Synthesis of a 3*H*-Pyrido[1,2-*b*]pyridazin-3-one from Pyridine *N*-Imide and Methylphenylcyclopropenone

By Albert Kascheres* and Decio Marchi Jr.

(Universidade Estadual de Campinas, Instituto de Quimica, CP 1170, Campinas—SP, Brasil 13.100)

Summary Pyridine N-imide (1a) reacts with methylphenyl-cyclopropenone (2) in methanol to give a β -aminoester (3) and the first reported 3H-pyrido[1,2-b]pyridazin-3-one (4); reaction in dichloromethane permitted isolation of the intermediate (5a).

Pyridine N-Imide (1a) has been shown to react as a nucleophile with diphenylcyclopropenone in methanol to produce methyl α -phenyl- β -amino-trans-cinnamate in high yield. In the course of investigating the mode of ring opening of the unsymmetrical methylphenylcyclopropenone² (2) in methanol, we have observed the formation of both the β -amino-ester (3) (31%, oil, unstable with respect to hydrolysis to the β -ketoester) and the 3H-pyrido[1,2-b]pyridazin-3-one (4) (67%, m.p. 201-203 °C). When the reaction was carried out in dichloromethane for 17 h, an intermediate (5a) (69%) was isolated in addition to (4) (27%). On standing, (5a) was readily oxidized to (4). This intermediate was also observed in methanol during shorter reaction times. Evidence for the isomer with a phenyl group in the 4-position was obtained from the n.m.r. spectrum of (5a), δ 3.75 (1H, d, H-4). Reaction of (1b) with (2) afforded (5b) (19%) the n.m.r. spectrum of which showed a singlet (1H) at δ 3.40, thus confirming the assignment. Treatment of (4) with an excess of sodium borohydride in ethanol produced (6) (61%, m.p. 164-166 °C).

This reaction of (2) apparently involves participation of (1a) not only as a nucleophile, as in the case of diphenyl-cyclopropenone, but also as a 1,3-dipolar compound. This result may also be contrasted with those obtained previously in the reactions of diphenylcyclopropenone with pyridine N-(acylimides).³ In these cases, reaction led exclusively to formation of 2,4,5-trisubstituted-6H-1,3-oxazin-6-ones by a pathway involving initial nucleophilic attack of the imide on the cyclopropenone ring followed by elimination of pyridine. Thus, the present report represents the first example of a possible 1,3-dipolar addition of a pyridine

N-imide to a cyclopropenone, further illustrating the effect of substituents on the reactivity of the cyclopropenone ring.⁴

The authors acknowledge financial assistance of Financiadora de Estudos e Projetos.

(Received, 19th December 1975; Com. 1403.)

- ¹ A. Kascheres and D. Marchi Jr., J. Org. Chem., 1975, 40, 2985.
- ² A. Krebs and J. Breckwoldt, Tetrahedron Letters, 1969, 3797.
- ³ T. Sasaki, K. Kanematsu, and A. Kakchi, J. Org. Chem., 1971, 36, 2451; J. W. Lown and K. Matsumoto, Canad. J. Chem., 1972, 50, 584.
- For a recent review of cyclopropenone chemistry, see K. T. Potts and J. S. Baum, Chem. Rev., 1974, 74, 189.