom migrates to C-7 during ring contraction. Our results are, however, in accord with the proposition by the same group that the ring contraction is initiated by abstraction of the *ent*-6 α -hydrogen itself.⁵

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