

Reaction of *o*-Hydroxyphenyl Alkyl Ketones with Phosphorus Trichloride. Formation of 1,3,2-Dioxaphosphorinans *via* α -Proton Abstraction

By GORDON M. L. CRAGG,* BETTE DAVIDOWITZ, and ROBIN G. F. GILES

(Department of Organic Chemistry, University of Cape Town, Private Bag, Rondebosch, Cape Town, Republic of South Africa)

and RAYMOND J. HAINES*

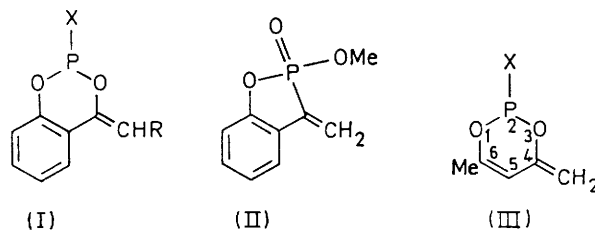
(Department of Chemistry, University of Natal, P.O. Box 375, Pietermaritzburg, Republic of South Africa)

Summary Treatment of PCl_3 with *o*- $\text{HOC}_6\text{H}_4\text{COCH}_2\text{R}$ ($\text{R} = \text{H}$ or Et) and NEt_3 affords 2-chloro-1,3,2-dioxaphosphorinans, which can be readily converted into the corresponding 2-alkoxy derivatives; evidence is presented for the formation of analogous derivatives on similar reaction of acetylacetone and PCl_3 .

As part of a general study of the influence of transition metals on the photochemistry of organic molecules, we attempted the synthesis of tertiary aryl phosphites of the type $\text{P}(\text{OC}_6\text{H}_4\text{COCH}_2\text{R})_3$ ($\text{R} = \text{H}$, Me , *etc.*) by reaction of PCl_3 with *o*- $\text{HOC}_6\text{H}_4\text{COCH}_2\text{R}$ in the presence of an appropriate base.¹ Mass and ^1H n.m.r. spectral data, however, indicated the product of the reaction involving PCl_3 with three-fold molar amounts of *o*- $\text{HOC}_6\text{H}_4\text{COMe}$ and NEt_3 to be the 1,3,2-dioxaphosphorinan (I; $\text{R} = \text{H}$; $\text{X} = \text{o-OC}_6\text{H}_4\text{-COMe}$) and not $\text{P}(\text{o-OC}_6\text{H}_4\text{COMe})_3$.

Although this species could not be isolated in a sufficiently pure form for complete characterisation, the 2-chloro-1,3,2-dioxaphosphorinan (I; $\text{R} = \text{H}$; $\text{X} = \text{Cl}$) was readily obtained by reaction of *o*- $\text{HOC}_6\text{H}_4\text{COMe}$, PCl_3 , and NEt_3 in a molar ratio of 1:2.5:2.5. This was, in turn, treated with various alcohols and phenol in the presence of NEt_3 to give the corresponding 2-alkoxy and 2-phenoxy derivatives

(I; $\text{R} = \text{H}$; $\text{X} = \text{OMe}$, OPr^t , OBu^t , OCH_2Ph , and OPh) (see Table), which were characterised by elemental analysis as well as by i.r. and ^1H n.m.r. spectroscopy and accurate mass determination of the molecular ions. The n.m.r. spectra of these derivatives contained, besides peaks associated with the phenylene and 2-alkoxy- and 2-phenoxy-groups, two broad resonances in the regions δ 4.6–4.9 and



