Cycloaddition of Dimethyl Acetylenedicarboxylate with 2-Imino-1,3-oxathioles

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Summary 2-Phenylimino-1,3-oxathioles (1), acting as masked 1,3-dipoles with carbon as the central atom, add dimethyl acetylenedicarboxylate to give aryl propionitriles (4) and dimethyl 2,3-dihydro-3-phenyl-2oxothiazole-4,5-dicarboxylate (3), probably as a result of thermal decomposition of a 1:1 cycloadduct (2).

In the course of our study of the addition of thiocarbonyl compounds to gem-dicyano epoxides we prepared the imino-oxathioles (1).2 The mesomeric form (1') shows that these compounds are potential 1,3-dipoles with carbon as the central atom. Compounds (1) are therefore similar to 3-imino-1,2,4-dithiazoles or 3-imino-1,2,4-thiadiazoles which have been shown recently to add dipolar ophiles.3-5† We have now shown that the imino-oxathioles (1) can also add dimethyl acetylenedicarboxylate (100 h in boiling xylene) to give the thiazolone (3), m.p. 125 °C, ν_{max} 1752, 1720, and 1707 cm⁻¹; $\delta(CDCl_3)$ 3.7 (3H, s) and 3.87 (3H, s), m/e 293.057 (99% yield), and the aryl propiolonitriles (4)⁸ [(4a) m.p. 41 °C, ν_{max} 2250, 2200, and 2146 cm⁻¹; (4b), m.p. 140 °C, ν_{max} similar to (4b)] (Scheme). These results may be rationalized by assuming that, in the first step, the reaction leads to the bicyclic tetravalent sulphur compound (2) which decomposes in boiling xylene leading to (3) and (4). In similar reactions described in the literature³⁻⁵ analogous tetravalent sulphur compounds were isolated only in the work reported in ref. 3, where the reaction was carried out at 70 °C for 1-2 h. Methyl maleate or methyl fumarate failed to react with (1). We are now attempting to prove whether or not this reaction is truly a 1,3-dipolar cycloaddition.

a; X = H $b; X = NO_2$

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† These reactions are claimed to be 1,3-dipolar cycloadditions but only ref. 5 describes the cis-addition of methyl maleate and methyl fumarate with 3-imino-1,2,4-thiadiazoles.

A. Robert and M. Baudy, unpublished results.
 K. Akiba, T. Tsuchiya, M. Ochiumi, and N. Inamoto, Tetrahedron Letters, 1975, 455.

⁴ K. Akiba, M. Ochiumi, T. Tsuchiya, and N. Inamoto, Tetrahedron Letters, 1975, 459; J. E. Oliver and A. B. De Milo, J. Org. Chem., 1974, 39, 2225; J. E. Oliver and R. T. Brown, ibid., p. 2228, and references cited therein.

⁵ J. Goerdeler and H. W. Linden, Tetrahedron Letters, 1975, 3387.

⁶ The phenyl propiolonitrile (4; X = H) was not purified, but was identified by g.l.c. comparison with an authentic sample: I. Kazunari and N. Toshio, Yuki Gosei Kagaku Kyokai sli, 1970, 28, 80 (Chem. Abs., 1970, 72, 900,001q).

¹ M. Ferrey, A. Robert, and A. Foucaud, Compt. rend., 1973, 277C, 1153; Synthesis, 1976, 261; M. Baudy and A. Robert, I.C.S. Chem. Comm., 1976, 23.