## Conversion of Dialdehydes into Cyclic $\alpha$ -Ketols by Thiazolium Salts: Synthesis of Cyclic 2-Hydroxy-2-enones

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Summary Pentanedials and hexanedials are cyclised by thiazolium salts and triethylamine in acetonitrile into 2-hydroxy-cyclopentanones and -cyclohexanones, easily oxidised to the 2-hydroxy-2-enones.

The thiazolium ion-catalysed dimerisation of aldehydes to acyloins, modelled on the cyanide ion-catalysed benzoin

condensation<sup>2</sup> and particularly on reactions catalysed by enzymes containing thiamine,<sup>3</sup> has hardly been used in synthesis, probably because it seems applicable only to symmetrical acyloins.<sup>4</sup> We point out that thiazolium ion-catalysed cyclisation of dialdehydes is a useful synthesis of cyclic  $\alpha$ -ketols and thence of 2-hydroxy-2-enones, several of which are important flavouring materials.<sup>5</sup>

Treatment of anhydrous glutaraldehyde (Ia) (4.44 mmol) in acetonitrile with 3-benzyl-4-methylthiazolium chloride (0.44 mmol) and triethylamine (0.88 mmol) at 80 °C gave 2-hydroxycyclopentanone in 78% yield, which was oxidised

TABLE Yields of 2-hydroxycyclopentenones formed on cylisation and oxidation of substituted pentanedials

Pentanedial (I)			% Yield of corresponding 2-hydroxycyclopentenone (II)	
	$\mathbf{R^{1}}$	$\mathbb{R}^2$	${f R^3}$	
a	H	H	H	63
b	$\mathbf{H}$	H	Me	63
С	Me	Me	H	88
d	Me	H	Me	74
e -CH <sub>2</sub> CH=CHCH <sub>2</sub> CH <sub>2</sub> - H			- H	68

by copper(II) acetate in a mixture of methanol, water, and acetic acid to 2-hydroxycyclopent-2-en-1-one (IIa) (81%). A range of substituted glutaraldehydes was synthesised by addition of the appropriate  $\alpha\beta$ -unsaturated aldehyde to an enol ether6 or preferably enamine7 and hydrolysis of the resulting dihydropyran. 2-Hydroxy-3-methylcyclopent-2en-1-one (IIb) ('corylone', 'cyclotene' etc.) was prepared in 50% overall yield from prop-2-enal and 1-piperidinopropene as illustrated in the Scheme. The Table records the yields of other hydroxycyclopentenones formed on cyclisation and oxidation of the appropriate dialdehyde.

Unfortunately, the isomeric a-ketols formed in the cyclisation of an unsymmetrical dialdehyde are equilibrated by the triethylamine during the reaction, so that in its present form the thiazolium ion-catalysed cyclisation cannot be used to make one specific isomer. (This does not matter, of course, if the  $\alpha$ -diketone is required.)

The thiazolium salt and triethylamine in acetonitrile cyclised hexanedial (III) into 2-hydroxycyclohexanone (IV) in 78% yield which rapidly dimerised to a mixture of the hemi-acetals (V).

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