

Stereochemical Dependence of Geminal Selenium–Carbon Coupling Constants

By HANS J. REICH* and JOHN E. TREND

(Department of Chemistry, University of Wisconsin, Madison, Wisconsin 53706)

Summary Geminal ^{13}C – ^{77}Se and vicinal ^1H – ^{77}Se n.m.r. couplings are stereospecific in a selenide, selenoxide, selenonium salt, and selenonium ylide; coupling is largest to the nuclei nearest the selenium lone pair(s).

Our interest in organoselenium stereochemistry¹ as well as in the application of organoselenium reagents in organic chemistry² has prompted us to measure n.m.r. properties of a variety of selenium-containing compounds. In the course of these studies we observed a pronounced stereochemical dependence of 2J (^{13}C - ^{77}Se) and 3J (H-Se) in several conformationally rigid systems.³

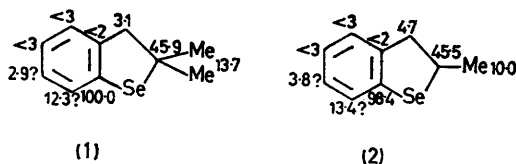
compounds the ^{13}C assignments were compatible with an upfield steric γ -shift⁶ for the Me group *cis* to the substituent on selenium. In each case the three geminal C-Se couplings follow the same pattern: one methyl carbon atom (the one eclipsed relative to the lone pair) is substantially more strongly coupled than the other two. The selenides (1) and (2) also fit this pattern. Three-bond proton-selenium coup-

TABLE. ^1H and ^{13}C N.m.r. data† for (3a-c)

| | (3a) ^a | | | | (3b) ^b | | | | (3c) ^c | | | |
|------------------|---------------------------|---------------------|--------------------|--------------------|---------------------------|---------------------|--------------------|--------------------|---------------------------|---------------------|--------------------|--------------------|
| | δ_{H} | δ_{C} | $^3J(\text{H-Se})$ | $^2J(\text{C-Se})$ | δ_{H} | δ_{C} | $^3J(\text{H-Se})$ | $^2J(\text{C-Se})$ | δ_{H} | δ_{C} | $^3J(\text{H-Se})$ | $^2J(\text{C-Se})$ |
| <i>cis</i> -Me | 1.93 | 24.0 | 10.5 | <6 | 1.65 | 21.5 | 9.5 | 8.4 | 1.62 | 24.6 | 12.0 | 13.7 |
| <i>trans</i> -Me | 1.80 | 26.9 | 23.0 | 25.7 | 1.26 | 23.6 | 19.0 | 26.9 | 1.71 | 28.3 | 24.0 | 34.7 |
| CH_2 | 3.53 ^d 3.87 | 46.6 | | <5 | 3.13 ^d 3.71 | 47.9 | | <5 | 3.17 ^d 4.04 | 52.2 | | <8 |

^a In CD_2Cl_2 solution at 25 °C. ^b In CDCl_3 solution at -15 °C. ^c In CDCl_3 solution at 25 °C. ^d J_{gem} 16-17 Hz.

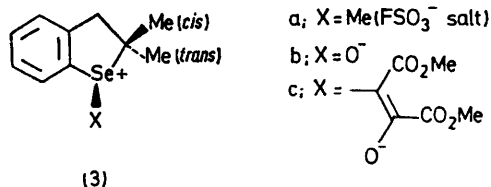
Selenium-carbon couplings in two dihydrobenzoselephen derivatives (1) and (2) are shown in the Figure.† The geminal couplings to selenium vary substantially: coupling is large to the CH_2 and one *ortho* carbon atom, but small to the CH_2 and the other *ortho* carbon atom.

FIGURE. J (Se-C) coupling constants (in Hz) for (1) and (2).

More striking examples are provided by the selenonium salt (3a), the selenoxide (3b) and the ylide (3c) (Table). All compounds are pyramidal at selenium and configurationally stable on the n.m.r. time-scale, as shown by the observation of diastereotopic ^1H and ^{13}C signals. Assignments for (3a) were established for the ^1H resonances by homonuclear NOE-INDOR experiments,⁴ and correlation of ^1H and ^{13}C resonances by single-frequency off-resonance decoupling, $^{13}\text{C}\{\text{H}\}$. Assignments for (3b) were made on the basis of the larger europium complex-induced shifts for the proton and carbon *cis*-methyl resonance.⁵ In all three

lings follow the trend set by the carbon couplings, although stereochemical dependence is less pronounced.

We feel that these regularities, if found to be general, may be an important aid for determining structures and conformations of organoselenium compounds, and, by analogy, for organosulphur compounds for which such n.m.r. information is not available.



Several reports of stereospecific 2J (^{13}C - ^{31}P) for phospholens,⁷ phosphetans,⁸ and acyclic phosphines⁹ have appeared. These couplings show remarkably similar magnitude and stereochemistry to those reported above for 3-co-ordinate selenium.

Acknowledgement is made to the donors of the Petroleum Research Fund, administered by the American Chemical Society, for support of this research.

(Received, 6th February, 1976; Com. 123.)

† Signs of coupling constants were not determined.

¹ H. J. Reich, *J. Amer. Chem. Soc.*, 1973, **95**, 964; H. J. Reich and J. E. Trend, *Canad. J. Chem.*, 1975, **53**, 1922.

² H. J. Reich, J. M. Renga, and I. L. Reich, *J. Amer. Chem. Soc.*, 1975, **96**, 5434.

³ One-bond Se-C couplings have been reported: W. McFarlane and D. S. Rycroft, *J.C.S. Chem. Comm.*, 1973, 10.

⁴ W. von Phillipsborn, *Angew. Chem. Internat. Edn.*, 1971, **10**, 472; J. H. Noggle and R. E. Schirmer, 'The Nuclear Overhauser Effect, Chemical Applications,' Academic Press, New York, 1971, p. 96.

⁵ Sulphoxides complex through oxygen: R. R. Fraser, T. Durst, M. R. McClary, R. Vian, and Y. Y. Wigfield, *Internat. J. Sulfur Chem. A.*, 1971, **1**, 133.

⁶ Upfield γ -shifts have been observed for phospholens (ref. 7) and sulphoxides (G. W. Buchanan and T. Durst, *Tetrahedron Letters*, 1975, 1683).

⁷ J. J. Breen, S. I. Featherman, L. D. Quin, and R. C. Stocks, *J.C.S. Chem. Comm.*, 1972, 657.

⁸ G. A. Gray and S. E. Cremer, *J.C.S. Chem. Comm.*, 1972, 367.

⁹ M.-P. Simonnin, R.-M. Lequan, and F. W. Wehrli, *J.C.S. Chem. Comm.*, 1972, 1204; S. Sørensen, R. S. Hansen, and H. J. Jakobsen, *J. Amer. Chem. Soc.*, 1972, **94**, 5900.