## X-Ray Structure of Cannabispiran: a Novel Cannabis Constituent

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Summary Cannabispiran, a spiro-compound with a novel skeletal class was isolated from the leaves of Cannabis sativa L. and X-ray crystallographic studies indicated that the compound was 7'-hydroxy-5'-methoxyspiro-(cyclohexane-1,1'-indan)-4-one (1).

Combined gas chromatographic—mass spectrometric analysis of the alcoholic extract of an Indian Cannabis (marihuana) variant revealed the presence of a novel constituent which had the same relative retention time  $^1$  as (—)- $\Delta^8$ -transtetrahydrocannabinol both before silylation (0.43) and after silylation (0.21). This compound was isolated by adsorption chromatography on silica gel and was obtained as colourless crystals, m.p. 178—179 °C, and was optically inactive. Analysis indicated the formula  $\rm C_{15}H_{18}O_3$  which was confirmed by high-resolution mass spectral data.

The i.r. spectrum showed major peaks at  $\nu_{\rm max}$  (CHCl<sub>3</sub>) 3600, 3335, 1715, 1602, and 1503 cm<sup>-1</sup> while the <sup>1</sup>H-n.m.r. spectrum (100 MHz; CDCl<sub>3</sub>) did not show any resonances due to methyl groups but did reveal a methoxy-singlet at  $\delta$  3·75; furthermore, it exhibited a pair of spin-coupled 2H triplets at  $\delta$  2·22 and 2·95 (J 7·0 Hz), an 8H multiplet centred at  $\delta$  2·56, an exchangeable signal at  $\delta$  5·24, and two 1H doublets at  $\delta$  6·12 and 6·36 (J 1·8 Hz) which were ascribed to the two aromatic protons.

The paucity of the compound (yields from the plant were ca. 0.002% by dry weight) precluded its structure elucidation by chemical means and thus an X-ray crystallographic study was undertaken. Crystal data: space group Pbca, orthorhombic, a=10.388(5), b=14.754(7), c=16.950(8) Å, U=2597.7(9) ų, Z=8, F(000)=1056. The structure determination was carried out using 852 reflections with  $I>2\sigma(I)$  collected by counter methods, using Mo- $K_{\alpha}$ -radiation ( $2\theta_{\rm max}=45$ ). The phase problem was solved by direct methods.² The structure model was refined to an R factor of 0.15. Inclusion of the hydrogen atoms in calculated positions and full-matrix least-squares refinement of all positional parameters, anisotropic thermal parameters for all non-hydrogen atoms and isotropic thermal parameters for hydrogen atoms, resulted in an R factor

of 0.074 and weighted  $R_{\rm w}$  of 0.054. Atomic parameters and a listing of observed and calculated structure factors are available from the authors upon request. All programs used are part of a local assembly of computer programs for CYBER-74 which have been described.<sup>3</sup> A view of the molecule is shown in the Figure. The indan system except for C(8), which deviates by 0.3 Å, is planar. Selected dihedral angles around C(9) are as follows (+ve  $\equiv$  right-handed screw): C(7)-C(8)-C(9)-C(2)  $-26.0(7)^{\circ}$ , C(1)-C(2)-C(9)-C(10)  $135.7(7)^{\circ}$ , and C(1)-C(2)-C(9)-C(14)  $-102.6(7)^{\circ}$ .

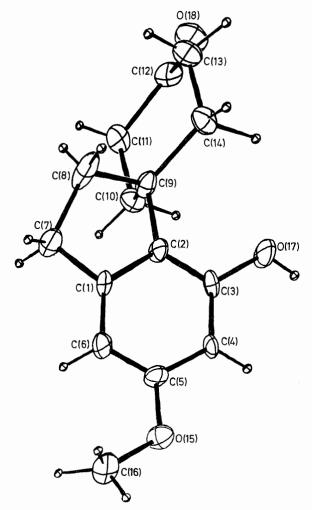


FIGURE. A view of the molecule of (1) down on the plane of the phenyl ring. The thermal ellipsoids are shown with 25% probability. The hydrogen atoms are drawn artificially small.

We have named the compound cannabispiran (1) and its structure is related to the synthetic compound spiro(cyclohexane-1,2'-indan)-1',4-dione which was reported to potentiate the estrogenic activity of stilbestrol.4

This work was supported by a Contract from the National Institute on Drug Abuse and by the Research Institute of Pharmaceutical Sciences of the University of Mississippi. (Received, 12th May 1976; Com. 533.)

<sup>1</sup> Gas chromatography was performed on a column of 2% OV-17 on 100/120 mesh Gas Chrom Q at 210 °C using androst-4-ene-3,17-dione as internal standard as described by C. E. Turner and K. Hadley, J. Pharm. Sci., 1973, 62, 251.

<sup>2</sup> G. Germain, P. Main, and M. Woolfson, Acta Cryst., 1971, A27, 368.

<sup>3</sup> P. Groth, Acta Chem. Scand., 1973, 27, 1837.

<sup>4</sup> D. J. Bailey, N. S. Doggett, L. Y. Ng, and T. Qazi, J. Medicin. Chem., 1975, 19, 438.