

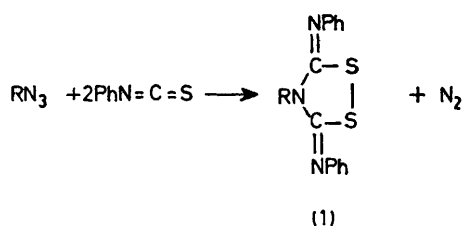
## Reaction of Primary Alkyl Azides with Phenyl Isothiocyanate

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**Summary** The reaction of primary alkyl azides with phenyl isothiocyanate gives 4-alkyl-3,5-bis(phenylimino)-1,2,4-dithiazolidines as the main product.

HYDRAZOIC ACID and sodium azide react with phenyl isothiocyanate to form 5-phenylamino-1,2,3,4-thiatriazole and 1-phenyl-1,2,3,4-tetrazoline-5-thione respectively.<sup>1</sup> Tri-n-butyltin azide and triphenyltin azide give the corresponding 1:1 adducts.<sup>2</sup> Alkyl azides react with equimolar amounts of sulphonyl isothiocyanates to yield the 1,3-dipolar cycloaddition products, 4-alkyl-5-sulphonylimino-1,2,3,4-thiatriazolines.<sup>3</sup>



**a**; R = Me[CH<sub>2</sub>]<sub>3</sub>

**b**; R = Me[CH<sub>2</sub>]<sub>4</sub>

**c**; R = Me<sub>2</sub>CH[CH<sub>2</sub>]<sub>3</sub>

**d**; R = EtCH(Me)CH<sub>2</sub>

**e**; R = Me[CH<sub>2</sub>]<sub>5</sub>

**f**; R = PhCH<sub>2</sub>

**g**; R = C<sub>10</sub>H<sub>7</sub>CH<sub>2</sub>

Primary alkyl azides are now shown to react with phenyl isothiocyanate in 1:2 molar ratios giving 4-alkyl-3,5-bis(phenylimino)-1,2,4-dithiazolidines (**1**) in 29–65% yield. Traces of 4-phenyl-3,5-bis(phenylimino)-1,2,4-dithiazolidine

(**2**) are formed in the preparation of (**1a–e**). The yellow, crystalline dithiazolidines are isolated after heating the pure reactants at 80–100 °C in the absence of solvent. The low solubility of (**2**) in light petroleum (b.p. 40–60 °C) enables efficient separation from (**1a–e**).

All products are characterised by i.r. [ $\nu_{\text{max}}$  (KBr) 1610 and 1580 (C=N), and 1300 (C–N) cm<sup>-1</sup>]<sup>4</sup> and mass spectral data and by a single crystal X-ray analysis of (**1f**; R = PhCH<sub>2</sub>). A *m/e* 285 ion, common to all mass spectra, represents replacement of the alkyl (**1a–g**) or phenyl groups (**2**) attached to the ring nitrogen atom of the molecular ion by a hydrogen atom. Two important fragmentations from this ion, *m/e* 167 and *m/e* 118, can be explained by cleavage of the S(2)–C(3) and N(4)–C(5) bonds of the dithiazolidine ring.

The reaction mechanism may involve stepwise addition of the 2 molecules of PhNCS *via* a 3-membered ring intermediate, RN–S–C=NPh. A similar route has been proposed for the formation of 1-phenyl-5-phenylimino-1,2,4-dithiazolidine-3-thione by the reaction of PhN<sub>3</sub> with CS<sub>2</sub> in the presence of AlCl<sub>3</sub>. The formation of (**2**) is not yet understood. Under identical conditions transfer of a phenyl group from PhNCS to (**1a–e**) does not occur and PhN<sub>3</sub> which could be formed as a by-product in the main reaction, does not react with PhNCS.

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