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Synthesis of Methyl Adirubine

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Summary Starting from the cyclopentene (2) and methyl tryptophanate, a total synthesis of methyl adirubine (1b) has been completed.

A MEMBER of the rapidly growing family of 5-carboxy indole alkaloids, adirubine, was the first corynanthé base to be isolated in which the α -amino-acid unit of precursor tryptophan remains intact.¹ We report a synthesis of methyl adirubine (**1b**) which, in addition to being the first of a corynanthé representative for this class, also permits elaboration of the stereostructure of adirubine to (**1a**).†

The (\pm) -diol ether (3a),² a mixture of diastereoisomers best prepared by hydroxylation of (2) with KMnO₄³ rather than OsO₄,² was converted (98%) into the (\pm) -acetal (3b) (b.p. 140 °C at 0.5 mmHg) by means of acetone-2,2-dimethoxypropane (toluene-*p*-sulphonic acid catalyst). After catalytic hydrogenolysis (Pd) of the ether group in (3b) to give the alcohol (3c) (100%) (b.p. 130 °C at 0.5 mmHg), Collins oxidation⁴ produced (85%) the relatively unstable (\pm) -acetal aldehyde (3d) [δ (CDCl₃; 60 MHz) 0.91 (3H, t, CHCMe₂), and 9.38 and 9.45 (1H, 2 × s, CHO diastereoisomers)].

The amino-acid unit was attached by sodium cyanoborohydride-promoted⁵ reductive alkylation of the aldehyde (**3d**) with methyl L-tryptophanate in acetonitrile at room temperature. The resulting glassy mixture of the diastereoisomers of (**4a**) (55%) was treated immediately with H₂O-MeOH-HCl to give, after 5 days at room temperature, the diol diastereoisomers (**4b**) in 73% yield (71%)



 \dagger On the basis of spectral and other considerations, R. T. Brown and his co-workers (ref. 1b) previously suggested the C-3, C-5, and C-15 relationship portrayed in (1).

conversion) after t.l.c. (silica gel). Without further purification, the diol mixture was cleaved by periodic acid in tetrahydrofuran-aqueous MeCO₂H-MeCO₂Na buffer, and



after cyclization in situ of the intermediate dialdehyde to the tetracyclic aldehyde (5a), silver oxide oxidation provided the acid (5b), which was converted without purification into the dimethyl ester (5c) by MeOH-Me₂C(OMe)₂-HCl. After preliminary separation of the diastereoisomeric diester mixture [40% overall from the diol (4b)] by preparative layer chromatography (p.l.c.) on silica gel, multiple elution p.l.c. provided five relatively pure diastereoisomers of (5c). One of these, (5c') [m.p. $253 \cdot 5$ — 255 °C, m/e 384 0531 (M^+)], bore a striking spectral resemblance to an authentic sample of methyl adirubine:1b it displayed Bohlmann bands in the i.r. region (2890, 2870, 2810, and 2775 cm⁻¹); the c.d. spectrum (dioxan) revealed a strong positive Cotton effect, maximum 273 nm [$\Delta \epsilon$ + 4560; and n.m.r. (100 MHz) peaks were located at inter alia δ 0.91 (3H, t, CH₂Me), 3.74 and 3.85 (3H, 2 × s, CO₂Me), ca. 7.2 (5H, m, indole CH), and 7.82 (1H, s, indole NH); all characteristics consonant with structure (5c') having the α, α arrangement of hydrogen atoms at C-3 and C-15. Although the n.m.r. method of Trager et al.6 for stereochemical assignments at C-15 and C-20 in the indole alkaloid series was not applicable to methyl adirubine and its relatives, this spectral approach did permit cis assignment

of these two centres in the ester (5c'). As in the case of various corynantheidine (allo) derivatives,⁶ (5c') features a characteristically shaped, well defined C-18 methyl triplet at δ 0.91, very different from the corresponding signals in the n.m.r. spectra of dihydrocorynantheine (normal) and related cases.⁶ [We have shown that the tetracycles (6) and (7) give the n.m.r. patterns expected from ref. 6]. The configuration at C-5 follows from the stereochemical nature of the starting tryptophan.[‡]

Completion of the synthesis of (1b) involves C-16 formylation of (5a) followed by reduction⁷ of the intermediate β -hydroxyacrylate to β -hydroxypropionate, achieved by treatment with lithium dicyclohexylamide-methyl formate $in\ tetrahydrofuran-hexamethyl phosphoric\ triamide\ followed$ by sodium borohydride in MeOH. After t.l.c. (silica gel) of the product mixture, the two C-16 epimers corresponding to (1b) were isolated, one of them $[9\cdot3\%$ from (5c')] being identical in all respects (n.m.r., i.r., m.s., Rf, and m.p.) with an authentic sample of methyl adirubine. Accordingly, the stereochemistry depicted in (1b) follows from the foregoing findings.§

On a purely chemical basis, the biosynthesis of sarpagine, ajmaline, and related cases is better rationalized as proceeding via C-5 oxidative decarboxylation of an adirubine type molecule rather than via dehydrogenation of a tryptamine-derived tetracycle, to a 4-5 iminium ion, either process being followed by addition of the nucleophile at C-16 to C-5.8 The stereochemical relationship between adirubine and these other systems which is now apparent favours the former proposition, which can readily be tested through feeding of appropriate radio-labelled substances of the adirubine type.

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 \ddagger On the basis of similar spectral information, the most abundant product of the sequence (4b) \rightarrow (5c) is regarded as having the pseudo stereochemistry (C-3 β , C-15 α , and C-20 β). The data for the remaining three diastereoisomers which, along with (5c'), were formed only in small yield, were not sufficient for stereochemical assignments.

§ In a personal communication, Dr. R. T. Brown indicated that similar conclusions regarding the stereochemistry of adirubin were reached on the basis of a correlation with 5α , 20α -methoxycarbonyldihydromancunine.

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