## Novel Photochemical Aporphine Synthesis via Spirodienone Rearrangement: $(\pm)$ -Boldine

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Summary The first synthesis of  $(\pm)$ -boldine (5), by photocyclization of the  $(\pm)$ -bromodiphenol (1) to the  $(\pm)$ -spirodienone (3), followed by rearrangement to  $(\pm)$ -N-ethoxycarbonylnorboldine (4) and reduction with LiAlH<sub>4</sub>, is reported.

The past decade has witnessed the discovery of several photochemical syntheses of aporphine alkaloids, which have proceeded by several different mechanistic pathways.<sup>1</sup> We describe here a novel photochemical synthesis of  $(\pm)$ -boldine (5), which proceeds *via* the intermediacy of the proceythrinadienone (3).

Photolysis<sup>†</sup> of the  $(\pm)$ -bromodiphenol  $(1)^2$  in a solution of NaOH in absolute ethanol for 14 h gave the  $(\pm)$ -spirodienone (3), 34%,<sup>‡</sup> m.p. 183—184.5 °C (lit.<sup>3</sup> 144—146 °C),



and (±)-N-ethoxycarbonylnorboldine (4),§ 5%, m.p. 197– 199 °C (EtOH–Et<sub>2</sub>O), u.v.  $\lambda_{max}$  (EtOH) 304 (log  $\epsilon$  4·15), 284 (4·15), and 216 (4·61) nm; i.r.  $\lambda_{max}$  2·83 and 5·96  $\mu$ m;  $\delta$  (CDCl<sub>3</sub>) 7·96 (1H, s, H-11), 6·85 (1H, s, H-8), 6·70 (1H, s, H-3), 3·93 (3H, s, C-10 OMe), 3·59 (3H, s, C-1 OMe), 4·20 (2H, q, OCH<sub>2</sub>-Me), and 1·29 (3H, t, OCH<sub>2</sub>Me); m/e 385 (100%, M<sup>+</sup>), 355

(64), 340 (4), 323 (5), 310 (12), 296 (14), 283 (31), and 269 (27), along with  $(\pm)$ -*N*-ethoxycarbonylnorprotosinomenine (2), 15%, m.p. 153—154 °C (lit.<sup>3</sup> 148–150 °C) and the starting material (1), 8%. Treatment of (4) with LiAlH<sub>4</sub> in tetrahydrofuran under reflux for 18 h yielded  $(\pm)$ -boldine (5), isolated as the hydrobromide, 78%, m.p. 189—191 °C (free base, m.p. 159—162 °C). The u.v., n.m.r., and mass spectra were in good agreement with those reported for naturally occurring (+)-boldine.<sup>4</sup> When a solution of (1) and sodium acetate in absolute ethanol was irradiated for 15 h, (3) and (4) were isolated in 7 and 22% yield, respectively, along with (2), 22%, and the starting material (1), 9%.

These results led us to consider that photochemical transformation of the  $(\pm)$ -bromodiphenol (1) into  $(\pm)$ -N-ethoxycarbonylnorboldine (4) may proceed via the spirodienone intermediate (3). A solution of the  $(\pm)$ -spirodienone (3) and NaOH in absolute ethanol was irradiated for 3.5 h, whereupon (4) was isolated in 6% yield along with starting material, 50%. On the other hand, photolysis of (3) in a solution of sodium acetate in absolute ethanol for 4 h yielded (4), 44%, along with recovered (3), 10%. This ready photochemical rearrangement of (3) to (4) supports the proposed intermediacy of the spirodienone in the photochemical conversion of the  $(\pm)$ -bromodiphenol (1) into (+)-N-ethoxycarbonylnorboldine (4) and constitutes the first reported synthesis of an aporphine via rearrangement of a proerythrinadienone.¶ Furthermore, the demonstrated sequence  $(1) \rightarrow (3) \rightarrow (4) \rightarrow (5)$  constitutes the first total synthesis of  $(\pm)$ -boldine.

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† Photolyses were carried out in a quartz vessel under  $N_2$  and irradiation with GE G15T8 germicidal lamps. Products were separated by preparative t.l.c. using plates pre-coated with silica gel 60 F-254 (EM Reagents).

<sup>‡</sup> The spirodienone (3) was obtained in ca. 2% yield as a powder by phenolic oxidation<sup>3</sup> and ca. 18% yield as an oil by photolytic synthesis<sup>2</sup> under somewhat different conditions. The product with m.p. 183—184.5 °C was characterized as (3) by comparison of the u.v., i.r., n.m.r., and mass spectra with those reported in ref. 3.

§ All new compounds were characterized by concordant analytical and spectral data. The structural formulae containing asymmetric atoms refer to racemic compounds.

 $\P$  An unsuccessful attempt to convert the spirodienone (3) into an aporphine derivative by acid-catalysed rearrangement has been reported (ref. 2).

<sup>1</sup> For a recent review, see M. Shamma, 'The Isoquinoline Alkaloids,' Academic Press, New York, 1972, ch. 10; S. M. Kupchan and P. F. O'Brien, J.C.S. Chem. Comm., 1973, 915.

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<sup>4</sup> M. Shamma, Experientia, 1960, 16, 484; A. H. Jackson and J. A. Martin, J. Chem. Soc. (C), 1966, 2181 and 2222.