

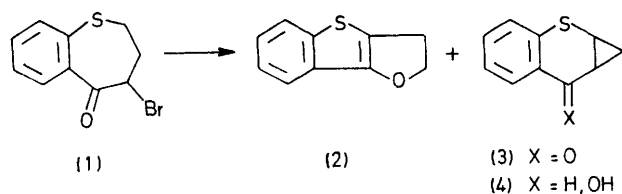
A Novel Rearrangement of a Benzo[*b*]thiepin Derivative¹

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Summary The unexpected transformation of 4-bromo-2,3-dihydrobenzo[*b*]thiepin-5(4*H*)-one (**1**) into 2,3-dihydro-[1]benzothieno[3,2-*b*]furan (**2**) and 1,1a,7,7a-tetrahydrobenzo[*b*]cyclopropa[*e*]thiopyran-7-one (**3**) is reported.

THE recent report² on the ring contraction of 4-bromo-2,3,4,5-tetrahydrobenzo[*b*]thiepin-5-ol to 2-(2-bromoethyl)-benzo[*b*]thiophen prompts us to report our preliminary work



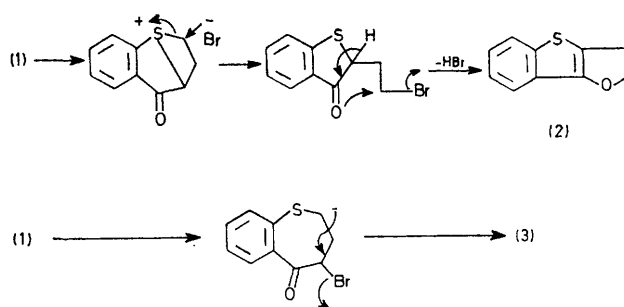
on a novel rearrangement of the thiepinone (1). It was recently reported that identifiable products could not be isolated from the reaction of (1) with a wide variety of bases.³ In our hands, reactions of (1) with LiBr-Li₂CO₃ in refluxing HCONMe₂ gave two crystalline compounds which we have identified as (2) (m.p. 58–60 °C; 15% yield) and (3) (m.p. 70–71 °C; 1%) on the basis of their elemental and spectral analyses.† Compound (2) was recovered unchanged from attempted reduction with LiAlH₄, whereas compound (3) gave the alcohol (4), m.p. 93–94 °C, with LiAlH₄.

† All new compounds gave analytical figures for C, H, and S to within $\pm 0.20\%$ of theory; spectral data, compound (2): δ (CDCl₃) 3.27 and 5.00 (each 2H, t, *J* 8.5 Hz, 2- and 3-H, respectively), and 7.13–7.83 (4H, ArH); ν_{\max} (KBr) 1590, 1450, 755, and 730 cm⁻¹; λ_{\max} (EtOH) 210, 251, 304, and 313 nm; *m/e* 176 (*M*⁺, 100%), 147 (74%), and 115 (30%); the peak at *m/e* 144 (*M*⁺ - S) was of <5% intensity; compound (3): δ (CDCl₃) 1.25–1.62 (2H, m, 1-H), 1.68–1.98 (2H, m, 1a- and 7a-H), and 7.02–7.62 and 7.68–7.92 (each 2H, m, ArH); ν_{\max} (KBr) 1695, 1000, 1455, 1310, 910, and 655 cm⁻¹; *m/e* 176 (*M*⁺, 100%), 147 (82%), 136 (17%), 115 (30%), and 108 (21%).

¹ For previous paper in the series: Heterocycles, see P. M. Weintraub, *J. Medicin. Chem.*, 1972, **15**, 419.

² A. Chatterjee and B. K. Sen, *J.C.S. Chem. Comm.*, 1974, 626.

³ V. S. Traynelis, J. C. Sih, and D. M. Borgnaes, *J. Org. Chem.*, 1973, **38**, 2629.



SCHEME

We rationalize the formation of (2) as shown in the Scheme. A similar mechanism was proposed in ref. 2. We postulate that (3) is formed by anion formation α to the sulphur with subsequent intramolecular displacement of Br⁻.

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