

## Specific Dehydrocyclisation of n-Hexane over Supported Platinum Crystallites Prepared from Platinum Carbonyl Cluster Anion Salts

By MASARU ICHIKAWA\*

(Sagami Chemical Research Center, 4-4-1 Nishi-Onuma, Sagamihara, Kanagawa, Japan)

**Summary** Highly dispersed Pt-aggregates have been prepared by pyrolysis of the  $\text{Pt}_{15}$ — $\text{Pt}_3$  carbonyl anion salts on  $\gamma\text{-Al}_2\text{O}_3$  and  $\text{SiO}_2$  which catalyse the specific dehydrocyclisation of n-hexane.

WE have tried to prepare highly dispersed Pt-crystallites by pyrolysis of well-defined Pt-cluster compounds, the particle size of which may be conveniently controlled down to  $<10 \text{ \AA}$ . The trigonal prismatic salts,  $[\{\text{Pt}_3(\text{CO})_6\}_n]^{2-} \cdot (\text{NET}_4^+)_2$  ( $n = 1-5$ ), were prepared by Chini's method,<sup>1</sup>

and deposited in high dispersion (0.05—1 Pt wt %) from tetrahydrofuran solutions over silica gel,  $\gamma$ -Al<sub>2</sub>O<sub>3</sub>, or Vycor glass (no. 7930) with surface area 240—380 m<sup>2</sup> g<sup>-1</sup>. The dispersed salts decomposed completely when pyrolysed *in vacuo* or under He or H<sub>2</sub> to produce stoichiometric amounts of CO and NEt<sub>3</sub>: [Pt<sub>15</sub>(CO)<sub>30</sub>][NEt<sub>4</sub>]<sub>2</sub> at 150, [Pt<sub>9</sub>(CO)<sub>18</sub>][NEt<sub>4</sub>]<sub>2</sub> at 165, and [Pt<sub>6</sub>(CO)<sub>12</sub>][NEt<sub>4</sub>]<sub>2</sub> at 170 °C. Also, the terminal (2025—1980 cm<sup>-1</sup>) and bridging (1870—1810 cm<sup>-1</sup>) carbonyl absorptions in the i.r. spectra of the supported Pt-carbonyl cluster salts disappeared when the Al<sub>2</sub>O<sub>3</sub> disc was evacuated at 180—250 °C for 0.5 h.

of n-hexane were measured between 250 and 340 °C in the presence of an excess of hydrogen (n-hexane:H<sub>2</sub> = 1:10) over the supported Pt-crystallites. For conversions of <2%, the composition of the products was determined by g.l.c. using a silicon SE-25 column at 0 °C. The dehydrogenation products (MCP + CH) were preferentially (with >75% selectivity) obtained over the Pt-aggregates prepared from Pt<sub>15</sub>, Pt<sub>12</sub>, Pt<sub>9</sub>, and Pt<sub>6</sub> salts on  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> and Vycor glass. The smaller Pt-aggregates deposited from [Pt<sub>6</sub>(CO)<sub>12</sub>][NEt<sub>4</sub>]<sub>2</sub> and [Pt<sub>3</sub>(CO)<sub>3</sub>(PPh<sub>3</sub>)<sub>4</sub>] on  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> resulted in the formation of methylcyclopentane with 80—94% selectivity and negligible skeletal isomerisation

TABLE. Reaction of n-hexane over the supported Pt-aggregate catalysts; C<sub>6</sub> reaction products<sup>a</sup>

Type of catalyst <sup>b</sup>	Dispersion (Pt wt %)	Reaction temp. (°C)	C <sub>6</sub> Reaction products (mol %)			
			2-MP	3-MP	MCP	CH + B
[Pt <sub>15</sub> (CO) <sub>30</sub> ][NEt <sub>4</sub> ] <sub>2</sub> <sup>c</sup>	0.69	295	17.2	8.2	61.7	12.9
		320	15.6	6.4	65.7	12.3
" " <sup>d</sup>	0.15	320	21.4	12.5	58.5	7.6
[Pt <sub>12</sub> (CO) <sub>24</sub> ][NEt <sub>4</sub> ] <sub>2</sub> <sup>c</sup>	0.54	275	15.2	6.8	64.1	13.9
[Pt <sub>9</sub> (CO) <sub>18</sub> ][NEt <sub>4</sub> ] <sub>2</sub> <sup>c</sup>	0.26	320	9.6	6.4	72.8	11.2
	0.50	320	11.4	5.2	74.4	9.0
[Pt <sub>6</sub> (CO) <sub>12</sub> ][NEt <sub>4</sub> ] <sub>2</sub> <sup>c</sup>	0.18	320	5.4	3.7	83.6	7.3
	0.25	325	5.2	2.6	87.0	5.2
[Pt <sub>3</sub> (CO) <sub>3</sub> (PPh <sub>3</sub> ) <sub>4</sub> ]	0.45	325	3.7	1.7	92.6	2.0