5.0–5.3, readily assigned to the exocyclic methylene group. The absence of peaks between 1650 and 1750 cm^{-1} in the i.r. spectra of these compounds is also consistent with the proposed structure (I). Reaction of (I; $\text{R} = \text{H}$; $\text{X} = \text{OMe}$) with $[\text{Fe}(\text{cp})(\text{CO})_2\text{I}]$ ($\text{cp} = \eta\text{-C}_5\text{H}_5$) in refluxing benzene afforded a single product, the spectroscopic data of which

are consistent with it being $[\text{Fe}(\text{cp})(\text{CO})(\text{MeOPC}_6\text{H}_4\text{COCH}_2)\text{I}]$ [$\nu(\text{C}=\text{O})$, 1977 cm^{-1} in CH_2Cl_2 ; δ 3.76 (d, J 11.6 Hz, OMe), 4.34 (s, C_5H_5) and 4.36 and 5.13 (d, J 2 Hz, $=\text{CH}_2$)]. The formation of this complex provides support for structure (I) as opposed to the other possible structure (II), since the latter would not be expected to give rise to a stable substituted derivative of $[\text{Fe}(\text{cp})(\text{CO})_2\text{I}]$. Further evidence against structure (II) is the absence of any band corresponding to $\text{P}=\text{O}$ in the i.r. spectra of the products. In addition, treatment of the derivatives (I; $\text{R} = \text{H}$; $\text{X} = \text{OMe}$ and OPh) in ether with anhydrous HCl regenerated the ketone fragment, $o\text{-OC}_6\text{H}_4\text{COMe}$, as indicated by the dis-

TABLE

Compound	B.p./°C (mmHg)	Isolated yield (%)
(I; $\text{R} = \text{H}$)		
{ $\text{X} = \text{Cl}$	72—76 (0.3)	45 ^a
{ $\text{X} = \text{OMe}$	64—67 (0.35)	65 ^b
{ $\text{X} = \text{OCHMe}_2$	73—78 (0.27)	65 ^b
{ $\text{X} = \text{OCMe}_3$	83—88 (0.20)	55 ^b
{ $\text{X} = \text{OPh}$	123—127 (0.30)	57 ^b
{ $\text{X} = \text{OCH}_2\text{Ph}$	140—144 (0.30)	34 ^b
(I; $\text{R} = \text{Et}$; $\text{X} = \text{Cl}$)	c	c
(I; $\text{R} = \text{Et}$; $\text{X} = \text{OMe}$)	93—98 (0.27)	55 ^b
(III; $\text{X} = \text{Cl}$)	28—38 (0.35)	20 ^a
(III; $\text{X} = \text{OMe}$)	23 (0.55)	15 ^d

^a Based on ketone. ^b Based on (I; $\text{X} = \text{Cl}$; $\text{R} = \text{H}$ or Et).
^c Product not distilled. ^d Based on (III; $\text{X} = \text{Cl}$).

appearance of the resonances due to the exocyclic methylene group, and the appearance of a singlet at δ 2.64 due to the methyl ketone, in the ^1H n.m.r. spectra. The reaction of $o\text{-HOC}_6\text{H}_4\text{COPr}^n$, PCl_3 , and NEt_3 in a molar ratio of 1:2.5:2.5 also afforded the 2-chloro-1,3,2-dioxaphosphorinan (I; $\text{R} = \text{Et}$; $\text{X} = \text{Cl}$), which was characterised as the methoxy derivative, (I; $\text{R} = \text{Et}$; $\text{X} = \text{OMe}$), formed by reaction with methanol in the presence of NEt_3 (see Table).

Although there is no n.m.r. evidence for the presence of the enol tautomer of $o\text{-HOC}_6\text{H}_4\text{COCH}_2\text{R}$ in solution, (I; $\text{R} = \text{H}$ or Et ; $\text{X} = \text{Cl}$) may be considered as having been formed from $o\text{-HOC}_6\text{H}_4\text{C}(\text{OH})=\text{CHR}$ by proton abstraction from the two hydroxy groups. Attempts were thus made to establish whether an analogous dioxaphosphorinan can be formed by reaction of PCl_3 with acetylacetone in the presence of NEt_3 ; reaction under similar conditions to those used above did, in fact, afford a product, proposed to be the 2-chloro derivative (III; $\text{X} = \text{Cl}$), on the basis of ^1H n.m.r. evidence [δ 5.50 (d, J 3 Hz, 5-H), 4.50 and 4.35 (br s, $=\text{CH}_2$), and 1.92 (s, Me)]. Treatment of (III; $\text{X} = \text{Cl}$) with

