

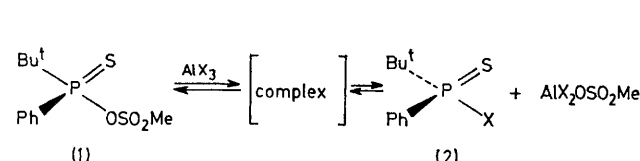
Novel Reaction of Phosphinothioic Methanesulphonic Anhydride with Aluminium Halides. Stereoselective Synthesis of Phosphinothioic Halides

By JAN MICHALSKI* and ZBIGNIEW SKRZYPCZYŃSKI

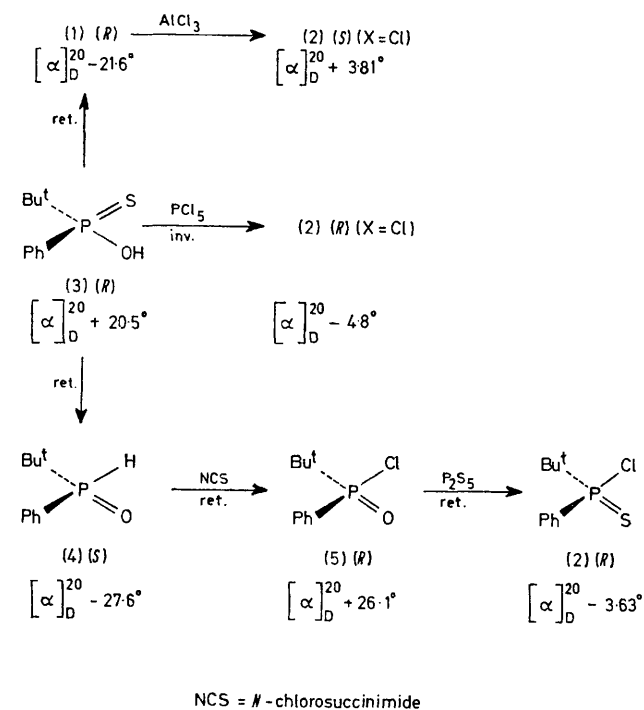
(Centre of Molecular and Macromolecular Studies, Polish Academy of Sciences, 90-362 Łódź, Boczna 5, Poland)

Summary Phosphinothioic methanesulphonic anhydrides (**1**) react with aluminium chloride or bromide to form complexes which after decomposition with water or methanol give halides (**2**) in high yield, with retention of configuration at phosphorus in the case of aluminium chloride.

IN connection with our studies of reactions of acid-catalysed nucleophilic displacement at four-co-ordinated phosphorus centres¹ we now report a novel displacement reaction with the aid of aluminium halides.



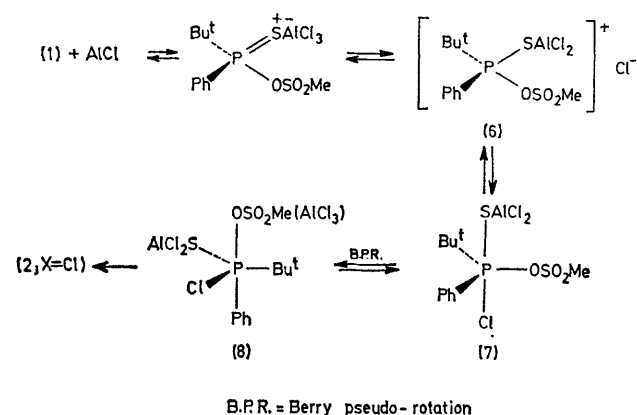
Racemic *t*-butylphenylphosphinothioic methanesulphonic anhydride (**1**) reacts with a small excess of aluminium chloride or bromide at ambient temperature in CH_2Cl_2 , over 10 min, to give a complex which after decomposition by



SCHEME 1

water or methanol and extraction with chloroform gives a high yield of corresponding chloride (**2**; X = Cl), or bromide (**2**; X = Br), finally purified by sublimation *in vacuo*. The stereochemistry of the reaction was studied using optically active (**1**) (Scheme 1).

Since the anhydride (**1**) is of the same configuration at the phosphorus atom as the starting phosphinothioic acid (**3**), the sign and rotation value of the chloride (**2**; X = Cl) are indicative of retention of configuration at phosphorus with relatively high stereoselectivity.



SCHEME 2

This assignment follows from the following stereochemical correlations. The chloridate (**2**; X = Cl) can be obtained by two independent routes of well established stereochemistry. The reaction of the acid (**3**) with phosphorus pentachloride leads to the chloride of the opposite configuration. This type of reaction is known to proceed with inversion of configuration at the phosphorus atom.² The chloride of the opposite configuration was also obtained by a series of reactions involving desulphurization of the thioic acid (**3**) to *t*-butylphenylphosphine oxide (**4**),^{3,4} chlorination of (**4**) by *N*-chlorosuccinimide to the phosphinic chloride (**5**)^{3,5} and finally replacement of oxygen by sulphur (P_2S_5).^{5,6} All reactions of this series proceed with retention of configuration at the phosphorus centre. However the ligand metathesis, equivalent to an additional inversion, results in formation of (**2**) of opposite configuration to that observed in the reaction of (**1**) with aluminium chloride. It is of interest to note that optically active (**2**) undergoes slow racemisation in the presence of AlCl_3 under the reaction conditions lowering somewhat the stereoselectivity of the reaction under discussion.

These findings suggest that reaction of aluminium chloride with the anhydride (**1**) proceeds *via* complexation†

† A similar complex is presumably formed with aluminium bromide but no reaction is observed using aluminium fluoride presumably as the result of lack of release of a suitable nucleophile from the reagent.

of the former at the sulphur atom attached to the phosphorus, leading to the phosphonium salt (6) which undergoes nucleophilic displacement *via* the five-co-ordinate intermediate (7) (Scheme 2). Co-ordination makes the sulphur atom highly apicophilic, forcing the sulphonate leaving group to enter an

equatorial position. After one Berry-pseudorotation leading to the trigonal bipyramid (8), the sulphonate group may depart, assisted most likely by further complexation with AlCl_3 .⁷

(Received, 17th September 1976; Com. 1065.)

¹ B. Krawiecka, J. Michalski, and Z. Skrzypczyński, *J.C.S. Chem. Comm.*, 1974, 1022.

² J. Michalski and M. Mikołajczyk, *Chem. and Ind.*, 1964, 661; *Tetrahedron*, 1966, **22**, 3059.

³ L. J. Szafraniec, L. P. Reiff, and H. S. Aaron, *J. Amer. Chem. Soc.*, 1970, **92**, 5275.

⁴ J. Michalski and Z. Skrzypczyński, *J. Organometallic Chem.*, 1975, **97**, C31-32.

⁵ L. J. Szafraniec, L. P. Reiff, and H. S. Aaron, *J. Amer. Chem. Soc.*, 1970, **92**, 6391.

⁶ J. Omelańczuk and M. Mikołajczyk, *Tetrahedron*, 1971, **27**, 5587.

⁷ W. W. Putschhova, E. N. Gurjanova, W. N. Lava, and E. N. Priležajeva, *Doklady Akad. Nauk S.S.S.R.*, 1972, **207**, 886; E. C. Dart and G. Holt, *J.C.S. Perkin I*, 1974, 1403.