

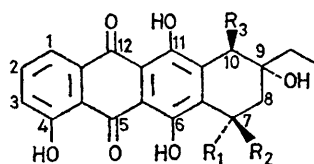
Total Synthesis of Rhodomycin Aglycones

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Summary The tetracyclic ketone (VIa), readily available from 1,4,5-trimethoxyanthraquinone, has been converted by C-9 vinylation and subsequent transformations into the four rhodomycin aglycones (I)—(IV).

RHODOMYCINONES (I)—(V) are the principal aglycones of the aminoglycoside antibiotics of the rhodomycin group,¹ which include mycetin A, B, and C.² We now report the first total synthesis of the racemic rhodomycinones (I)—(IV) by methods which could prove useful for the construction of similar aglycones in the citromycin³ and isorhodomycin⁴ series.

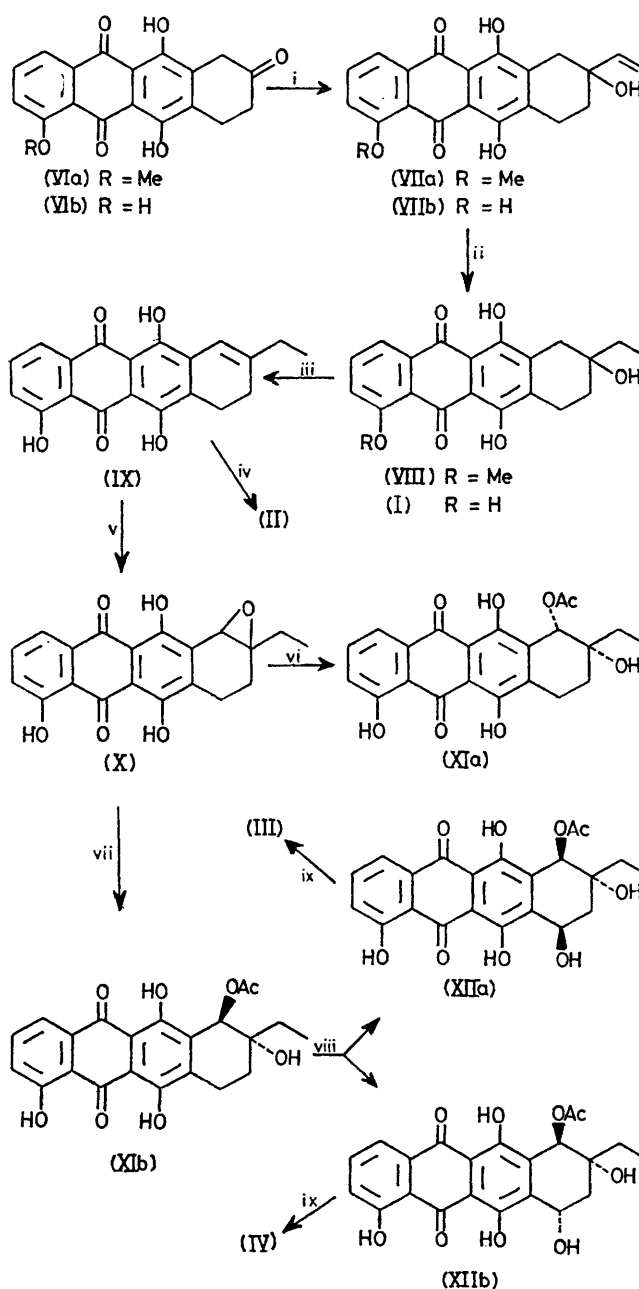


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|---------------------------------------|-----------------------------------|
| (I) $R^1 = R^2 = R^3 = H$ | 10-deoxy- γ -rhodomycinone |
| (II) $R^1 = R^2 = H, R^3 = OH$ | γ -rhodomycinone |
| (III) $R^1 = H, R^2 = R^3 = OH$ | α -rhodomycinone |
| (IV) $R^1 = R^3 = OH, R^2 = H$ | β -rhodomycinone |
| (V) $R^1 = OH, R^2 = H, R^3 = CO_2Me$ | ϵ -rhodomycinone |

