

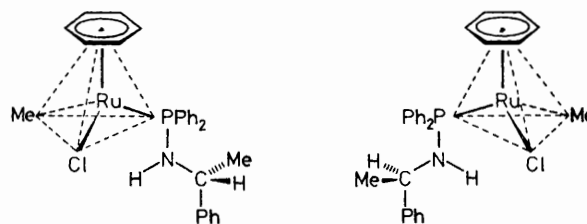
## Synthesis of Optically Active $\eta^6$ -Benzene-Ruthenium Complexes

By HENRI BRUNNER\* and ROBERT G. GASTINGER

(Universität Regensburg, Institut für Chemie, Universitätsstrasse 31, 8400 Regensburg, Germany)

**Summary** The first examples of optically active arene-ruthenium complexes (+)-(1) and (-)-(1), prepared by the stepwise reaction of benzene(dichloro)ruthenium dimer with dimethylmercury and (*R*)-(+)-Ph<sub>2</sub>PNHCH(Me)Ph and separated by chromatography, have stable configurations even in solution at 60 °C.

THE importance of chirality as a probe to follow the stereochemical course of reactions of organometallic complexes has been emphasized in recent years.<sup>1</sup> Although the number and variety of organometallic complexes which contain an asymmetric metal atom are rapidly increasing, there is, as yet, no report of the resolution of a chiral organometallic arene-ruthenium complex. A recent communication described the preparation but not resolution of chiral  $\eta^6$ -arene-ruthenium compounds.<sup>2</sup> Moreover, there is only one reported example of a resolved arene complex which contains chromium as the asymmetric metal atom.<sup>3</sup> We now report the first examples, (+)-(1) and (-)-(1), of optically active Ru complexes, which differ only in the configuration at the asymmetric Ru atom.



(1)

(+)-(1):  $[\alpha]_{578}^{25} + 645^\circ$ ,  $[\alpha]_{546}^{25} + 970^\circ$ ,  $[\alpha]_{436}^{25} - 1080^\circ$ ; (-)-(1):  $[\alpha]_{578}^{25} - 565^\circ$ ,  $[\alpha]_{546}^{25} - 840^\circ$ ,  $[\alpha]_{436}^{25} + 765^\circ$  (1 mg ml<sup>-1</sup>, C<sub>6</sub>H<sub>6</sub>).

A convenient synthesis of  $[\eta^6\text{-C}_6\text{H}_6\text{RuCl(Me)(PR}_3\text{)}]$ ,<sup>4,5</sup> where R = Ph, has been described and involves the alkylation of  $[(\eta^6\text{-C}_6\text{H}_6\text{RuCl}_2)_2]$  by dimethylmercury in acetonitrile followed by addition of a tertiary phosphine.<sup>4</sup> By employing the optically active phosphine (*R*)-(+)-Ph<sub>2</sub>PNHCH(Me)Ph<sup>6</sup> a pair of diastereoisomers is obtained which can be separated by dry-column chromatographic techniques<sup>7</sup> using alumina and ether-acetonitrile (15:1).

The more soluble isomer (–)-(1) (16% yield) as isolated from the first band during chromatography contained a small amount of the isomer (+)-(1) (ca. 10%). Pure (–)-(1) could be obtained by repeating the chromatographic step. Recrystallization from either benzene-hexane at 5 °C or acetonitrile at –30 °C yielded yellow needles of (–)-(1), m.p. 143–145 °C. The less soluble isomer (+)-(1) (19% yield) was obtained optically pure from the second band of the chromatogram as a yellow solid. Recrystallization from acetonitrile at –30 °C afforded orange prisms, m.p. 175–177 °C.

