

## Five- to Nine-bond $^{31}\text{P}$ – $^{31}\text{P}$ Spin Coupling Constants in Derivatives of Benzene and Naphthalene

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**Summary** The 1,5-, 1,6-, 2,7-, and 2,6-isomers of bis(diethoxyphosphino)methyl)naphthalene show interbenzylic  $^{31}\text{P}$ – $^{31}\text{P}$  spin coupling constants over seven, eight, and nine bonds, respectively, which are primarily determined by  $\pi$ -electron contributions.

SPIN coupling constants involving a  $^{31}\text{P}$  nucleus are usually rather large<sup>1</sup> because of the high effective nuclear charge of  $^{31}\text{P}$  compared to most other 'common' magnetic nuclei. Consequently, one expects to detect spin coupling between two phosphorus atoms even if they are separated by a considerable distance, yet  $^4J_{\text{PP}}$  is the longest-range coupling

reported in the literature.<sup>1</sup> The present communication describes  $^{31}\text{P}$ – $^{31}\text{P}$  couplings of considerable magnitude over a large number of bonds.

From the second-order nature of the proton-decoupled  $^{13}\text{C}$  n.m.r. spectra of the bis(phosphino)methylbenzenes (**1a**)–(**1c**) we concluded<sup>2</sup> that there exist sizeable  $J_{\text{PP}}$  couplings over five, six, and seven bonds, respectively,<sup>†</sup> and we have now studied the diphosphonates (**2a**)–(**2e**) which exhibit interbenzylic  $J_{\text{PP}}$  over up to nine bonds (Table). For the symmetrical molecules,  $J_{\text{PP}}$  was derived by two independent methods: (a) by observation of the outer  $^{13}\text{C}$  satellites in the  $^{31}\text{P}$  spectra ( $^1J_{\text{PC}}$  ca. 137 Hz) from which

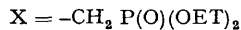
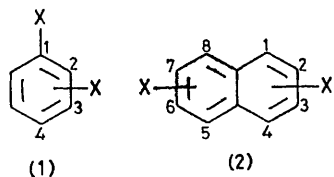
TABLE

$^{31}\text{P}$ – $^{31}\text{P}$ Spin coupling constants <sup>a</sup> in (1) and (2)								
	(1a)	(1b)	(1c)	(2a)	(2b)	(2c)	(2d)	(2e)
$^nJ_{\text{PP}}^b$	9.0(5)	3.1(6)	7.8(7)	1.4(7)	1.8(8)	1.4(8)	4.0(9)	5.3(5)

<sup>a</sup> Given in Hz, accuracy  $\pm 0.2$  Hz,  $(\text{CD}_3)_2\text{CO}$  (1) or  $\text{CDCl}_3$  (2) solutions. <sup>b</sup> Number of bonds separating the coupled nuclei given in parentheses.

<sup>†</sup> By analogy with the corresponding  $J_{\text{HH}}$  (M. Barfield and B. Chakrabarti, *Chem. Rev.*, 1969, **69**, 757) these  $J_{\text{PP}}$  are termed 'interbenzylic' coupling constants.

$J_{PP}$  can be measured directly and (b) by analysis of the five-line<sup>3</sup> ester  $^{13}\text{CH}_2\text{-}\{^1\text{H}\}$  and  $^{13}\text{CH}_3\text{-}\{^1\text{H}\}$  spectra which represent the X parts of AA'X spectra and from which  $J_{AA'}$  can be derived, because  $J_{AX}$  and  $J_{AX'}$  are known from



- |               |               |
|---------------|---------------|
| a; 1,2-isomer | a; 1,5-isomer |
| b; 1,3-isomer | b; 1,6-isomer |
| c; 1,4-isomer | c; 2,7-isomer |
|               | d; 2,6-isomer |
|               | e; 2,3-isomer |

the related monophosphonates.<sup>2</sup> In the unsymmetrical compound (**2b**),  $^8J_{PP}$  is taken directly from the  $^{31}\text{P}\text{-}\{^1\text{H}\}$  spectrum which shows an AB pattern.

The large magnitudes of the interbenzylic  $J_{PP}$ , in particular  $^9J_{PP} = 4.0\text{ Hz}$ , point to considerable  $\pi$ -electron contribution to these couplings. In the preferred conformations<sup>4</sup> of the present benzyl-type phosphonates the C-P bonds lie parallel to the adjacent carbon  $p_z$  orbitals. This geometry is most favourable for  $\sigma$ - $\pi$  exchange polarization which transmits the coupling information to the aromatic  $\pi$ -system and it is responsible for the large magnitude of the observed  $J_{PP}$ .

It is interesting to note the large decrease of  $^5J_{PP}$  as the mobile bond order of the intervening aromatic bond is decreased; compare (**1a**) with (**2e**).

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<sup>1</sup> G. Mavel, in 'Annual Reports on NMR Spectroscopy,' ed. E. F. Mooney, Vol. 5B, Academic Press, London, 1973, p. 1; R. Grinter, in 'Specialist Periodical Reports; Nuclear Magnetic Resonance,' Senior Reporter R. K. Harris, Vols. 1-5, The Chemical Society, 1972-1976, ch. 2.

<sup>2</sup> L. Ernst, *Org. Magnetic Resonance*, 1977, **9**, 35.

<sup>3</sup> M. Fild and W. Althoff, *J.C.S. Chem. Comm.*, 1973, 933.

<sup>4</sup> Cf. e.g. K. Chum, J. B. Rowbotham, and T. Schaefer, *Canad. J. Chem.*, 1974, **52**, 3489.