Conversion of 1,4,5-trimethoxyanthraquinone into the crystalline tetracyclic ketone (VIa) in 25% yield over three steps has recently been reported.⁵ Demethylation of the ketone (VIa) ($AlCl_3$ in CH_2Cl_2 , room temp.) produced the ketone (VIb) (90%, m.p. 263—266 °C). Although the reaction of this ketone with $EtMgBr$ led mainly to enolization, reaction with $CH_2=CHMgBr$ was normal, giving the vinyl carbinol (VIIb) (50%), which on di-imide† reduction gave 96% of the red, crystalline (\pm)-10-deoxy- γ -rhodomycinone (I), m.p. 203—204 °C.‡

Similarly, the ketone (VIa) could be converted by way of the vinyl carbinol (VIIa) (m.p. 205—206 °C) into the alcohol (VIII), m.p. 209—210 °C, in 73% overall yield. Aluminium chloride effected both demethylation and dehydration to give 72% of the olefin (IX), m.p. 180—183 °C, δ ($CDCl_3$) 6.73 (s, 10-H). Stereospecific *trans*-hydroxylation⁶ converted (IX) in 70% yield into (\pm)- γ -rhodomycinone (II), m.p. 254—257 °C (decomp.), spectroscopically identical to the natural material.⁷

Attempts to introduce oxygen at C-7 by homolytic bromination⁵ of (II) led to oxidation at C-10. Therefore the olefin (IX) was transformed to the crystalline epoxide (X), m.p. 209—210 °C in 76% yield. Acetic acid at 85 °C cleaved (X) stereospecifically to give predominantly the 10-acetate (XIa) of (\pm)-*epi*- γ -rhodomycinone, R_f 0.38,



SCHEME. Reagents: i, $CH_2=CHMgBr$, tetrahydrofuran, -78 °C; ii, $KO_2CN_2CO_2K$, HOAc, pyridine, 65 °C; iii, $AlCl_3$, CH_2Cl_2 , 25 °C; iv, *o*-sulphobenzoic anhydride, see ref. 6, 30% H_2O_2 , acetone, 25 °C; v, *m*-chloroperbenzoic acid, CH_2Cl_2 , 25 °C; vi, HOAc, 85 °C; vii, HOAc-NaOAc, 85 °C; viii, Br_2 , CCl_4 , *hv*, then $AgOCOCF_3$, Me_2SO ; ix, 0.5N NaOH, EtOH, room temp.

† Hydrogenation of carbinols (VII) over Pd or Pt catalysts gave complex mixtures unsuitable for further transformations.

‡ Mass spectra, t.l.c. properties, and n.m.r. spectra of synthetic (I) were identical to those reported by Brockmann and Niemeyer, see ref. 1.

ν (CHCl_3) 1716 cm^{-1} .[§] In contrast, a mixture of acetic acid and 2–4% sodium acetate at 85°C gave both (XIa) and (\pm)- γ -rhodomycinone-10-acetate (XIb), R_f 0.31, ν (CHCl_3) 1738 cm^{-1} , in the ratio 1:9.[§] The acetate (XIb) was brominated, and the very labile C-7 bromination product was treated with silver trifluoroacetate and then hydrolysed to afford a 1:1 mixture of C-7 alcohols (50% yield), readily separable by preparative t.l.c. into (\pm)- α -rhodomycinone-10-acetate (XIIa), δ (CDCl_3) 5.28 (7-H, $\nu_{\frac{1}{2}}$ 16 Hz), and (\pm)- β -rhodomycinone-10-acetate (XIIb), δ (CDCl_3) 5.28 (7-H, $\nu_{\frac{1}{2}}$ 7 Hz). Hydrolysis of (XIIa) and (XIIb) yielded, respectively,

(\pm)- α -rhodomycinone (III) and (\pm)- β -rhodomycinone (IV), having mass spectrometric and chromatographic properties indistinguishable from the values recorded by Brockmann and Niemeyer.⁷

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[§] These R_f values refer to 0.5 mm silica gel 60 F-254 t.l.c. plates (Brinkmann), employing 3% MeOH in CH_2Cl_2 . The ester C=O shifts may reflect intramolecular H-bonding in the *cis* acetate (XIa), cf. H. B. Henbest and B. J. Lovell, *J. Chem. Soc.*, 1957, 1965.

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