Both diastereoisomers (+)-(1) and (–)-(1) gave satisfactory elemental analyses. Their mass spectra showed the molecular ion. The c.d. spectra of (+)-(1) and (–)-(1) are depicted in the Figure. Compounds (+)-(1) and (–)-(1) differ in their <sup>1</sup>H n.m.r. spectra (CDCl<sub>3</sub>, Me<sub>4</sub>Si): (+)-(1),  $\tau$  8.70 [d,  $J(\text{Me-P})$  6.2 Hz, Ru–Me], 8.99 [d,  $J(\text{Me-CH})$  7.2 Hz, C–Me], 6.00 (m, C–H), 4.93 [d,  $J(\text{C}_6\text{H}_6\text{-P})$  0.9 Hz,  $\eta^6\text{-C}_6\text{H}_6$ ], and 2.68 (m, Ph); (–)-(1),  $\tau$  9.17 [d,  $J(\text{Me-P})$  6.3 Hz, Ru–Me], 8.97 [d,  $J(\text{Me-CH})$  7.2 Hz, C–Me], 6.00 (m, C–H), 4.87 [d,  $J(\text{C}_6\text{H}_6\text{-P})$  0.9 Hz,  $\eta^6\text{-C}_6\text{H}_6$ ], and 2.76 (m, Ph). The difference in the Ru–Me resonances and, to some extent, in the  $\eta^6\text{-C}_6\text{H}_6$  resonances is sufficiently large to allow their use as a gauge of optical purity as well as a monitor of epimerization reactions.

Both isomers have stable configurations at room temperature in the solid state as well as in benzene, toluene, or chloroform solutions when kept oxygen-free. Moreover, no

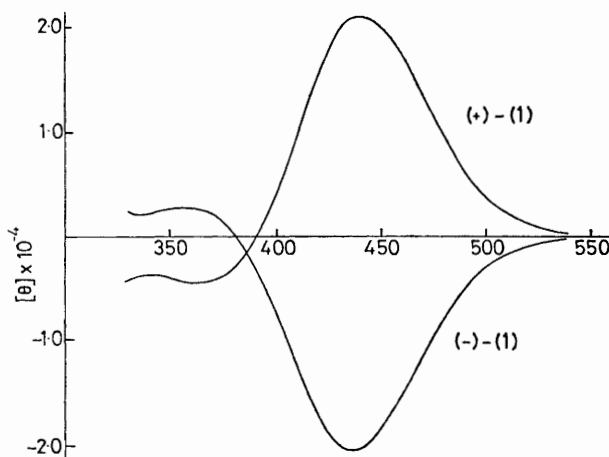


FIGURE. C.d. spectra of (+)-(1) and (–)-(1) in ca. 0.002 M C<sub>6</sub>H<sub>6</sub> solution at 20 °C.

epimerization can be observed upon heating these solutions up to 60 °C for at least 18 h. Therefore, the utility of (+)-(1) and (–)-(1) in investigations of the stereochemistry of reactions at Ru–ligand bonds is evident.

We thank the Deutsche Forschungsgemeinschaft and the Fonds der Chemischen Industrie for financial support.

(Received, 29th March 1977; Com. 294.)

<sup>1</sup> H. Brunner, *Ann. New York Acad. Sci.*, 1974, **239**, 213; H. Brunner, *Topics Current Chem.*, 1975, **56**, 67.

<sup>2</sup> D. F. Dersnah and M. C. Baird, *J. Organometallic Chem.*, 1977, **127**, C55.

<sup>3</sup> G. Simonneaux, A. Meyer, and G. Jaouen, *J.C.S. Chem. Comm.*, 1975, 69.

<sup>4</sup> R. A. Zelonka and M. C. Baird, *J. Organometallic Chem.*, 1972, **44**, 383.

<sup>5</sup> M. A. Bennett and A. K. Smith, *J.C.S. Dalton*, 1974, 233.

<sup>6</sup> H. Brunner and W. Rambold, *Angew. Chem.*, 1973, **85**, 1118; *Angew. Chem. Internat. Edn.*, 1973, **13**, 1013.

<sup>7</sup> B. Loev and M. M. Goodman, *Chem. and Ind.*, 1967, 2026.