A Distorted Trigonal-bipyramidal Phosphorane with a Diequatorial Four-membered Ring; X-Ray Crystal Structure of 4-p-Bromophenyl-2',2',3',4',4'-pentamethyl-3,3,6,6-tetrakistrifluoromethyl-2,7-dioxa-1-phospha(v)bicyclo[3,2,0]hept-4-ene-1-spiro-1'-phosph(v)etan

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(3), C(6)

1), C(6)

Summary Hexafluoroacetone and 1-phenylethynylphosphetan gave a 2:1 adduct shown by X-ray structure analysis to be a five-co-ordinate phosphorane having distorted trigonal bipyramidal geometry with a diequatorial phosphetan ring.

PREVIOUSLY¹ we have treated a number of phosphetans (1) with hexafluoroacetone (HFA) and studied the variable temperature ¹H n.m.r. spectra of the resulting adducts (2) in order to obtain data on the relative apicophilicities of the groups R. In continuation of these studies, the *trans*phenylethynylphosphetan (1; R = PhC:C) was treated with an excess of HFA at -78 °C in dichloromethane. The major product, m.p. 160 °C, was a 1:2 adduct† [τ (CDCl₃) 2:6-3:15 (5H, m), 7:72-7:98 (1H, m), 8:62 (6H, d, J 24 Hz),



8.70 (6H, d, J 24 Hz), and 9.10 (3H, dd, J 7 and 1.5 Hz), ³¹P (CDCl₃) 31.6 p.p.m. downfield from external 85% H₃PO₄, ¹⁹F (CDCl₃) 9.70 (6F, s) and 11.52 (6F, d, J 4 Hz) p.p.m. upfield from internal PhCF₃], which showed no acetylenic absorption in its i.r. spectrum. We were unable to assign a convincing structure to this adduct and accordingly prepared the analogous p-bromophenyl compound,[†] m.p. 176—177 °C ³¹P (CDCl₃) 30 p.p.m. downfield from external 85% H₃PO₄,[‡] and submitted it to single crystal X-ray analysis.

Crystal data: $C_{22}H_{20}BrF_{12}O_2P$, M = 655, orthorhombic, $Pca2_1$, a = 17.77, b = 8.71, c = 16.51 Å, U = 2555 Å³, Z = 4, $D_c = 1.70$ g cm⁻³. 890 Unique reflection intensities were measured between $0.1 < \sin \theta/\lambda < 0.63$ using monochromatic Mo- K_{α} ($\lambda = 0.7107$ Å) radiation and have been refined to R = 0.062 with anisotropic thermal parameters

TABLE

Bond angles (°) about phosphorus atom in $P(C_3HMe_5)[OC(CF_3)_2C(C_6H_4Br)CC(CF_3)_2O]$. Estimated standard deviations are 0.9°. Bonded atoms Angle Bonded atoms Angle O(1), O(2)159.7O(1), C(1) 10Ĩ·0 C(1), C(3) 82.9 O(1), C(3) **99**.0 O(2), C(6) C(1), C(6)133.4 71.6

141.1

88.1

for P, Br, and F atoms, and all hydrogen atoms included. The structure is shown in the Figure with selected bond lengths, and in formula (3). Despite the ³¹P n.m.r.

O(2), C(1)

O(2), C(3)

94.3

96·0

lengths, and in formula (3). Despite the ${}^{31}P$ n.m.r. chemical shift the compound is a five-co-ordinate phosphorane. From the angles given in the Table the geometry round the phosphorus atom is that of a distorted



FIGURE. The molecular structure of $P(C_{3}HMe_{5})[OC(CF_{3})_{2}-C(C_{6}H_{4}Br)CC(CF_{3})_{2}O]$. E.s.d.'s of bond lengths shown are P-O, 0.015; P-C 0.02 Å; C-O, C-C, 0.025 Å.

trigonal bipyramid with the phosphetan ring occupying two equatorial positions. Obviously the equatorial angles depart from the ideal 120° values to accommodate the fourmembered ring, otherwise distortions are small. The two apical atoms are bent slightly away from the phosphetan ring. The molecule has a plane of symmetry defined by the two apical and the third equatorial atoms and dictated by the CC double bond, the largest displacement out of this plane for the seven atoms of this PO_2C_4 system and the

† This compound gave correct elemental analysis and showed the expected mass peak and fragmentation pattern in its mass spectrum.

[‡] The other spectral properties were similar to those of the phenyl analogue.

attached phenyl carbon being 0.06 Å. The phenyl ring is perpendicular to this plane (88.6°) , presumably to minimise steric repulsions with the trifluoromethyl groups. Large variations in bond lengths within the rings is evidence of ring strain. The distortion from trigonal bipyramidal geometry is towards that of a 0° (2 + 3) structure² with the 'pair' being the O and C of the oxaphosphetan ring and the geometry could equally well be described in these terms.

The reaction between the phosphetans (1 R = ArC:C)and HFA presumably proceeds via the ylides (4) as shown.³

We have postulated^{1,4} that trigonal bipyramidal phosphoranes having four- or five-membered diequatorial rings are high energy intermediates or transition states in the

¹ R. K. Oram and S. Trippett, J.C.S. Perkin I, 1973, 1300. ² P. Gillespie, P. Hoffman, H. Klusacek, D. Marquarding, S. Pfohl, F. Ramirez, E. A. Tsolis, and I. Ugi, Angew. Chem. Internat. Edn., 1971, 10, 687; P. Gillespie, F. Ramirez, I. Ugi, and D. Marquarding, *ibid.*, 1973, 12, 91.

⁸ E. Duff, S. Trippett, and P. J. Whittle, *J.C.S. Perkin I*, 1973, 972. ⁴ S. A. Bone, S. Trippett, M. W. White, and P. J. Whittle, *Tetrahedron Letters*, 1974, 1795: S. A. Bone, S. Trippett, and P. J. Whittle, J.C.S. Perkin I, 1974, 2125.

Berry pseudo-rotations of a variety of spirophosphoranes, although the opinion has been expressed² that such diequatorial placement of small-membered rings is not possible and the turnstile rotation mechanism for the permutational isomerisation of phosphoranes was developed largely in order to avoid such placements. Ideal trigonal bipyramidal geometry was never assumed or expected in strained phosphoranes and structure (3) shows how little distortion from this is necessary in order to accommodate a diequatorial four-membered ring.

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