methanol in the presence of NEt_3 afforded an unstable product (m/e 160) postulated to be the 2-methoxy derivative (III; $\text{X} = \text{OMe}$), on the basis of n.m.r. evidence [δ 5.38 (d, J 3 Hz, 5-H), 4.34 and 4.12 (br s, $=\text{CH}_2$), 3.65 (d, J 11 Hz, OMe), and 1.90 (s, Me)] (see Table). The coupling of 3 Hz observed between ^{31}P and the 5-H in (III; $\text{X} = \text{Cl}$ or OMe) is similar to that observed for 4J ($\text{POC}=\text{CH}$) in *transoid* enolic phosphates;² the signals due to the $=\text{CH}_2$ -hydrogens, however, appear as broad singlets (ω_t 4 Hz), indicating weaker coupling of these protons to the phosphorus. Similar line broadening is observed for the exocyclic methylene hydrogen signals of the dioxaphosphorinans (I; $\text{R} = \text{H}$; $\text{X} = \text{Cl}$ or OR'), while, in the case of (I; $\text{R} = \text{Et}$; $\text{X} = \text{Cl}$ or OMe), the olefinic proton signal appears as a triplet (J 7 Hz) further split by coupling to phosphorus (J 2 Hz).

The formation of dienolate anions has been invoked in the treatment of β -diketones with strong bases,³ but the isolation of a dienolic derivative of a β -diketone has, to the best of our knowledge, not previously been reported. In this respect, the reactions of acetylacetone with BCl_3 , AlCl_3 , and SiCl_4 have been shown to afford $[\text{B}(\text{acac})_2]^+$, $[\text{Al}(\text{acac})_3]$, and $[\text{Si}(\text{acac})_3]^+$ (acac = acetylacetonate) respectively, with no α -proton abstraction being reported.⁴

The synthesis of the 2-methoxy derivative (I; $\text{R} = \text{H}$; $\text{X} = \text{OMe}$) illustrates a typical reaction procedure. o -Hydroxyacetophenone (0.15 mol) in ether (100 ml) was added dropwise with stirring to a solution of NEt_3 (0.375 mol) and PCl_3 (0.375 mol) in ether (300 ml). After stirring the mixture for 2 h the precipitate of $\text{NEt}_3\cdot\text{HCl}$ was filtered off through glass wool and the ether removed under reduced pressure. Distillation of the residue gave the 2-chloro derivative (I; $\text{R} = \text{H}$, $\text{X} = \text{Cl}$) (45%), b.p. 72—76 °C at 0.3 mmHg. NEt_3 (0.02 mol) was added to a solution of the 2-chloro derivative (0.02 mol) in ether (25 ml), followed by the dropwise addition of a solution of MeOH (0.06 mol) in ether (10 ml). After stirring the mixture for 2 h it was filtered through glass wool and the ether was removed under reduced pressure. Distillation of the residue gave the 2-methoxy derivative (65%), b.p. 64—67 °C at 0.35 mmHg. All operations were carried out under anhydrous conditions and under nitrogen.

The authors thank the South African Council for Scientific and Industrial Research and the Universities of Cape Town and Natal for financial support.

(Received, 10th May 1977; Com. 451.)

¹ W. Gerrard and H. R. Hudson, 'Organic Phosphorus Compounds,' vol. 5, eds., G. M. Kosolapoff and L. Maier, Wiley-Interscience, New York, 1972, p. 21.

² E. M. Gaydou, *Tetrahedron Letters*, 1972, 4469.

³ H. O. House, 'Modern Synthetic Reactions,' 2nd edn., Benjamin, Menlo Park, California, 1972, p. 554.

⁴ J. P. Fackler, Jr., *Progr. Inorg. Chem.*, 1966, 7, 361 and references quoted therein.