<sup>a</sup> 2-MP = 2-methylpentane, 3-MP = 3-methylpentane, MCP = methylcyclopentane, CH = cyclohexane, and B = benzene; only traces of 2,3-dimethylbutane and neohexane were detected. <sup>b</sup> The reaction was carried out after hydrogen reduction at 350 °C for 5 h and evacuation at 350 °C for 0.5 h. <sup>c</sup> On  $\gamma$ -Al<sub>2</sub>O<sub>3</sub>. <sup>d</sup> On Vycor glass. <sup>e</sup> The catalyst was prepared by hydrogen treatment at 250 °C of the dispersed [Pt<sub>3</sub>(CO)<sub>3</sub>(PPh<sub>3</sub>)<sub>4</sub>] on  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> and evacuation to remove the decomposition products, such as C<sub>6</sub>H<sub>6</sub> and Ph<sub>3</sub>PH.

When H<sub>2</sub> (50—250 Torr at 25 °C) was admitted into supported Pt-aggregates after pyrolysis *in vacuo* [designated here as (Pt<sub>3</sub>)<sub>n</sub>], it was found that all the Pt atoms of the dispersed Pt-crystallites displayed homogeneous hydrogen adsorption (H.A.), the stoichiometry of H per Pt atom on  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> or SiO<sub>2</sub> being 1.95, 1.7, 1.5, and 1.4 for Pt<sub>15</sub>, Pt<sub>12</sub>, Pt<sub>9</sub>, and Pt<sub>6</sub>, respectively. For CO adsorption on Al<sub>2</sub>O<sub>3</sub>, CO/Pt<sub>total</sub> = 0.45, 0.37, and 0.25 for Pt<sub>15</sub>, Pt<sub>9</sub>, and Pt<sub>6</sub>, respectively. Hydrogen titration (H.T.) of the oxygen adsorbed (O.A.) on the dispersed Pt-crystallites on Al<sub>2</sub>O<sub>3</sub> after evacuation was measured and the ratios for H.A.:O.A.:H.T. were found to be 2.5—3.7:1:3.8—4.2 for Pt<sub>15</sub>, Pt<sub>9</sub>, and Pt<sub>6</sub>. Although the Pt-aggregates here have not been proved to contain the same number of atoms as the clusters from which they are formed, the results indicate that they consist of crystallites of particle size <10 Å.<sup>2</sup> The H<sub>2</sub>-O<sub>2</sub> titration stoichiometries were reproduced in several runs after treatment with H<sub>2</sub> and evacuation at 350 °C.

The dehydrocyclisation and the skeletal isomerisation

of n-hexane occurred under these conditions. 2- and 3-methylpentanes were mainly formed from the simultaneous hydrogenolysis of the methylcyclopentane produced. It is also of interest that the skeletal isomerisation of neopentane proceeded, with less hydrogenolysis, over the Pt<sub>15</sub>- and Pt<sub>12</sub>-crystallites on  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> and Vycor glass at 270—340 °C under an excess of hydrogen. The ratio of isopentane:isobutane formed at lower conversion (<1%) was 5:20. In contrast, hydrocracking occurred preferentially over Pt-crystallites prepared from smaller Pt-clusters, such as Pt<sub>6</sub>- and Pt<sub>3</sub>-carbonyls, on  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> with negligible formation of isopentane (isopentane/isobutane = 0.1—0.01 at 310—350 °C). The results may suggest that the smaller Pt-aggregates prepared from the Pt-cluster compounds consist of lower-co-ordinated corner Pt atoms which favour the formation of adsorbed carbocyclic reaction intermediates as suggested by Anderson *et al.*<sup>3</sup>

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<sup>3</sup> J. R. Anderson, R. J. MacDonald, and Y. Shimoyama, *J. Catalysis*, 1971, **20**, 147.