

Rearrangements of 8-Oxa-4-thia-1-azabicyclo[4,3,0]non-2-en-7-ones: a New Synthesis of 4-Thiazolines

By ANTHONY G. W. BAXTER and RICHARD J. STOODLEY*

(Department of Organic Chemistry, The University, Newcastle upon Tyne NE1 7RU)

Summary The enethiolates, *e.g.* (13), formed from thiazino-oxazolidinones, *e.g.* (9), in the presence of potassium *t*-butoxide, are converted into 4-thiazolines *e.g.* (15), probably by an S_N2 -like pathway.

As part of an exploration of the chemistry of β -amino- α -mercaptoacrylates, we have examined some β -elimination reactions of methyl 7-oxo-8-oxa-4-thia-1-azabicyclo[4,3,0]non-2-ene-3-carboxylates.

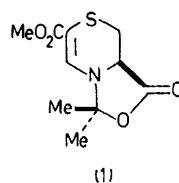
Treatment of the thiazino-oxazolidinone (1)¹ with KOBU^{\dagger} (1 mol. equiv.) in $[\text{D}_6]\text{H}_2\text{O}$ dimethyl sulphoxide caused the rapid formation (n.m.r. spectroscopy) of the salt (2). Acidification of the mixture yielded the acid (3) as an unstable oil and addition of MeI to the mixture afforded the methyl ester (4)[†] (26% after silica-gel chromatography). Under similar conditions the dimethylthiazino-oxazolidinone (8)² gave the salt (5) which was converted into the unstable acid (6) and the methyl ester (7)[†] (62% after silica-gel chromatography). The acids (3) and (6) were decomposed to methyl 2,2-dimethyl-4-thiazoline-5-carboxylate (10)[†] during silica-gel chromatography.

When KOBU^{\dagger} (1 mol. equiv.) was added to a solution of the derivative (1) in dimethyl sulphoxide containing MeI, the methylthioacrylate (11)[†] was the major product (42% after silica-gel chromatography).

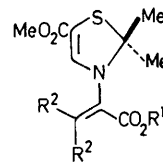
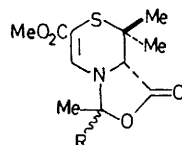
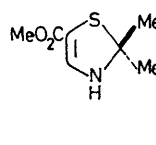
Evidently the foregoing reorganizations involve the intermediacy of the enethiolates, *e.g.* (12), formed by a β -elimination process. In principle, cleavage of the C–O bond of these intermediates may precede (S_N1 -like) or be coupled with (S_N2 -like) the attack by the thiolate anion. In an attempt to distinguish between these pathways, a thiazino-oxazolidinone possessing a chiral centre at position 9 was required. Treatment of the thiazine-carboxylic acid (14)³ with redistilled MeCHO in CDCl_3 containing MgSO_4 afforded (90%) the derivative (9)[†], m.p. 50–52°, $[\alpha]_D + 276^\circ$ (CHCl_3), as a single diastereoisomer. When treated with KOBU^{\dagger} followed by acid, the thiazino-oxazolidinone (9) was transformed into the 4-thiazoline (16), which gave (33%) the salt (17),[†] m.p. 130–132°, $[\alpha]_D + 9^\circ$ (CHCl_3), with *t*-butylamine. O.r.d. measurements $\{[\Phi] + 13,800^\circ$ (252), -160° (290), $+4800^\circ$ (333), and -4600° (382 nm) (EtOH)} confirmed that the salt (17) was optically active.

Although it is not possible to define the absolute stereochemistry or the optical purity of the 4-thiazoline (17), it is clear that chirality is maintained during the reorganization. This result implicates a S_N2 -like process in the conversion of the enethiolate (13) into the salt (15).

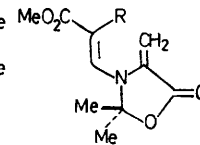
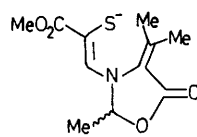
These results exemplify an unusual molecular rearrangement which provides a novel synthesis of 4-thiazolines, a relatively unexplored class of compound.⁴ There is some



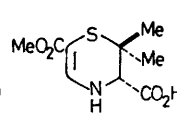
(1)

(2) $\text{R}^1 = \text{K}, \text{R}^2 = \text{H}$ (5) $\text{R}^1 = \text{K}, \text{R}^2 = \text{Me}$
(3) $\text{R}^1 = \text{R}^2 = \text{H}$ (6) $\text{R}^1 = \text{H}, \text{R}^2 = \text{Me}$
(4) $\text{R} = \text{Me}, \text{R} = \text{H}$ (7) $\text{R}^1 = \text{R}^2 = \text{Me}$ (8) $\text{R} = \text{Me}$
(9) $\text{R} = \text{H}$ 

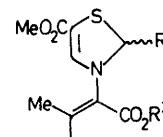
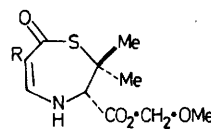
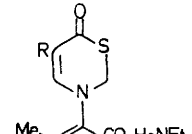
(10)

(11) $\text{R} = \text{MeS}$
(12) $\text{R} = \text{S}^-$ 

(13)



(14)

(15) $\text{R}^1 = \text{Me}, \text{R}^2 = \text{K}$
(16) $\text{R}^1 = \text{Me}, \text{R}^2 = \text{H}$
(17) $\text{R}^1 = \text{Me}, \text{R}^2 = \text{Bu}^{\dagger}\text{NH}_3^+$ (18) $\text{R} = \text{phthalimido}$ (19) $\text{R} = \text{phthalimido}$

analogy for this behaviour in the triethylamine-induced conversion of 1,4-thiazepines, *e.g.* (18), into 1,3-thiazines, *e.g.* (19).⁵

We thank the S.R.C. for a research studentship (to A.G.W.B.).

(Received, 10th January 1975; Com. 031.)

[†] The composition of new compounds was confirmed by elemental analysis and/or by mass spectroscopy. Structural assignments were supported by i.r., u.v., and n.m.r. spectroscopic evidence.

¹ A. R. Dunn, I. McMillan, and R. J. Stoodley, *Tetrahedron*, 1968, **24**, 2895.

² J. Kitchin and R. J. Stoodley, *J.C.S. Perkin I*, 1973, 22.

³ I. McMillan and R. J. Stoodley, *Tetrahedron Letters*, 1966, 1205; *J. Chem. Soc. (C)*, 1968, 2533.

⁴ F. Kurzer in 'Organic Compounds of Sulphur, Selenium, and Tellurium,' ed. D. H. Reid (Specialist Periodical Reports), The Chemical Society, London, 1970, Vol. 1, p. 399; 1971, Vol. 2, p. 630.

⁵ J. R. Jackson and R. J. Stoodley, *Chem. Comm.*, 1970, 14; *J.C.S. Perkin I*, 1972, 1063.