

Biosynthesis of Oleanene- and Ursene-type Triterpenes from [4-¹³C]Mevalonic Acid in Tissue Cultures of *Isodon japonicus* Hara

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Summary On the basis of the ¹³C-labelling patterns elucidated by ¹³C n.m.r. spectroscopy in oleanene- and ursene-type triterpenes (1)–(5) isolated from *Isodon japonicus* tissue cultures fed with [4-¹³C]mevalonic acid, Ruzicka's hypothesis for cyclisation of squalene 2,3-oxide to the pentacyclic triterpenes has been verified experimentally.

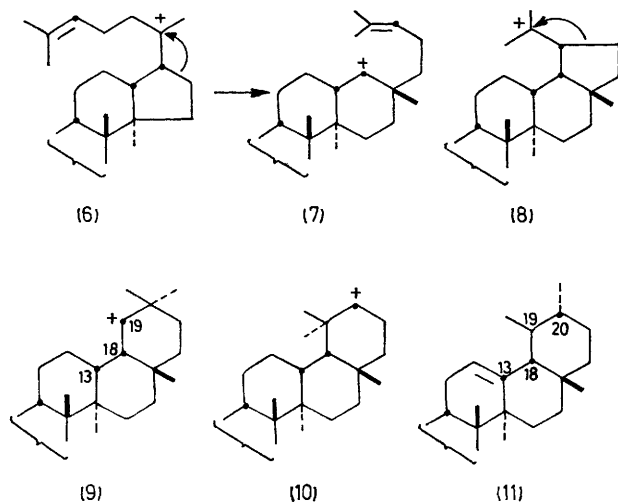
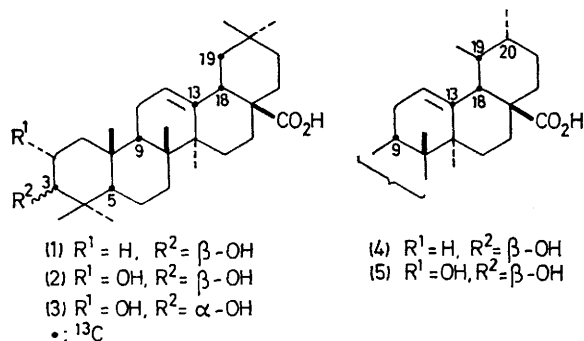
RUZICKA's group has proposed that pentacyclic triterpenes such as α - and β -amyrin arise by cyclisation of squalene† folded in a chair-chair-boat form with specific 1,2-hydride

shifts.¹ Although an alternative mechanism involving a 1,3-hydride shift was considered later, the postulated 1,2-hydride shift has good experimental support.^{2,3} However, the rearrangement of carbon atoms during formation of the olean-12-ene and the urs-12-ene skeleton from squalene 2,3-oxide has not yet been demonstrated owing to the difficulty of chemical degradation of the triterpenes biosynthetically labelled with ¹⁴C.

Recently, we have successfully assigned all ¹³C n.m.r. signals for a number of olean-12-enes^{4,5} and urs-12-enes.⁵ Having elucidated ¹³C-labelling patterns in some of these

†Recently, (S)-squalene 2,3-oxide was found to be the exclusive precursor of β -amyrin in plant systems; see D. H. R. Barton, T. R. Jarman, K. G. Watson, and D. A. Widdowson, *J.C.S. Chem. Comm.*, 1974, 861.

pentacyclic triterpenes synthesised from [4-¹³C]mevalonic acid in tissue cultures of *Isodon japonicus* Hara by ¹³C n.m.r. spectroscopy, we here show the occurrence of the rearrangement in Ruzicka's hypothesis for the cyclisation of squalene to olean-12-ene- and urs-12-ene-type triterpenes.



Tissue cultures of *Isodon japonicus* were grown in Linsmaier-Skoog liquid media; [4-¹³C]mevalonic acid was prepared from [2-¹³C]acetic acid by Cornforth's method.⁸ A solution of [4-¹³C]mevalonic acid (1 g; ca. 30% ¹³C) in 50% ethanol (7.2 ml) was distributed among 18 bottles containing tissue cultures of *Isodon japonicus* (18 × 300 ml); cells were harvested after two weeks and extracted with methanol. Oleanolic acid (1), maslinic acid (2), 3-epimaslinic acid (3), ursolic acid (4), and 2α-hydroxyursolic acid (5) were obtained by repeated preparative t.l.c. as described previously.^{5,6} ¹H-Noise-decoupled ¹³C Fourier transform n.m.r. spectra of the methyl esters of these triterpenes were then

compared for enriched and unenriched samples in CDCl₃ (see Table).

TABLE

Carbon-13 chemical shifts, δ(C), of enriched carbon atoms in CDCl₃^a

Compound	δ(C) values ^b					
	C-3	C-5	C-9	C-13	C-18	C-19
(1)	78.7	55.2	47.6	143.4	41.3	45.8
(2)	83.8	55.3	47.5	143.6	41.3	45.8
(3)	78.9	48.1	47.4	143.8	41.3	46.0
(4)	78.8	55.4	47.5	138.0	52.8	39.1
(5)	83.8	55.4	47.5	138.1	52.8	39.1
						(or C-20) ^c
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^a ¹H-Noise-decoupled ¹³C n.m.r. spectra were taken with a Varian NV-14 FT n.m.r. spectrometer operating at 15.09 MHz; precision ca. ±0.1. ^b Assignments and δ(C) values for the other carbon signals are reported in refs. 4 and 5. ^c We suggest that the enriched carbon is C-19 from Yb(fod)₃-induced ¹³C shifts for the enriched and unenriched samples, the induced shift being apparently larger for C-19 than for C-20. However, unambiguous assignment must await further evidence.

The ¹³C n.m.r. spectrum of methyl 3-epimaslinic acid clearly shows that the six carbons C-3, -5, -9, -13, -18, and -19, based on [4-¹³C]mevalonic acid, were enriched by ca. 2 times. Similar ¹³C-labelling patterns were observed in the spectra of the methyl esters of (1) and (2) with enrichments of ca. 5 and 2 times, respectively. The results are entirely in accord with Ruzicka's hypothesis for cyclisation from squalene to β-amyrin [(6) → (7) → (8) → (9) → β-amyrin].

In the biosynthesis of the ursene-type triterpenes (4) and (5), the D-ring is formed *via* process (6) → (7) as postulated for the biosynthesis of β-amyrin, since ¹³C enrichment was clearly observed at C-18 together with C-3, C-5, C-9, and C-13 in (4) and (5) by ca. 5 and 2 times, respectively. Two possible mechanisms for E-ring formation have been considered so far.^{1b,7,9} If the cation (8) or its equivalent is an intermediate, one of the enriched carbons originating from [4-¹³C]mevalonic acid should be situated at C-19. On the other hand, if the E-ring is formed *via* process (7) → (10) → (11), the enriched carbon should be C-20. As indicated in the Table, although one signal corresponding to C-19 or C-20 of (4) and (5) was enriched by ca. 5 and 2 times, respectively, assignment of the enriched signal remains rather ambiguous because the C-19 and C-20 signals in these compounds are quite close to each other.⁵ It is thus not possible at present to distinguish between these alternative mechanisms; further studies for differentiating between them are in progress.

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